

## Assignment #1: due Thursday, Oct. 21, 2010

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### Theoretical Astrophysics

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#### 1. Number of stars in the Milky Way 5 pt

Does the Milky Way Galaxy contain more stars than there are grains of sand in the beach volleyball court at the Neckarwiese? Please justify your answer using simple order-of-magnitude estimates.

#### 2. Timescale estimates for the Sun 20 pt

- (a) *Dynamical timescale:* The collapse timescale for a self-gravitating object is given by  $t_{\text{dyn}} \approx 1/\sqrt{G\rho}$ . Calculate it for the Sun assuming a mean density of  $\rho = 1.4 \text{ g cm}^{-3}$
- (b) *Sound crossing time:* The sound speed in the solar core is roughly  $350 \text{ km sec}^{-1}$ . How long does a sound wave need to cross the Sun assuming a constant sound speed throughout the Sun (the solar radius is  $6.96 \times 10^5 \text{ km}$ ).
- (c) *Nuclear timescale:* The Sun's energy is produced by the process of fusion of hydrogen into helium. If 10% of the solar mass is consumed in this process during the Sun's lifetime, how long does the Sun's energy production persist if the Sun's energy loss (i.e. luminosity:  $L_{\odot} = 3.846 \times 10^{33} \text{ erg s}^{-1}$ ) is constant during that time. Use the formula  $\tau = E/\dot{E}$  to estimate the nuclear time scale. Note that 0.7% of the hydrogen rest mass is turned into energy in the fusion process.
- (d) *Kelvin-Helmholtz timescale:* The Kelvin-Helmholtz timescale is the ratio of the gravitational energy of an object to its luminosity. Calculate the Kelvin-Helmholtz timescale for the Sun. Assume that the Sun has a constant density.

#### 3. Molecular excitation 15 pt

Giant molecular clouds (GMCs) are composed almost entirely of molecular hydrogen ( $\text{H}_2$ ), but also contain small quantities of tracer molecules. The most important of these tracers is carbon monoxide (CO). Assume that the molecular hydrogen in a GMC has a Maxwell-Boltzmann velocity distribution. Compute the temperature at which an  $\text{H}_2$  molecule with a kinetic energy equal to the mean kinetic energy of the distribution can excite a CO molecule from its ground state to the:

- (a)  $J = 1$  excited rotational level ( $\Delta E = 4.76 \times 10^{-4}$  eV)
- (b)  $v = 1$  excited vibrational level ( $\Delta E = 0.266$  eV)
- (c) B  $^1\Sigma^+$  excited electronic level ( $\Delta E = 10.5$  eV)

[Note:  $1 \text{ eV} \simeq 1.6 \times 10^{-12}$  erg]. Ignore the effects of any internal excitation of the  $\text{H}_2$  molecules, and the contribution of the CO to the mean molecular weight of the gas.

In a typical GMC, the temperature of the gas is in the range 10–20 K. Which of these levels will be excited?

#### 4. Relaxation to equilibrium

(bonus points 20 pt)

Consider an ideal gas with a distribution function  $f = f_0 + g$ , where  $f_0$  is the Maxwell distribution function and  $g$  is a small perturbation.

- (a) Give an expression for the collision term  $\dot{f}_c$  in the Boltzmann equation in terms of  $f_0$  and  $g$ . [Hint: Use the kinetic theory of elastic encounters and consider only terms up to first order in the perturbation term. ]
- (b) A coarse approximation for  $\dot{f}_c$  can be obtained by using the collision approximation (*Stoßzahlansatz*) as discussed in the lecture and by looking at the functional form of the individual terms in  $\dot{f}_c$ . Take one as being representative and show that

$$\dot{f}_c \sim -gn\sigma_{\text{tot}}\bar{u}_{\text{rel}}, \quad (1)$$

where  $n$  is the number density of particles,  $\sigma_{\text{tot}}$  is the total collision cross-section and  $\bar{u}_{\text{rel}}$  is the mean relative velocity between the particles.

- (c) Using Eq. 1, show that the Boltzmann equation can be written (approximately) as:

$$\frac{\partial f}{\partial t} + \vec{w} \cdot \vec{\nabla}_x f + \frac{\vec{F}}{m} \cdot \vec{\nabla}_w f = -\frac{f - f_0}{\tau}. \quad (2)$$

where  $m$  is the particle mass and  $\tau = 1/(n\sigma_{\text{tot}}\bar{u}_{\text{rel}})$ . Discuss the physical interpretation of  $\tau$ .