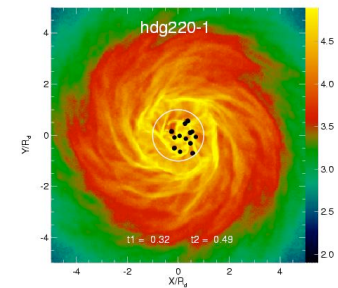
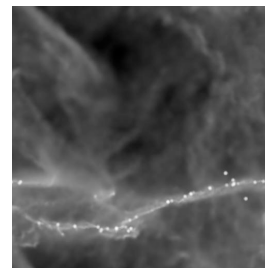
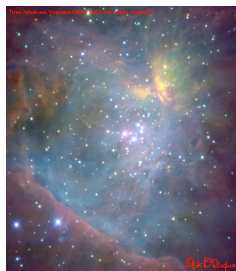
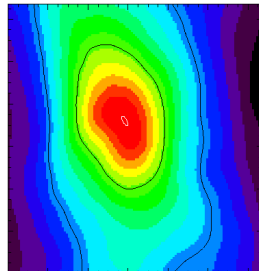
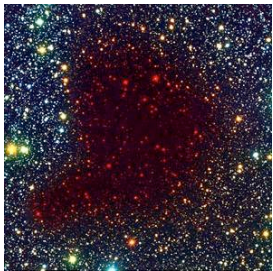


ISM Dynamics and Star Formation



Ralf Klessen

Zentrum für Astronomie der Universität Heidelberg
Institut für Theoretische Astrophysik





thanks to ...

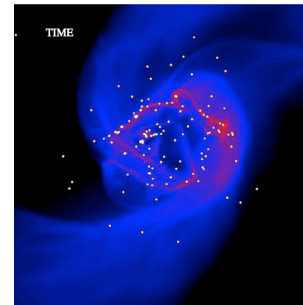
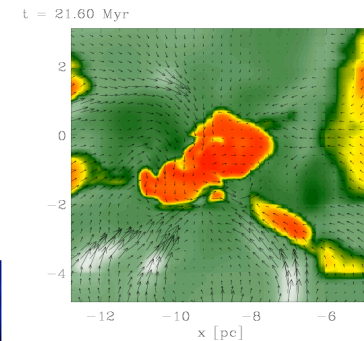
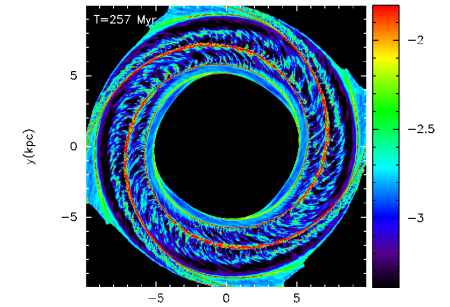
● many thanks to

- Robi Banerjee (ITA)
- Paul Clark (ITA)
- Clare Dobbs (Munich)
- Christoph Federrath (ITA)
- Simon Glover (ITA)
- Thomas Greif (MPA)
- Patrick Hennebelle (ENS)
- Mordecai Mac Low (AMNH)
- Dominik Schleicher (ESO)
- Enrique Vazquez-Semadeni (UNAM)



Agenda

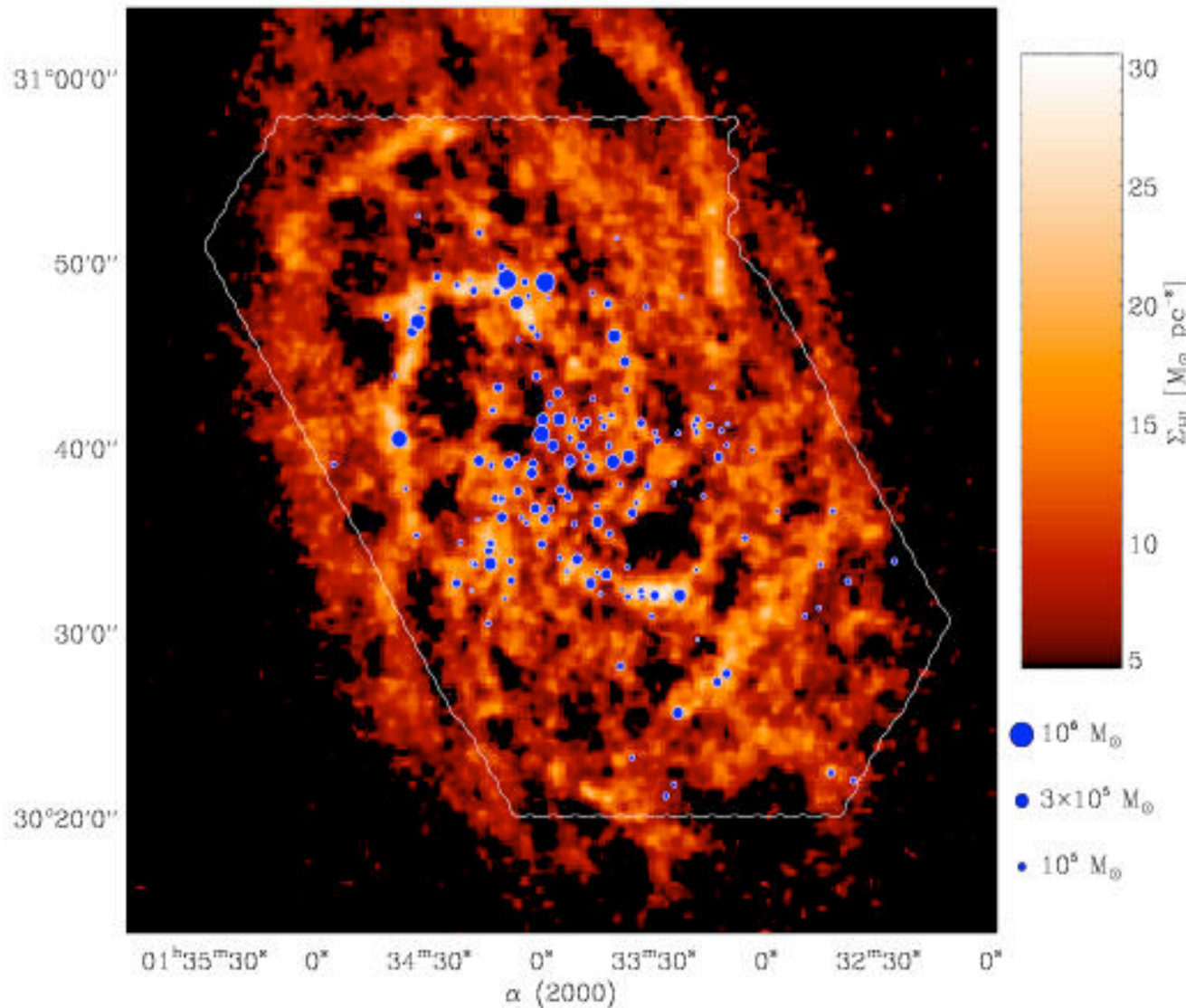
- phenomenology
 - what we need to explain
- dynamic star formation theory
 - gravity vs. turbulence (and all the rest)
- examples and predictions
 - formation of molecular clouds in galactic disks (H_2 & CO chemistry)
 - universal IMF: importance of turbulence and thermodynamics





phenomenology

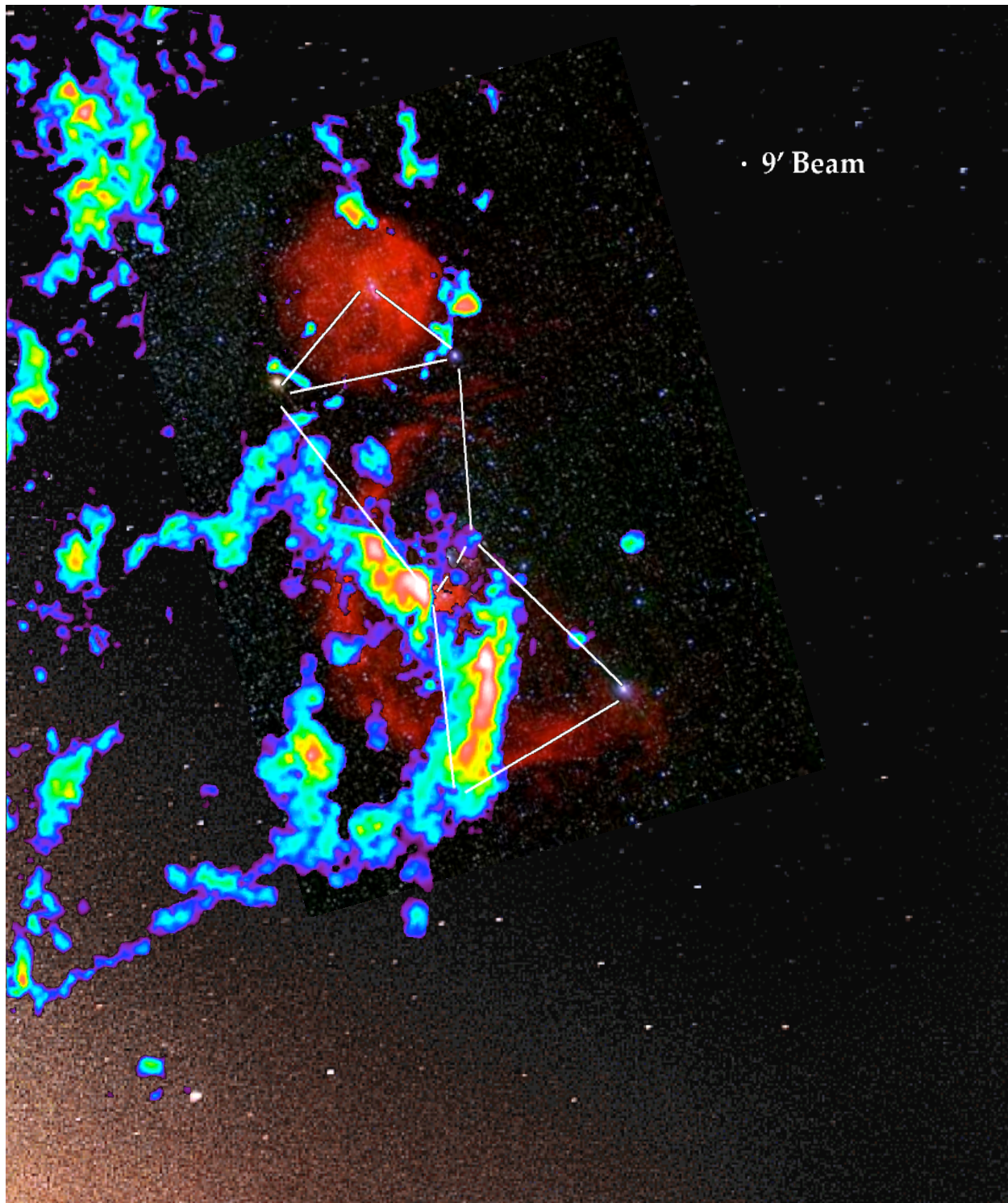
correlation between H₂ and HI



Compare H₂ - HI
in M33:

- H₂: BIMA-SONG Survey, see Blitz et al.
- HI: Observations with Westerbork Radio T.

H₂ clouds are seen in
regions of high HI
density
(in spiral arms and
filaments)

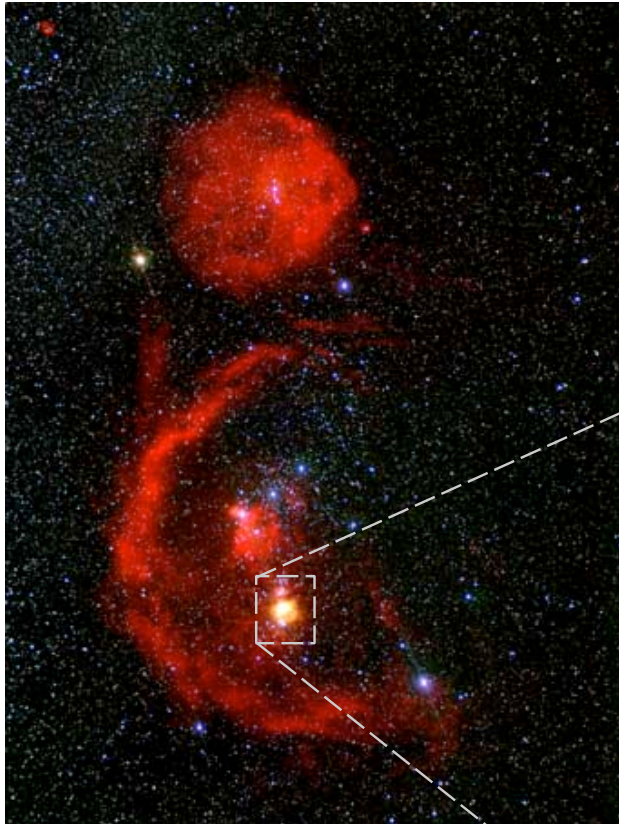


Star formation in Orion

We see

- *Stars* (in visible light)
- Atomic hydrogen (in $H\alpha$ -- red)
- Molecular hydrogen H_2 (radio emission -- color coded)

Local star forming region: The Trapezium Cluster in Orion



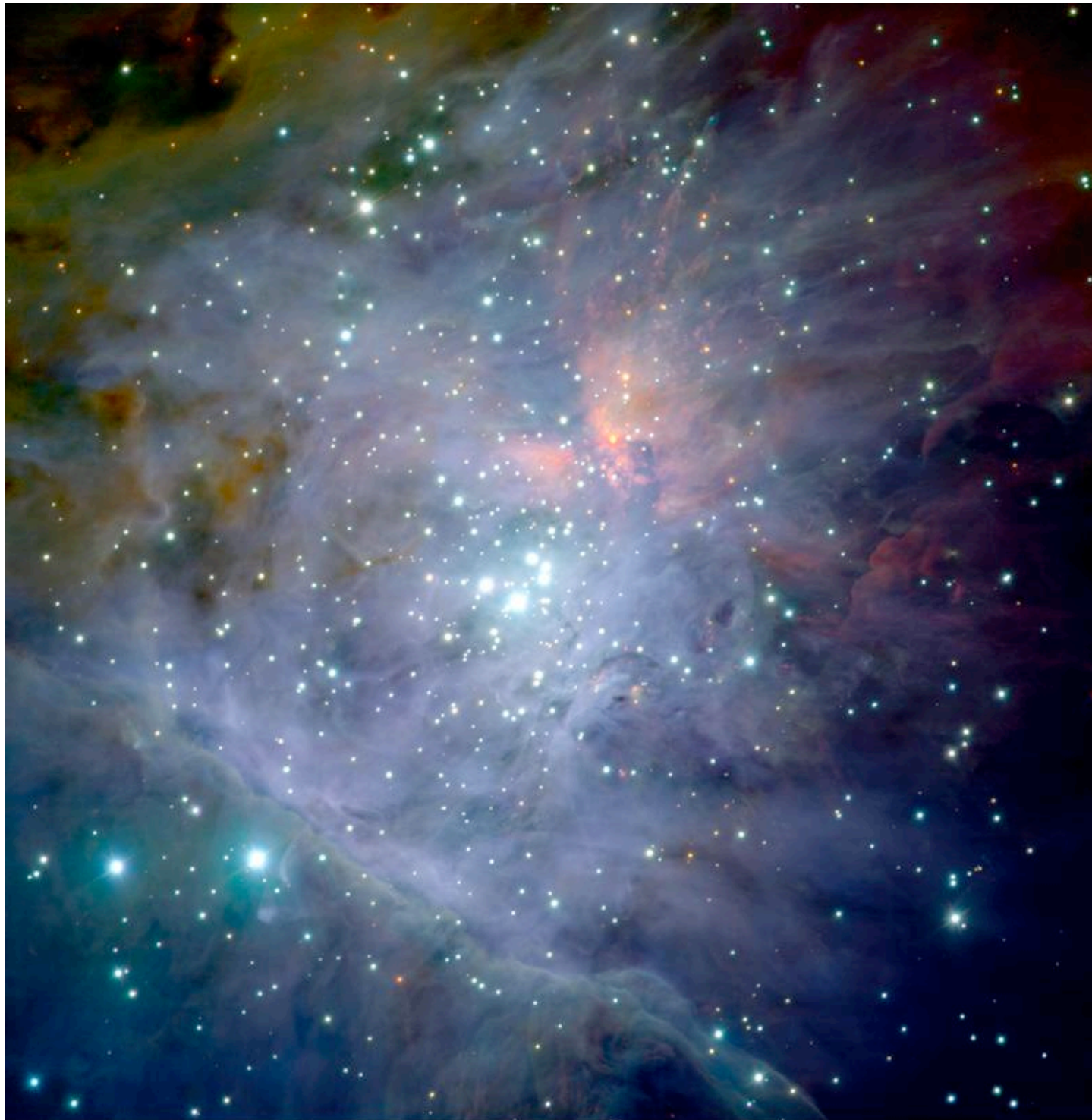
Orion molecular cloud

The Orion molecular cloud is the birth- place of several young embedded star clusters.

The Trapezium cluster is only visible in the IR and contains about 2000 newly born stars.



Trapezium cluster



Trapezium Cluster

(detail)

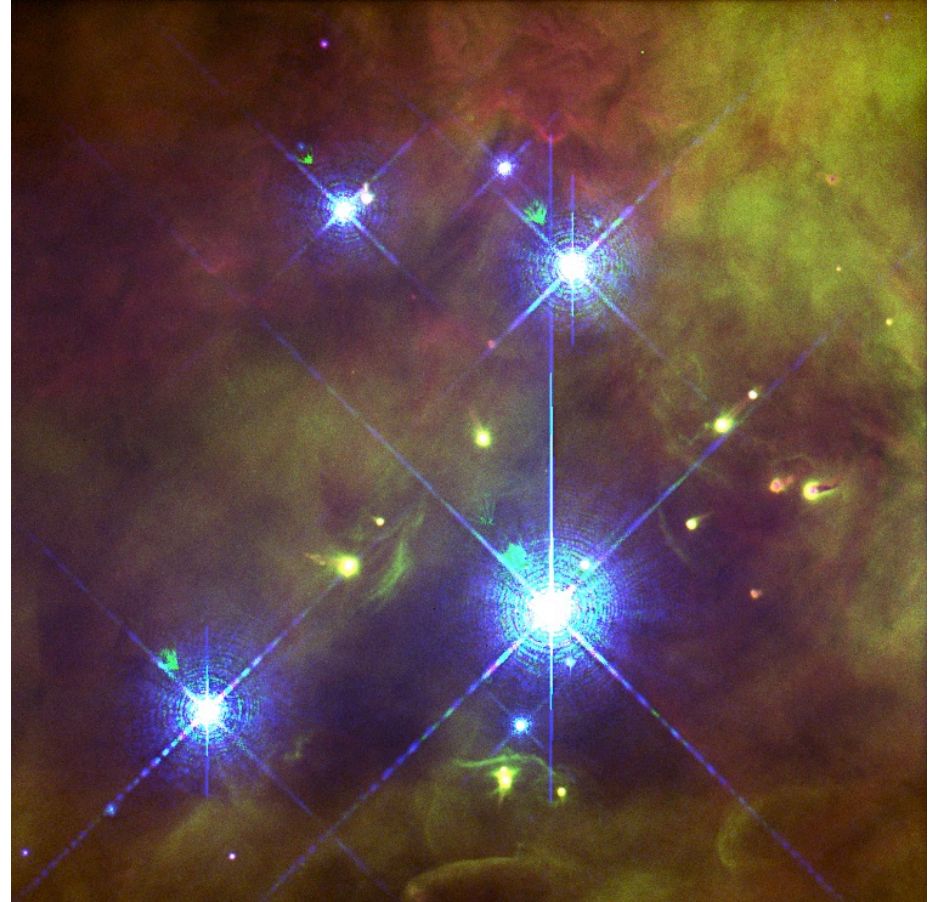
- stars form in **clusters**
- stars form in **molecular clouds**
- (proto)stellar **feedback** is important

(color composite J,H,K
by M. McCaughrean,
VLT, Paranal, Chile)

Trapezium Cluster: Central Region



Ionizing radiation from central star
 Θ 1C Orionis



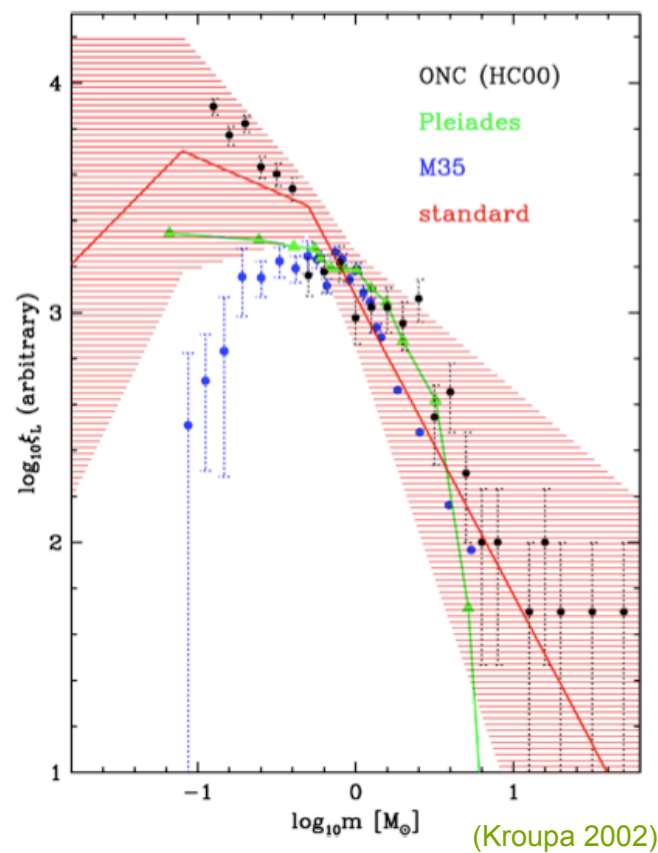
Proplyds: Evaporating ``protoplanetary`` disks
around young low-mass protostars

(images: Doug Johnstone et al.)



stellar mass function

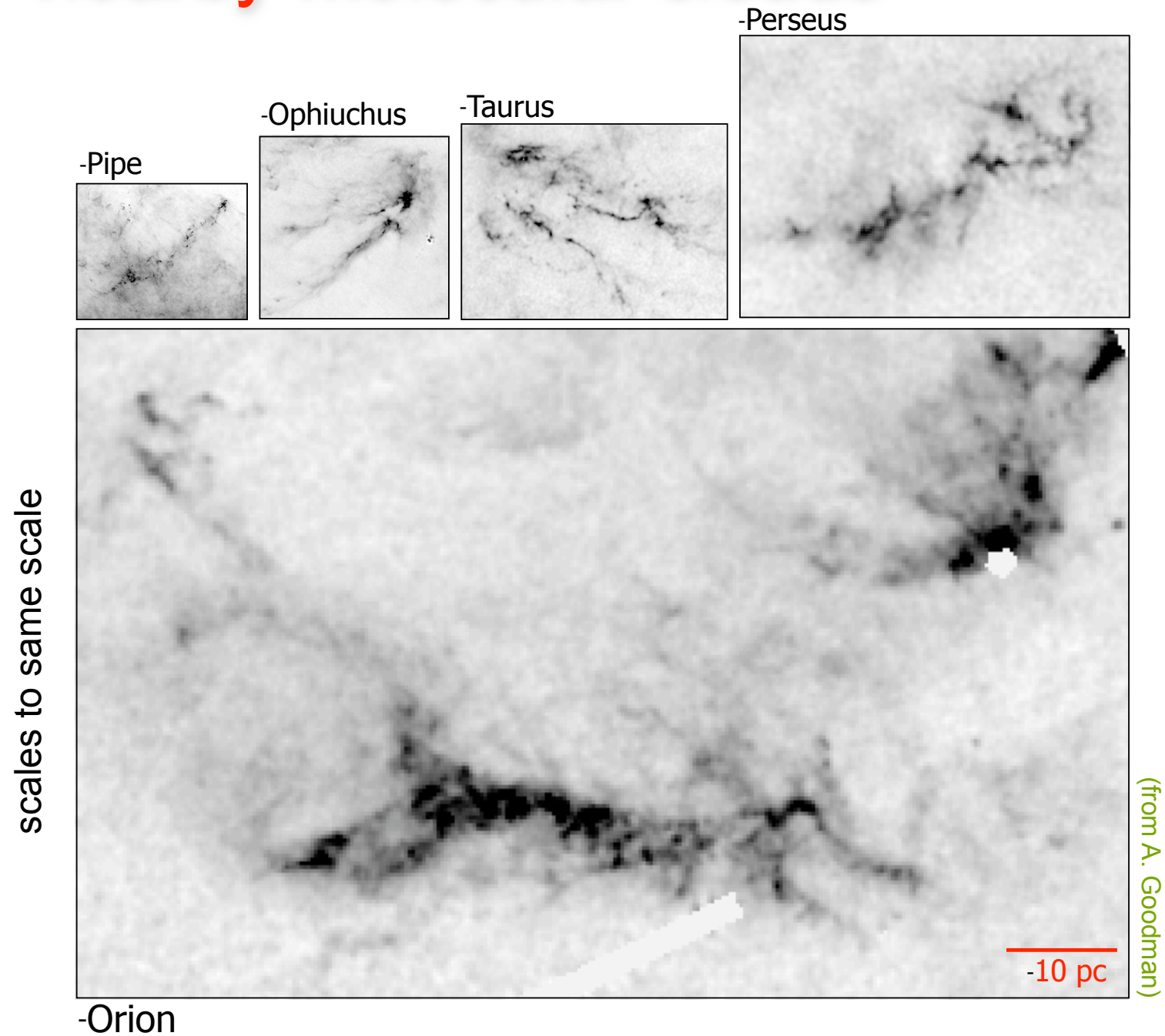
- stars seem to follow a universal mass function at birth --> IMF



Orion, NGC 3603, 30 Doradus
(Zinnecker & Yorke 2007)

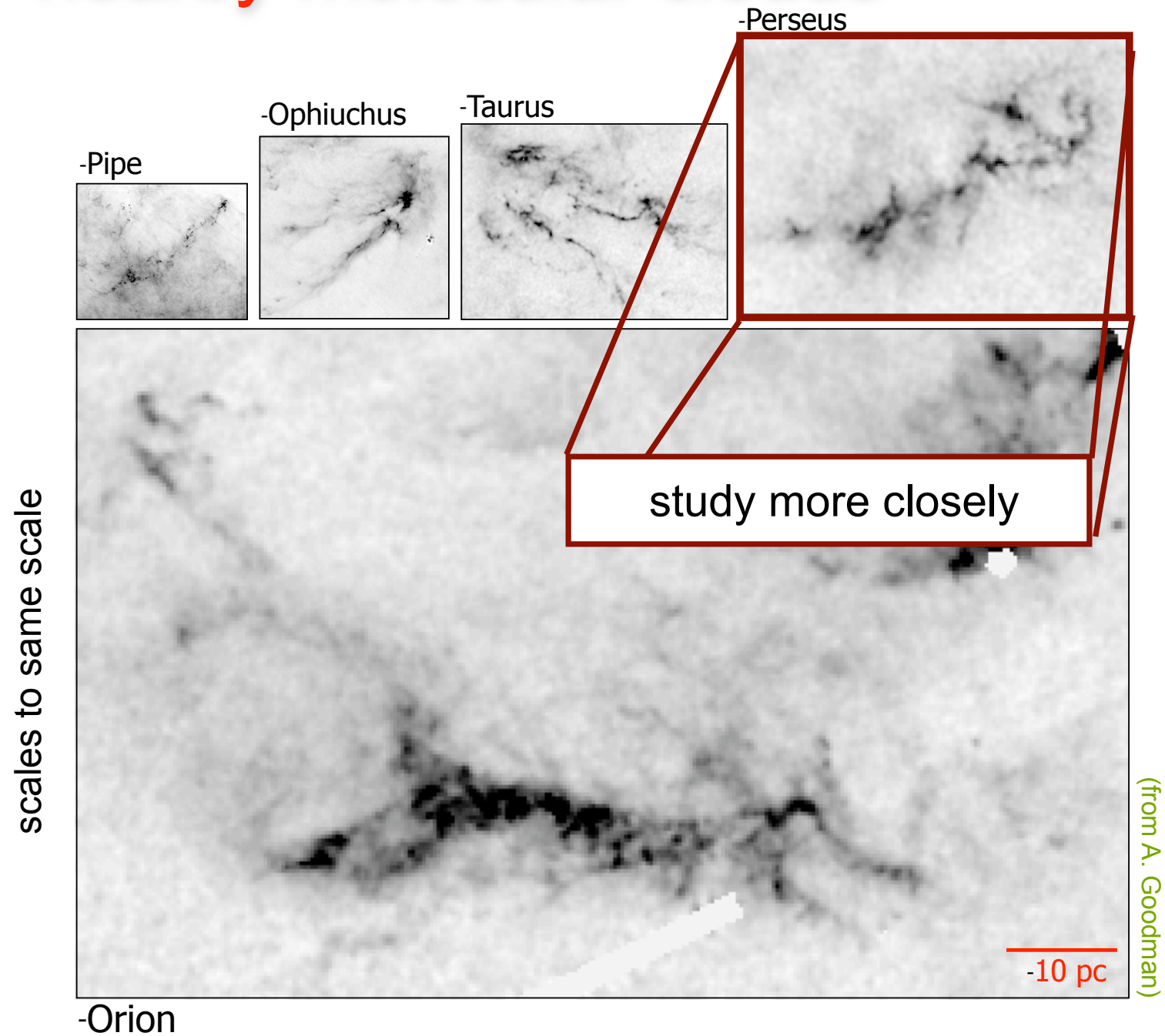


nearby molecular clouds





nearby molecular clouds



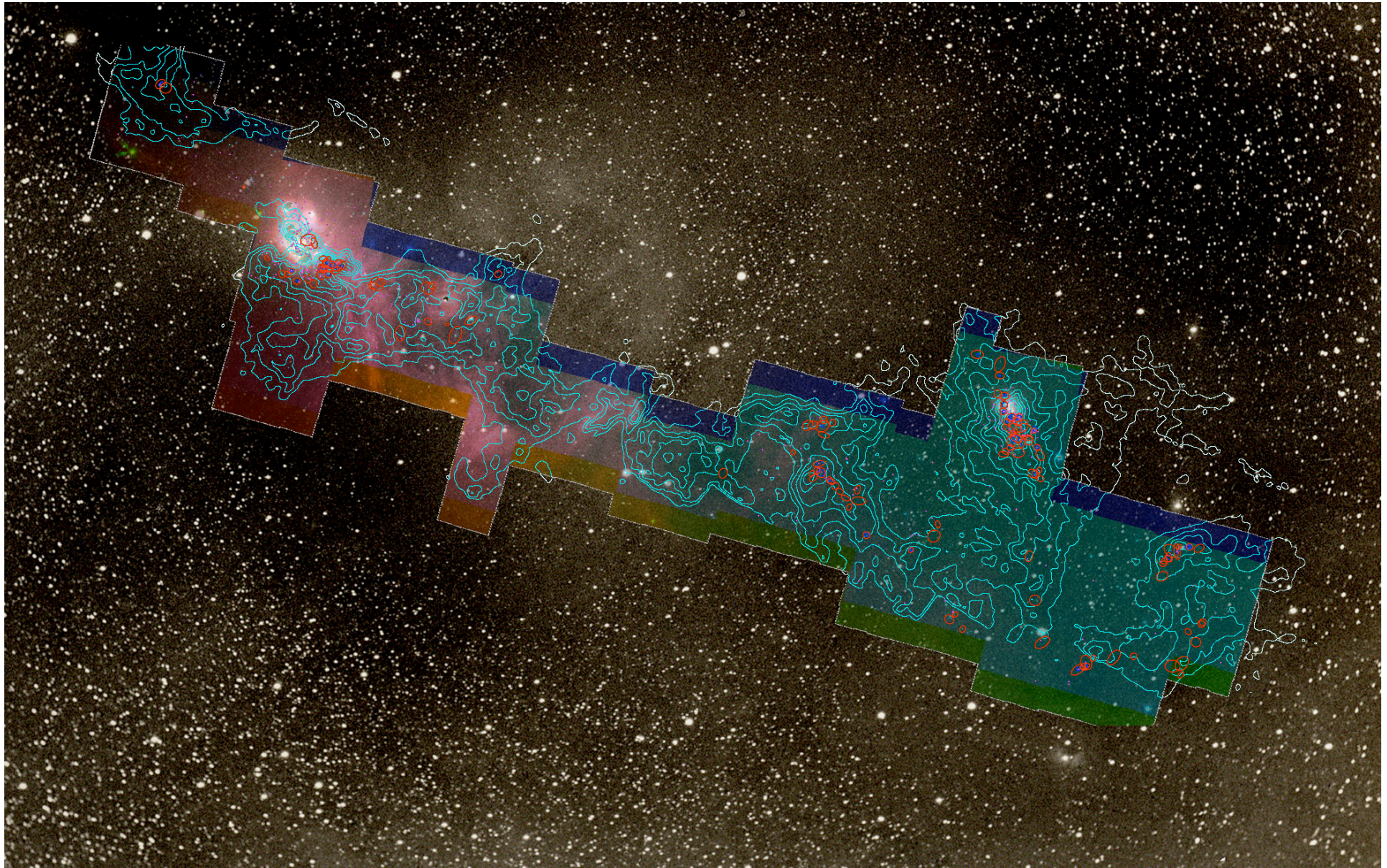
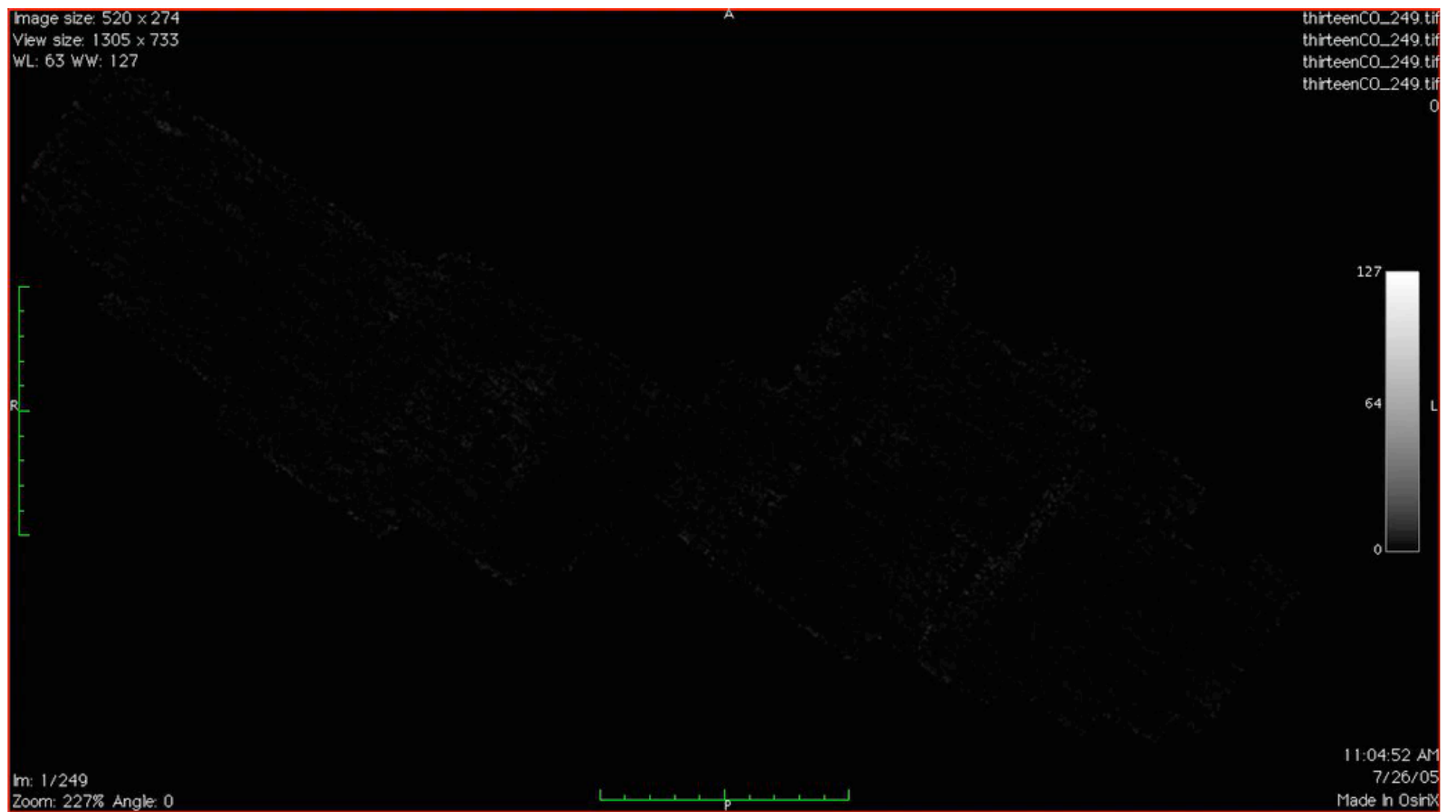


image from Alyssa Goodman: COMPLETE survey



LOS Geschwindigkeitsverteilung in Perseus

velocity cube from Alyssa Goodman: COMPLETE survey



what we need to consider ...

- *correlation* between large and small scales in galaxy (stars “know” where to form and when)
- all stars form in *molecular cloud* complexes (star formation linked to molecular cloud formation)
- molecular clouds are *turbulent* (understand turbulence to understand star formation)
- stars form in *clusters* (importance of dynamical interactions during formation)
- star formation has *universal* characteristics (e.g. initial mass function)



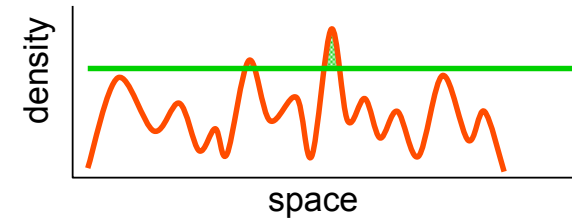
basic idea



dynamical SF in a nutshell

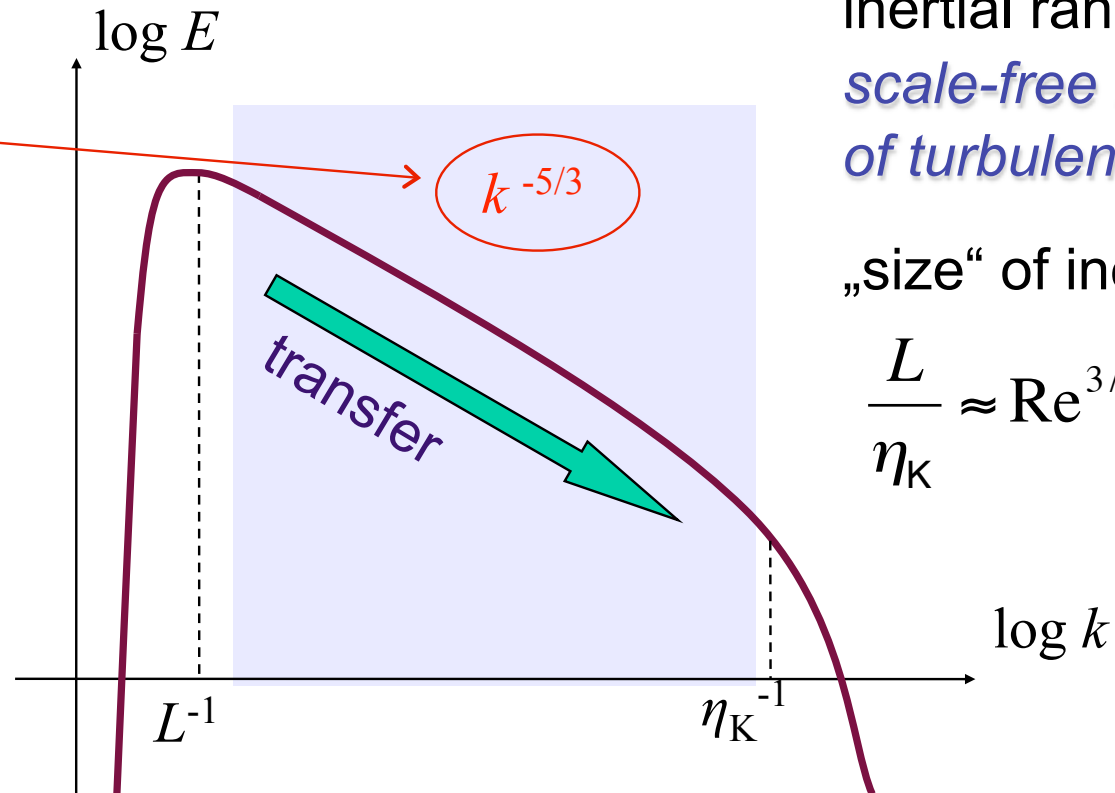
- interstellar gas is highly *inhomogeneous*
 - *gravitational instability*
 - *thermal instability*
 - *turbulent compression* (in shocks $\delta\rho/\rho \propto M^2$; in atomic gas: $M \approx 1...3$)
- cold *molecular clouds* can form rapidly in high-density regions at *stagnation points of convergent large-scale flows*
 - chemical *phase transition*: atomic \rightarrow molecular
 - process is *modulated* by large-scale *dynamics* in the galaxy
- inside *cold clouds*: turbulence is highly supersonic ($M \approx 1...20$)
 \rightarrow *turbulence* creates large density contrast,
gravity selects for collapse

 \longrightarrow **GRAVOTUBULENT FRAGMENTATION**
- *turbulent cascade*: local compression *within* a cloud provokes collapse
 \rightarrow formation of individual *stars* and *star clusters*



Turbulent cascade

Kolmogorov (1941) theory
incompressible turbulence



inertial range:
*scale-free behavior
of turbulence*

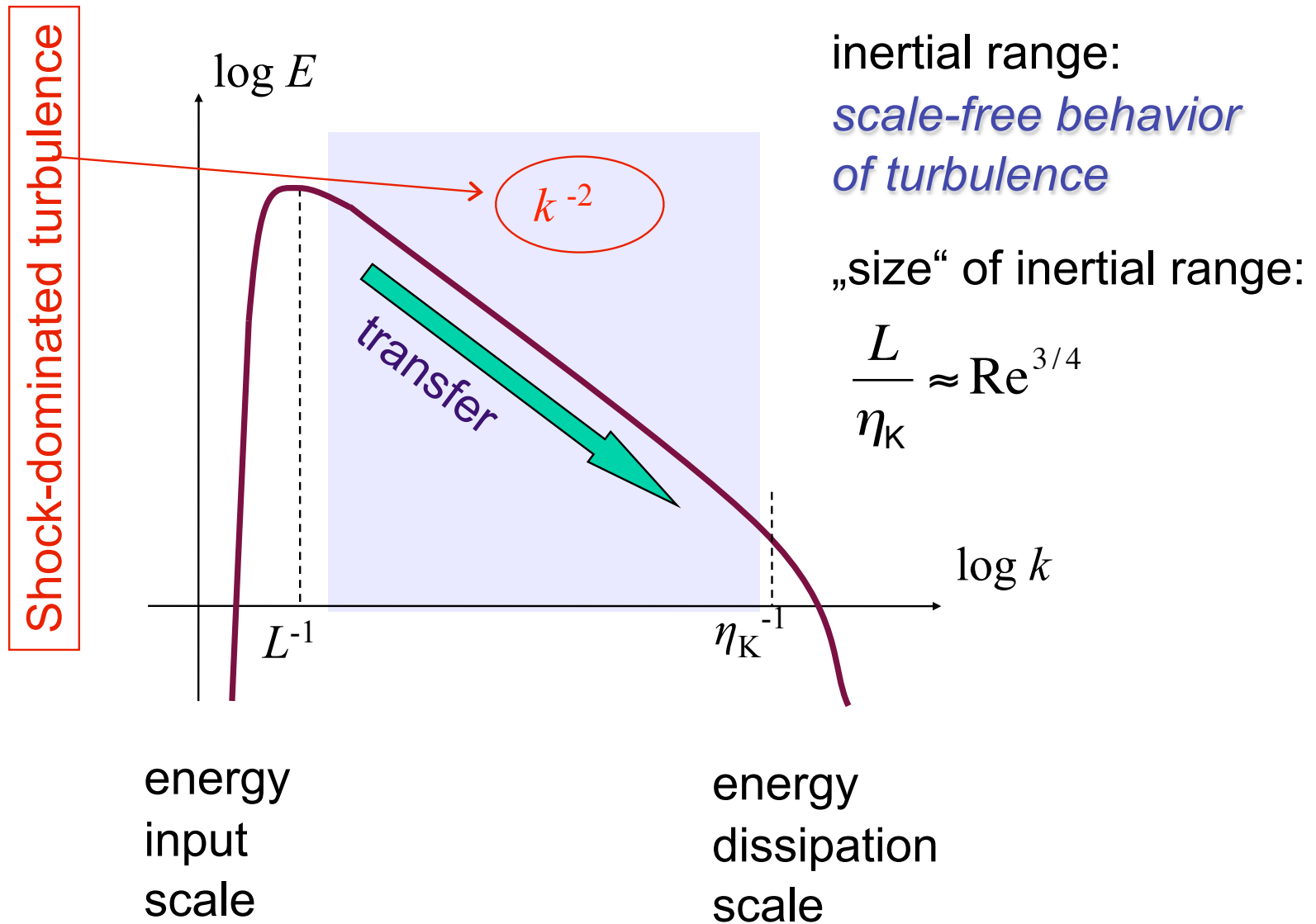
„size“ of inertial range:

$$\frac{L}{\eta_K} \approx \text{Re}^{3/4}$$

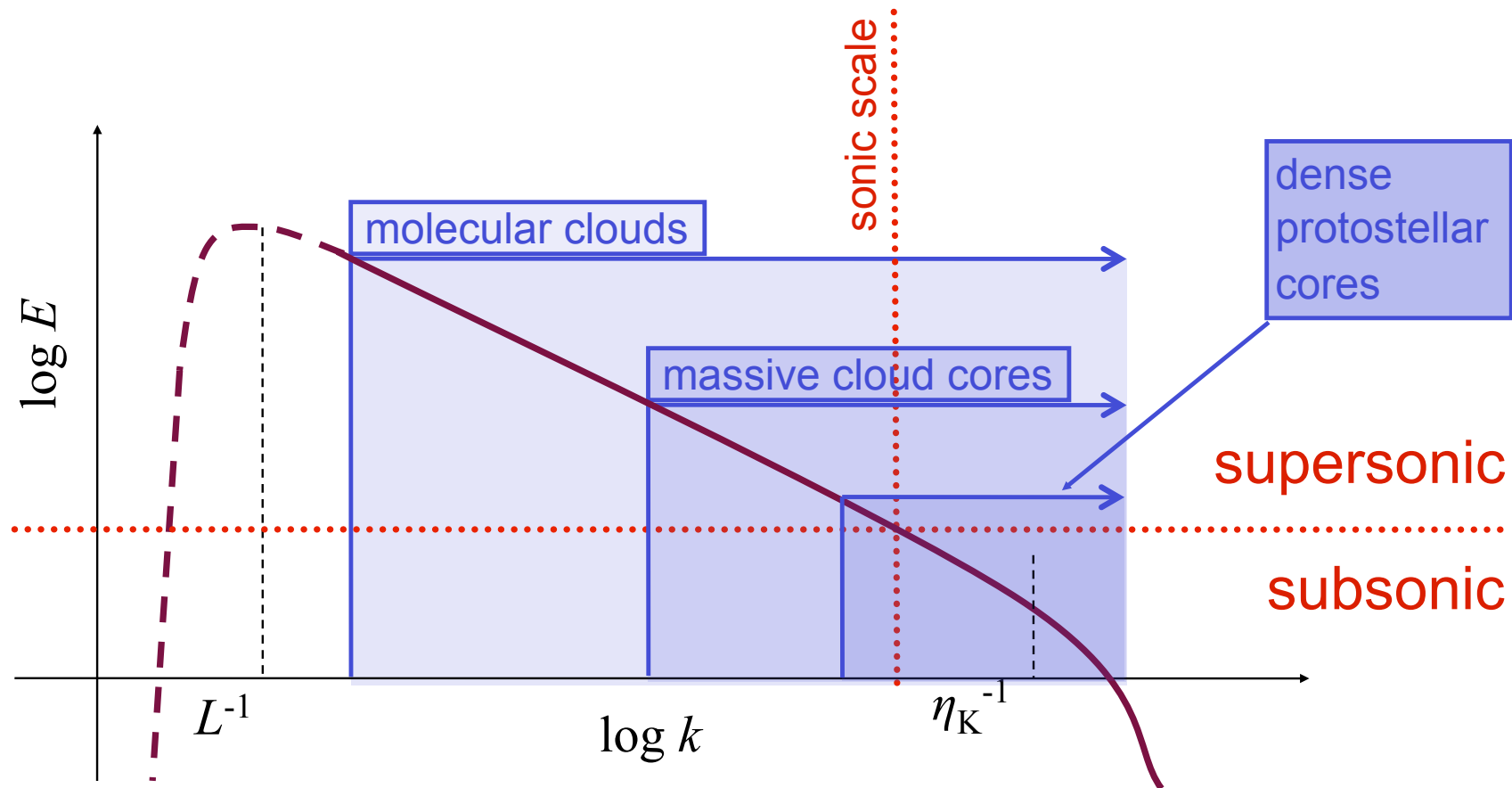
energy
input
scale

energy
dissipation
scale

Turbulent cascade



Turbulent cascade in ISM



energy source & scale
NOT known
 (supernovae, winds,
 spiral density waves?)

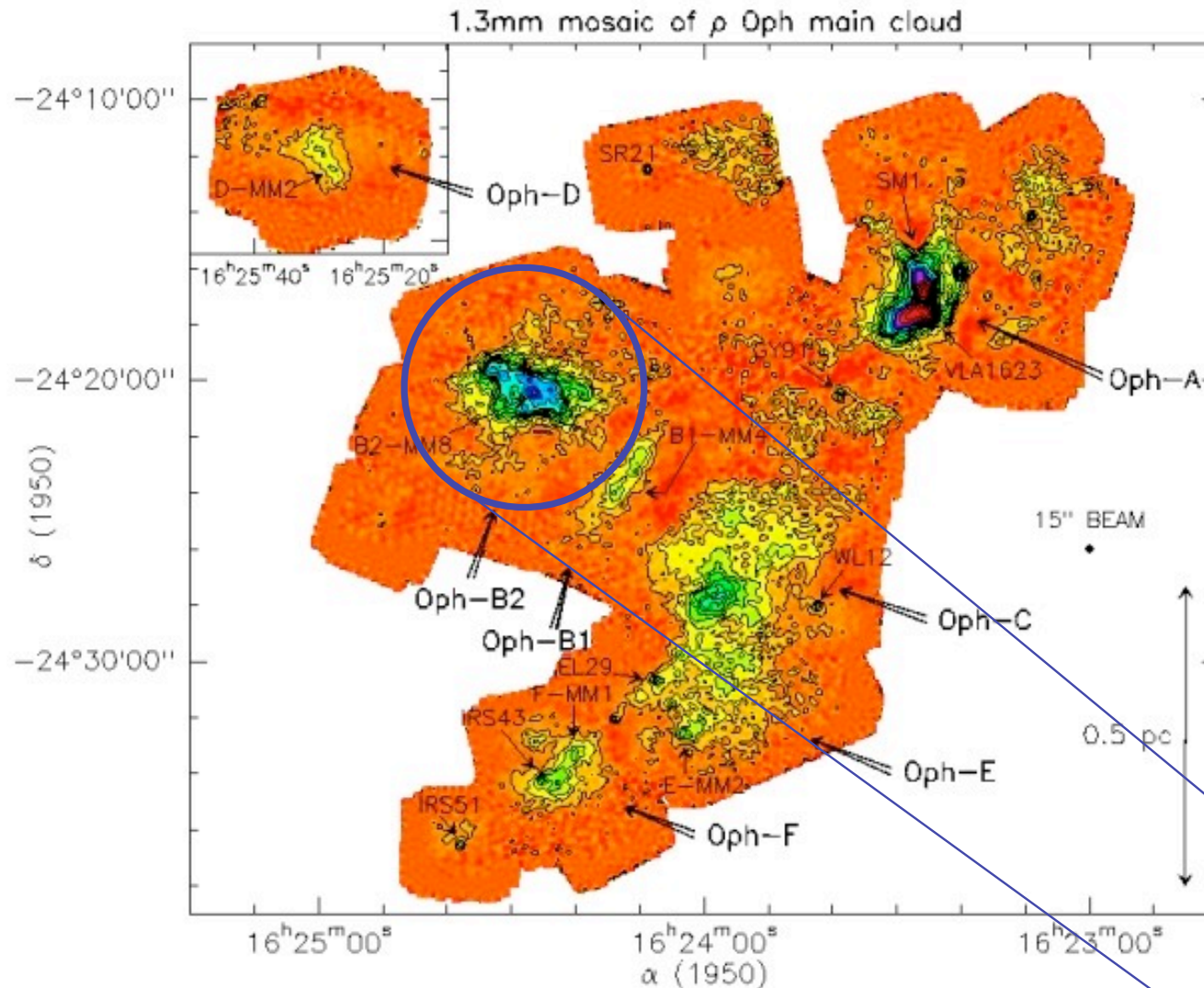
$$\sigma_{\text{rms}} \ll 1 \text{ km/s}$$

$$M_{\text{rms}} \leq 1$$

$$L \approx 0.1 \text{ pc}$$

dissipation scale not known
 (ambipolar diffusion,
 molecular diffusion?)

Density structure of MC's



(Motte, André, & Neri 1998)

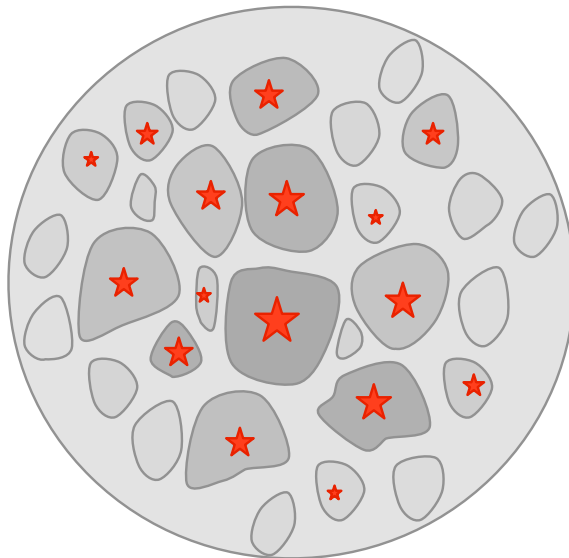
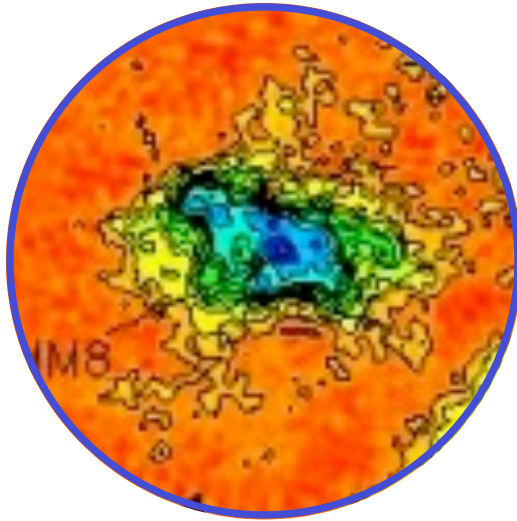
molecular clouds
are highly
inhomogeneous

stars form in the
densest and
coldest parts of
the cloud

ρ -Ophiuchus
cloud seen in dust
emission

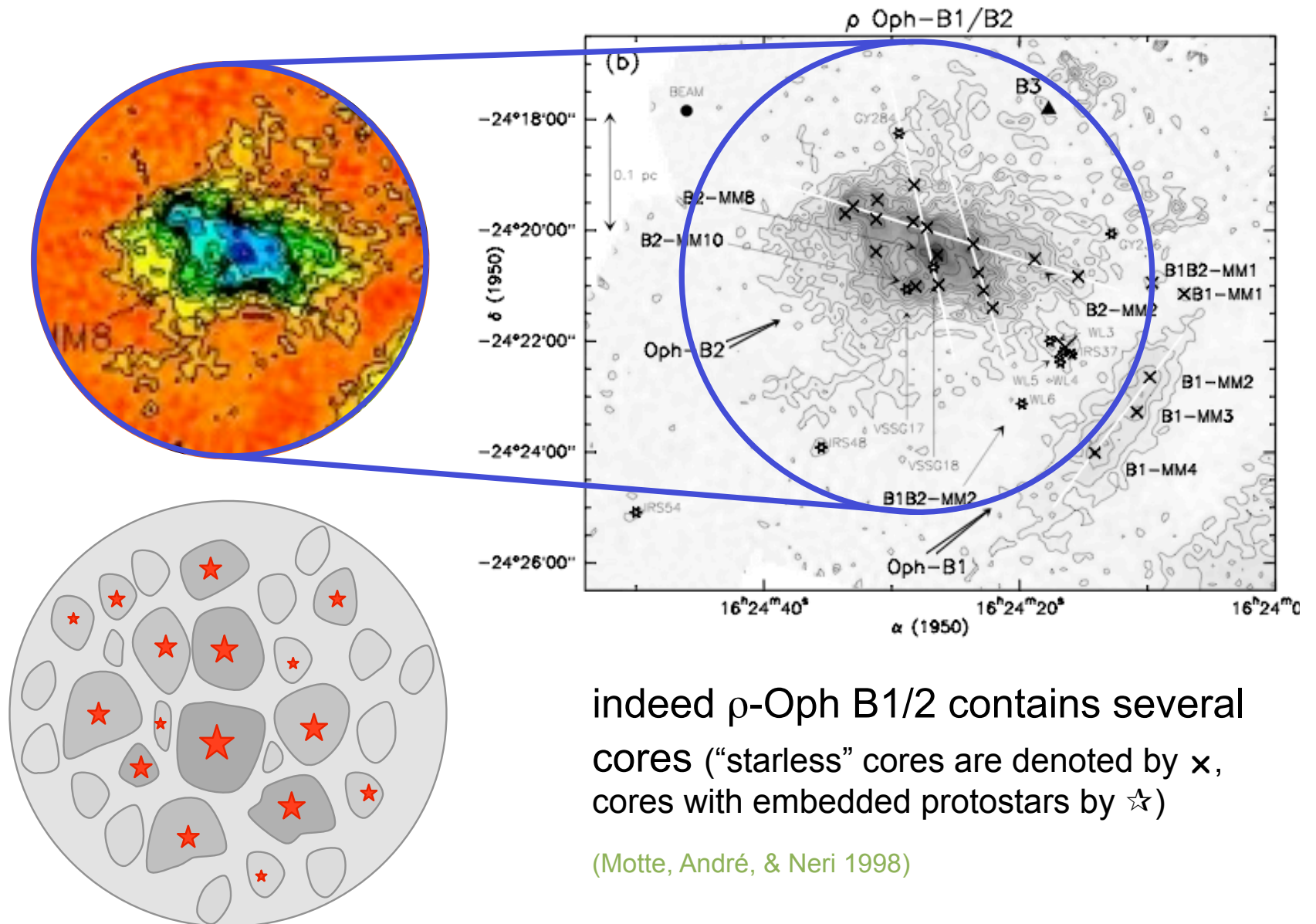
let's focus on
a cloud core
like this one

Evolution of cloud cores



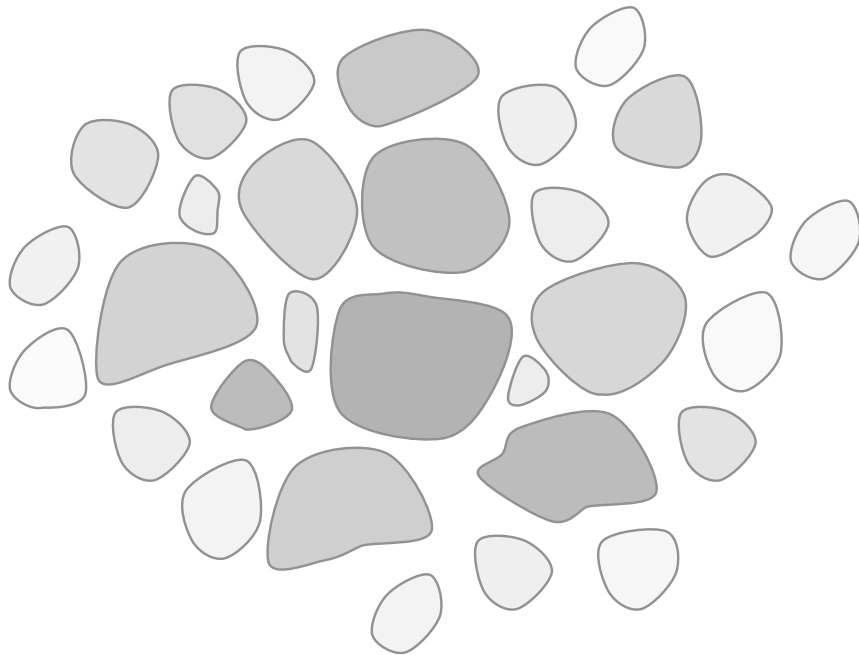
- How does this core evolve?
Does it form one single massive star or cluster with mass distribution?
- Turbulent cascade „goes through“ cloud core
--> NO *scale separation* possible
--> NO *effective sound speed*
- Turbulence is supersonic!
--> produces strong density contrasts:
 $\delta\rho/\rho \approx M^2$
--> with typical $M \approx 10$ --> $\delta\rho/\rho \approx 100!$
- many of the shock-generated fluctuations are Jeans unstable and go into collapse
- --> expectation: *core breaks up and forms a cluster of stars*

Evolution of cloud cores



Formation and evolution of cores

What happens to distribution of cloud cores?



Two extreme cases:

(1) turbulence dominates energy budget:

$$\alpha = E_{\text{kin}} / |E_{\text{pot}}| > 1$$

--> individual cores do *not* interact

--> *collapse of individual cores*
dominates *stellar mass growth*

--> *loose cluster of low-mass stars*

(2) turbulence decays, i.e. gravity

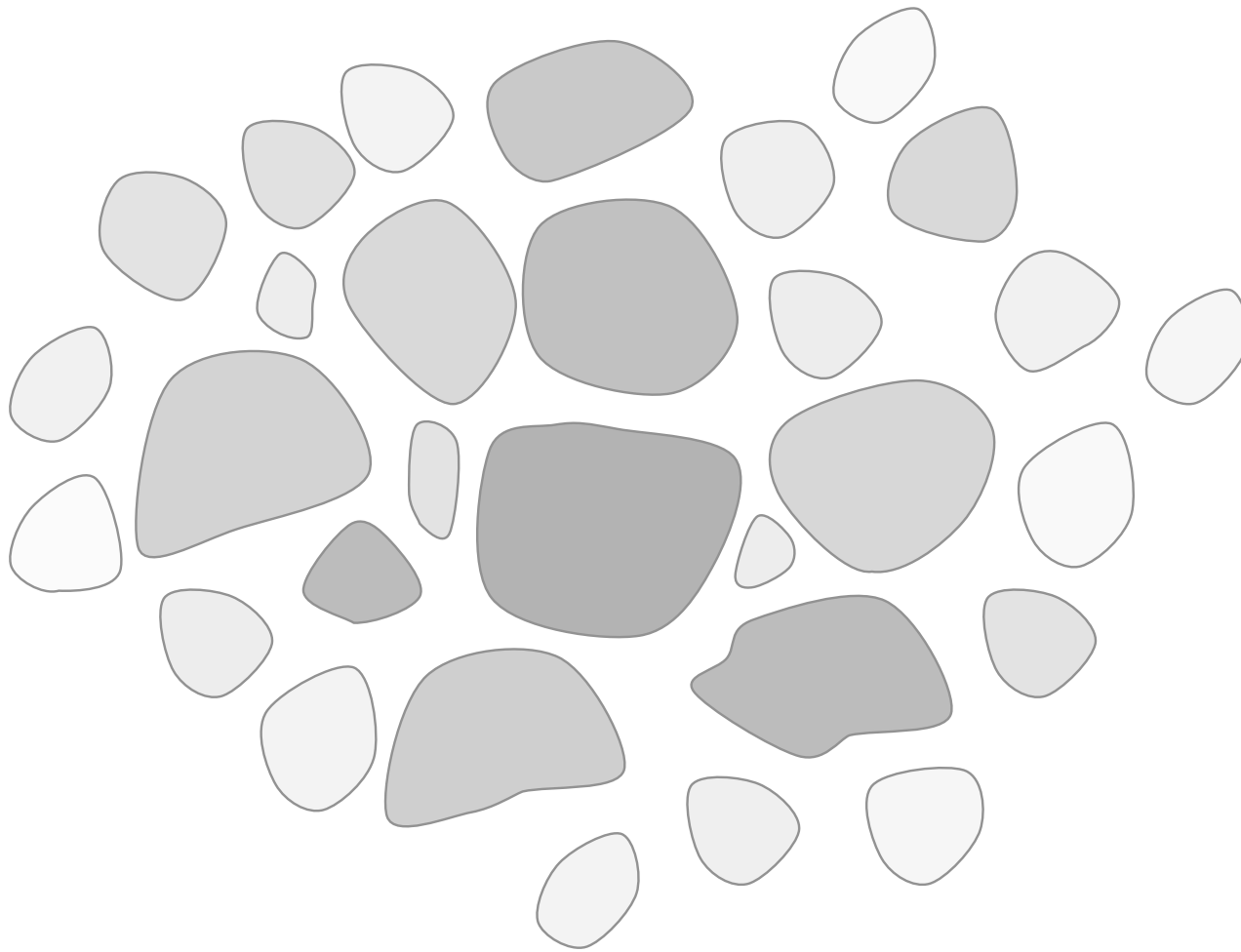
dominates: $\alpha = E_{\text{kin}} / |E_{\text{pot}}| < 1$

--> *global contraction*

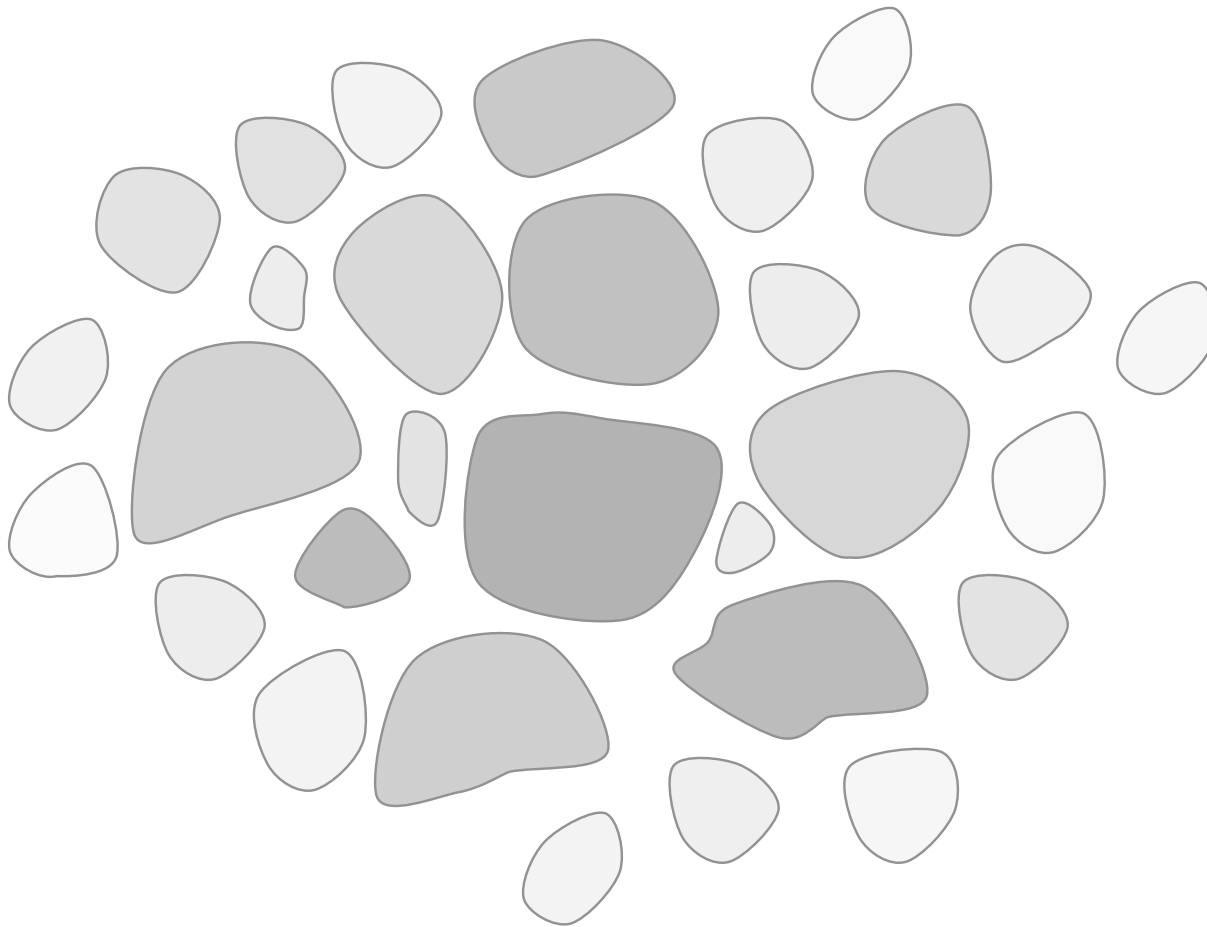
--> core do *interact* while collapsing

--> *competition* influences *mass growth*

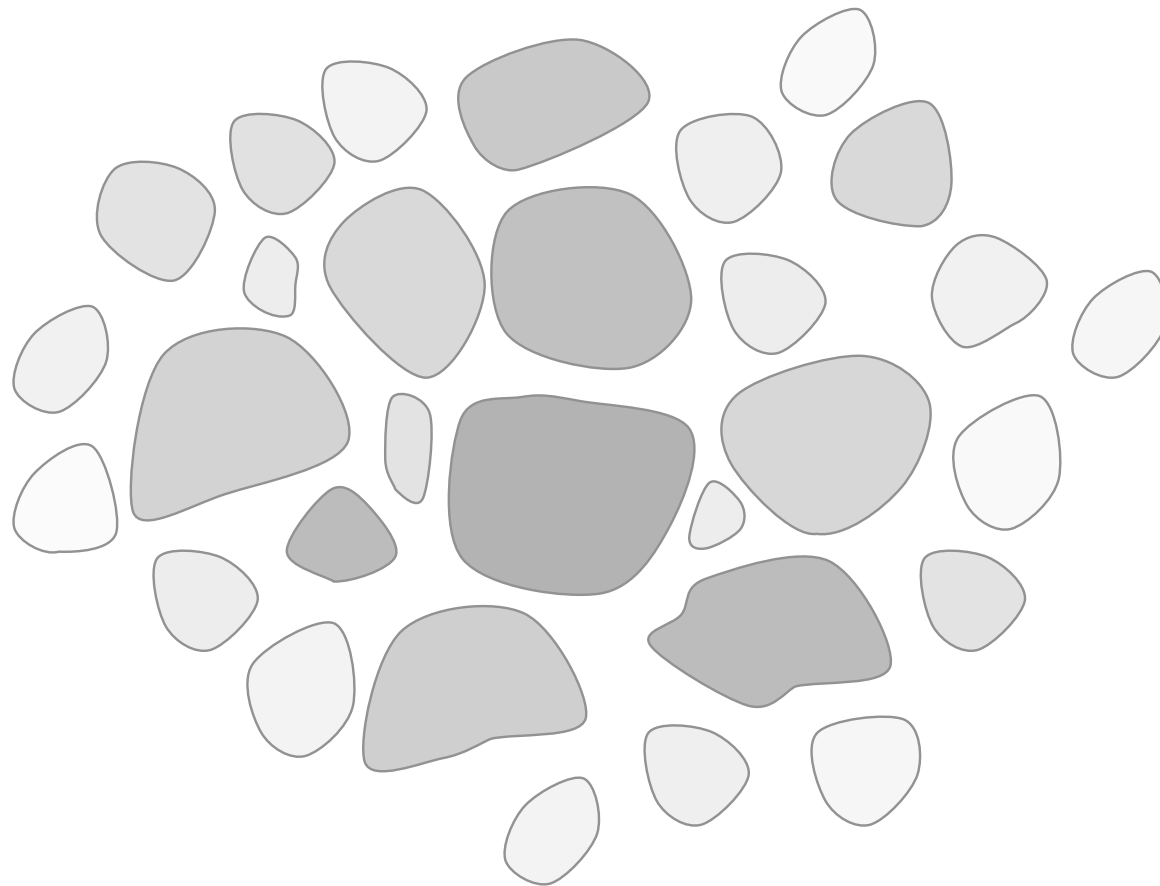
--> *dense cluster with high-mass stars*



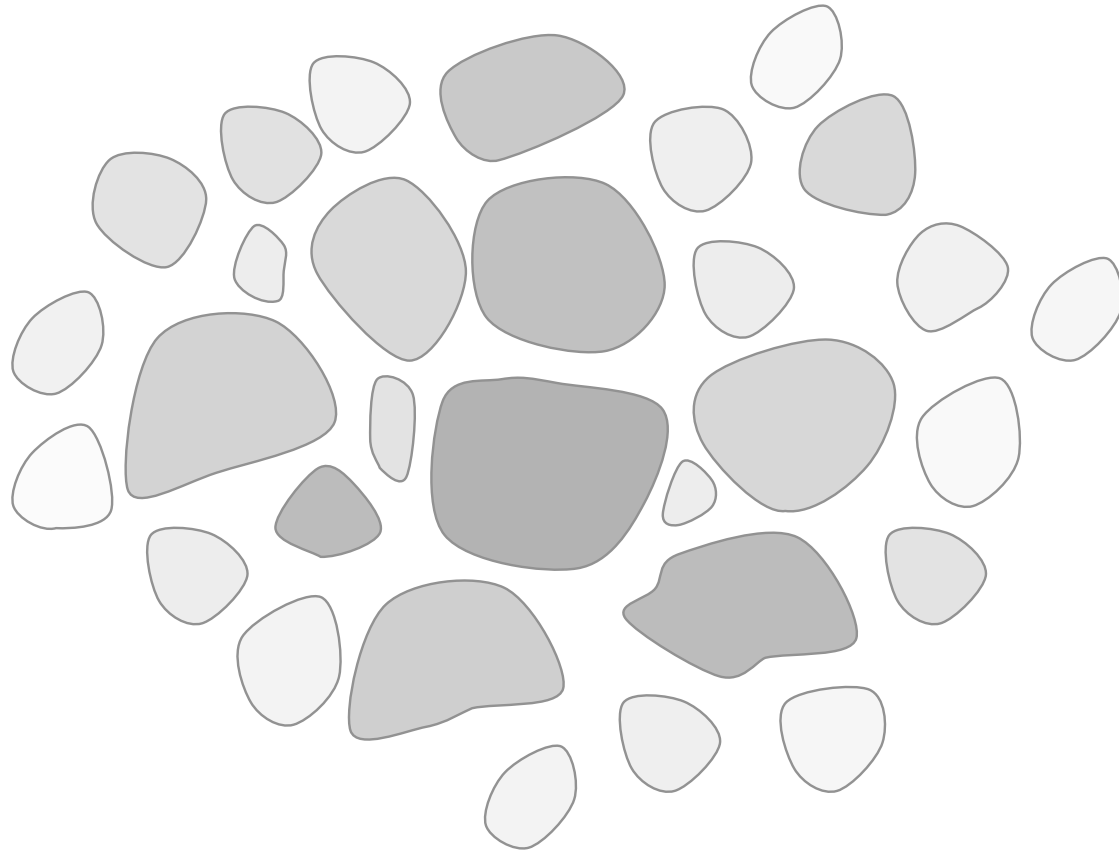
turbulence creates a hierarchy of clumps



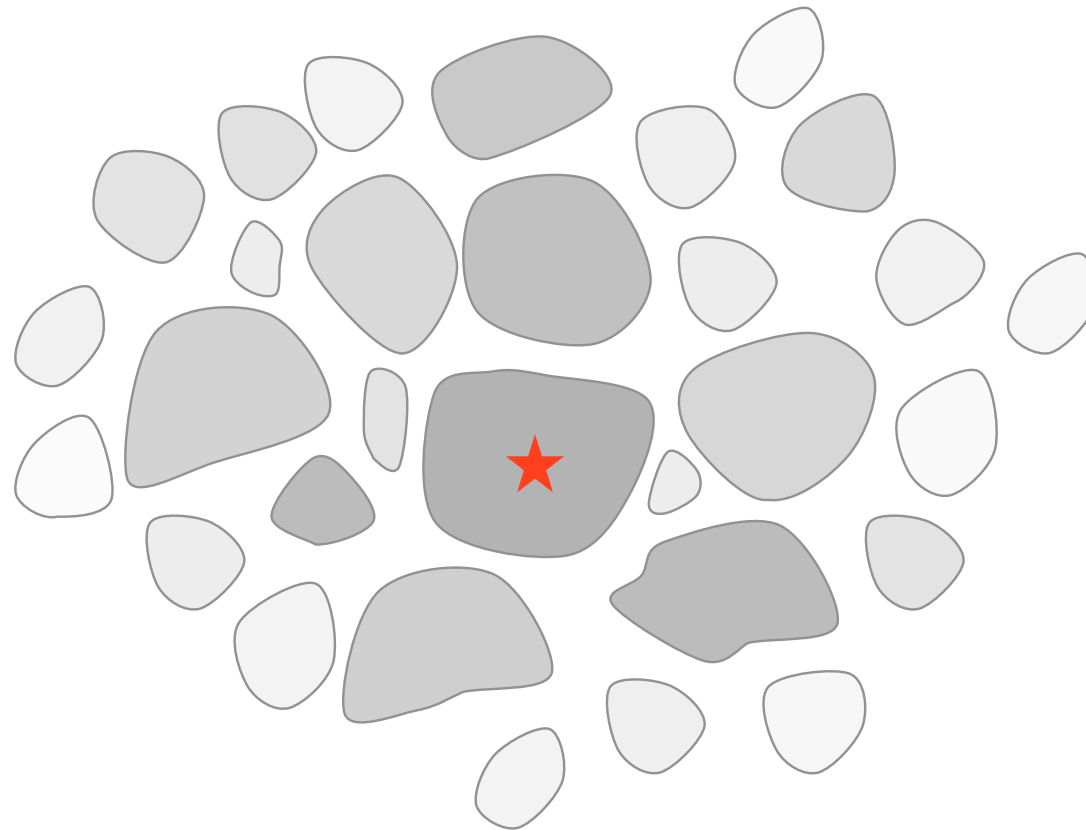
as turbulence decays locally, contraction sets in



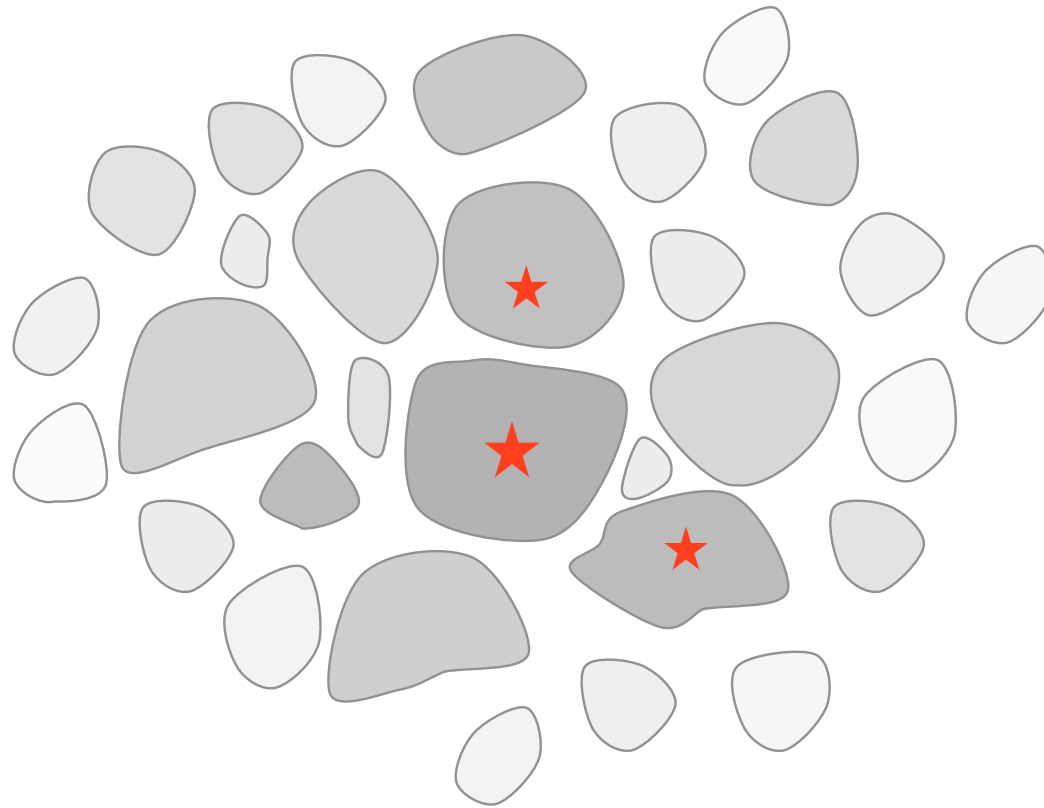
as turbulence decays locally, contraction sets in



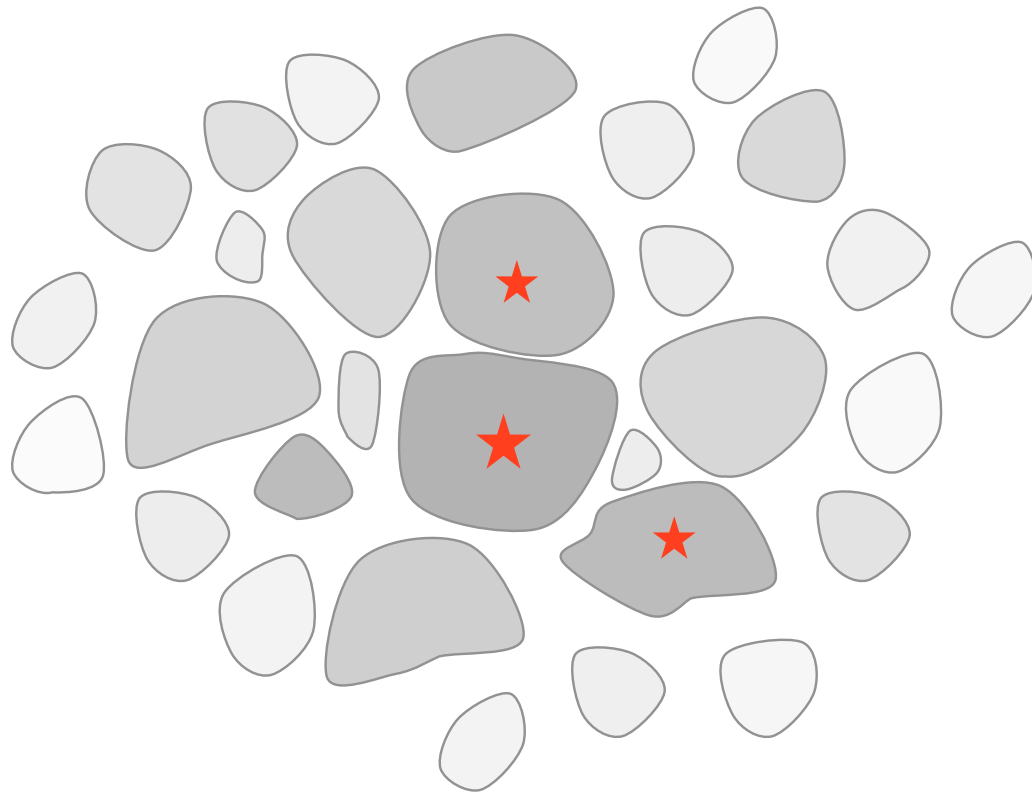
while region contracts, individual clumps collapse to form stars



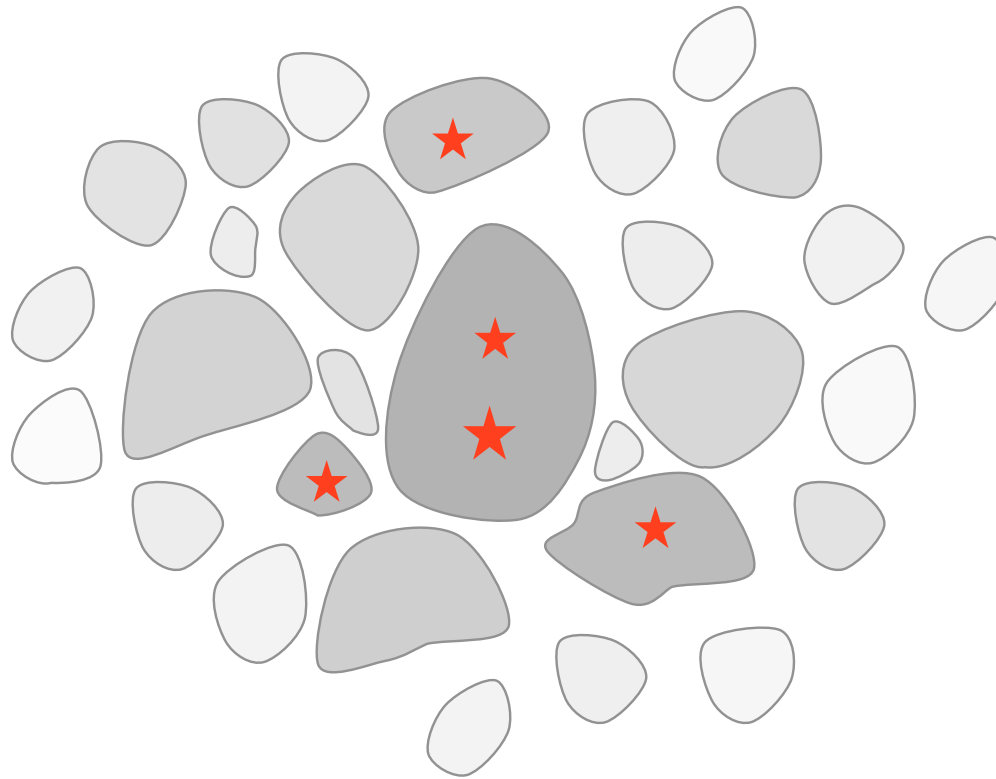
while region contracts, individual clumps collapse to form stars



individual clumps collapse to form stars

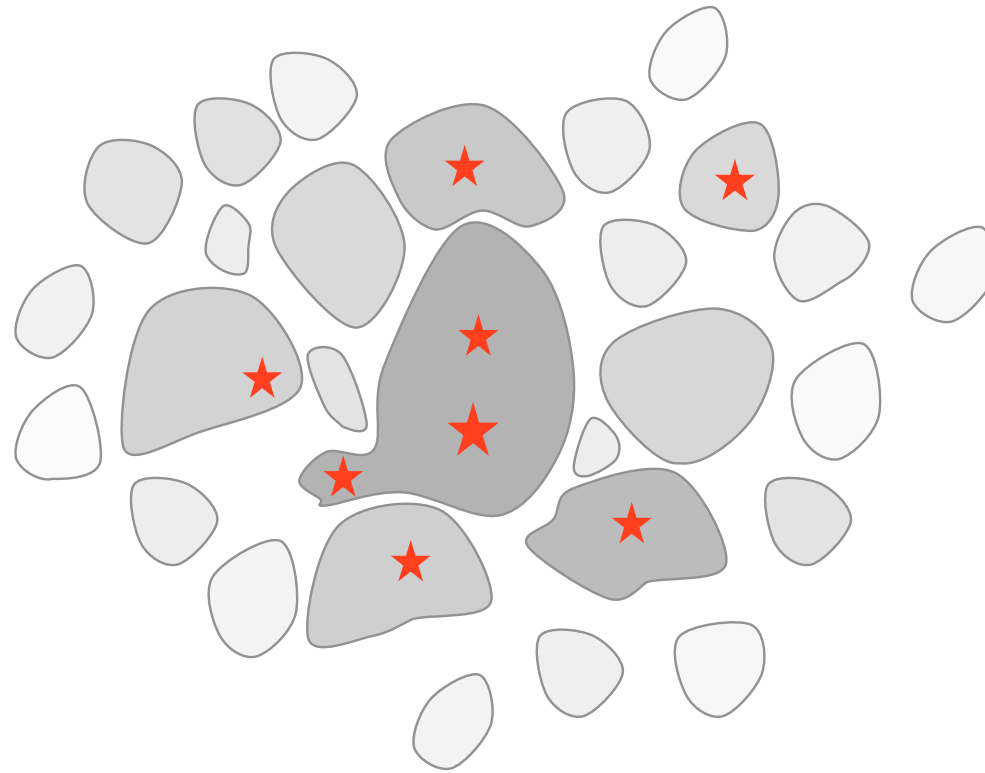


individual clumps collapse to form stars

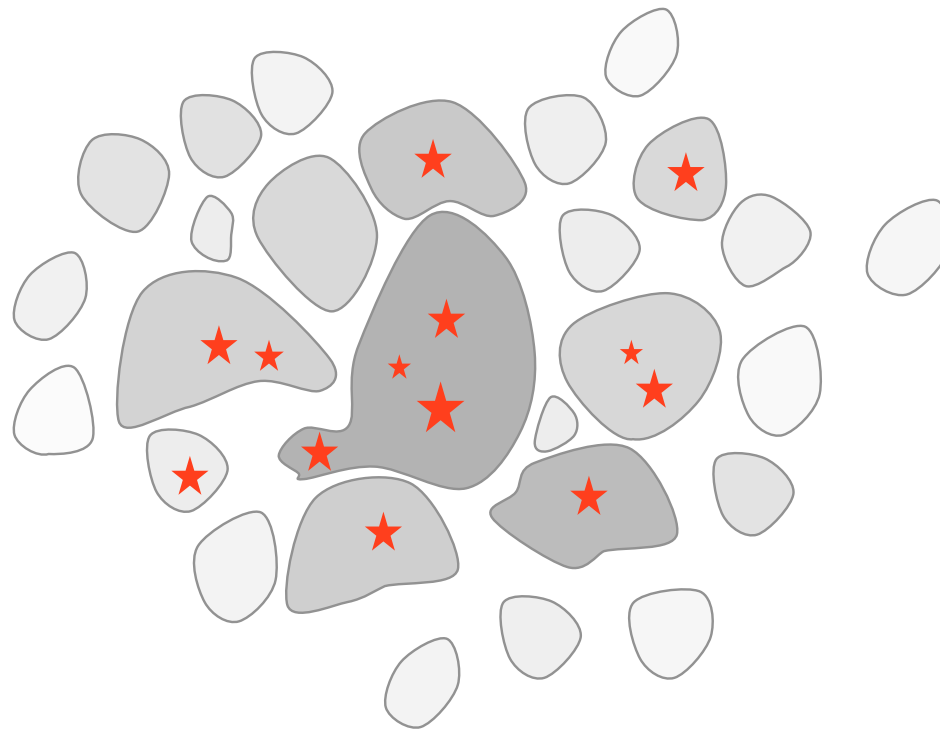


$$\alpha = E_{\text{kin}} / |E_{\text{pot}}| < 1$$

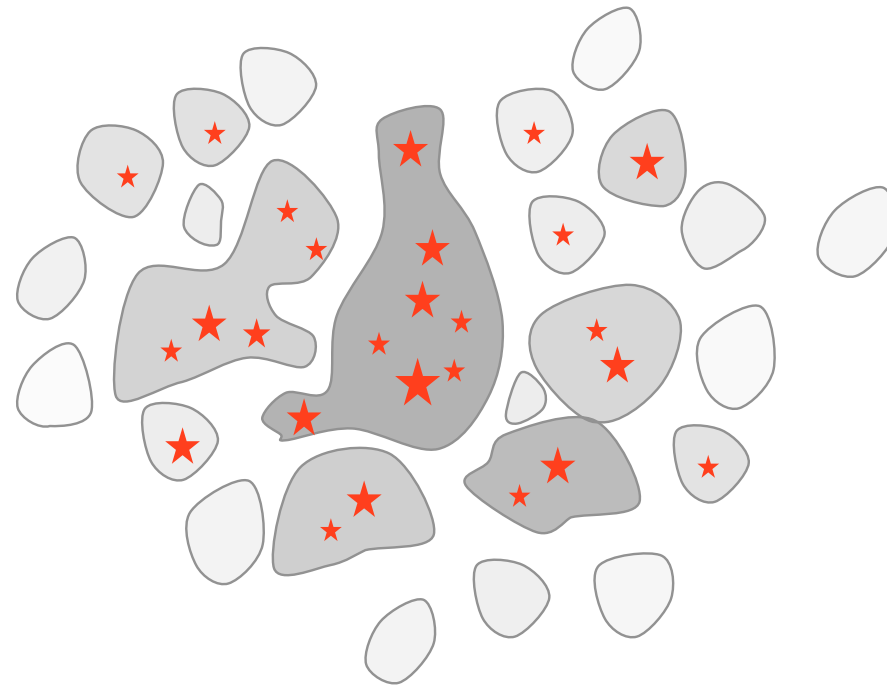
in *dense clusters*, clumps may merge while collapsing
--> then contain multiple protostars



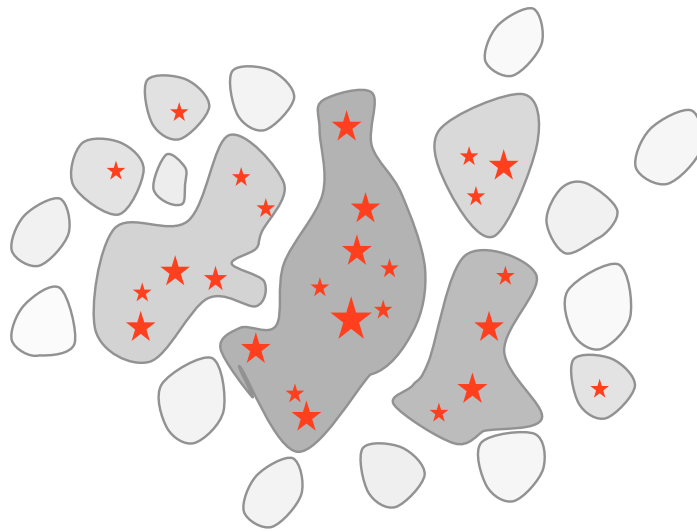
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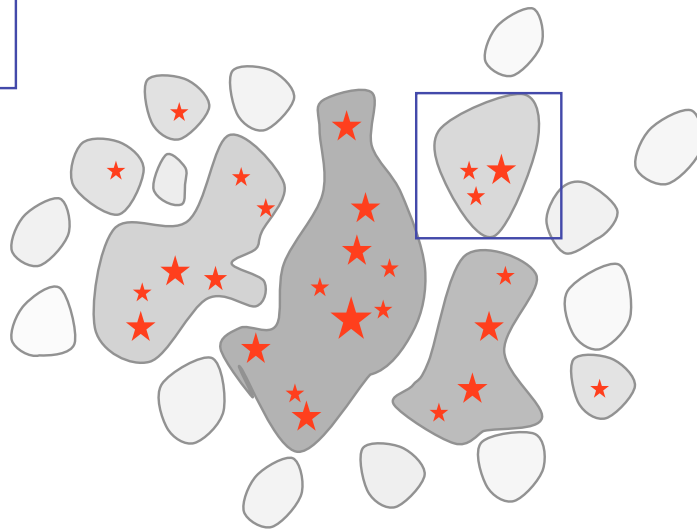
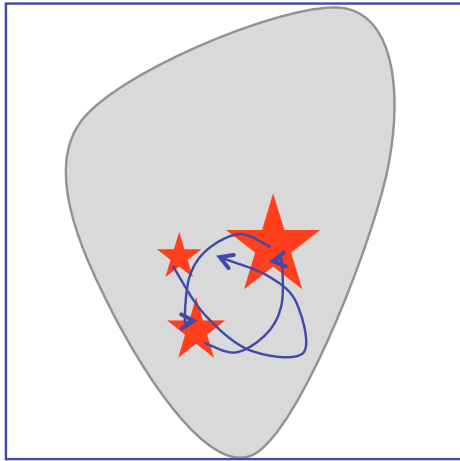
in *dense clusters*, clumps may merge while collapsing
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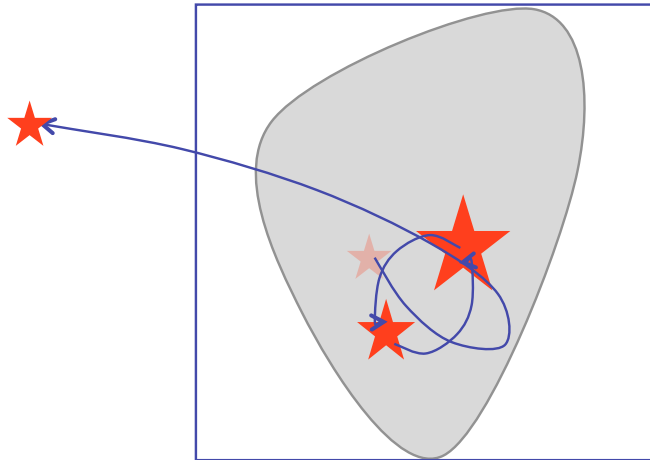
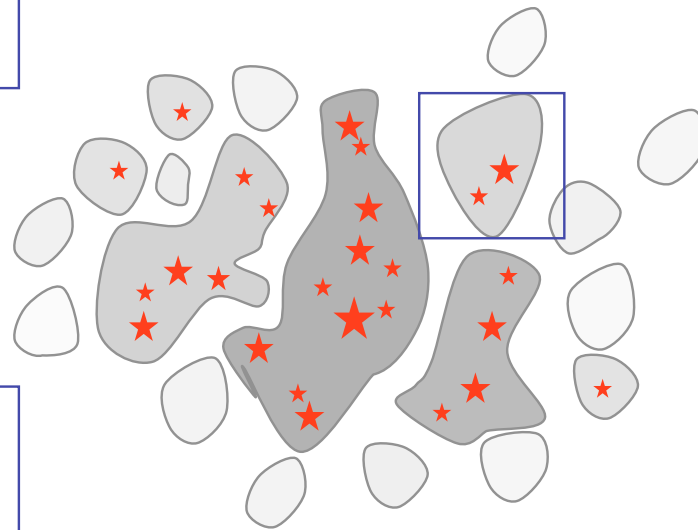
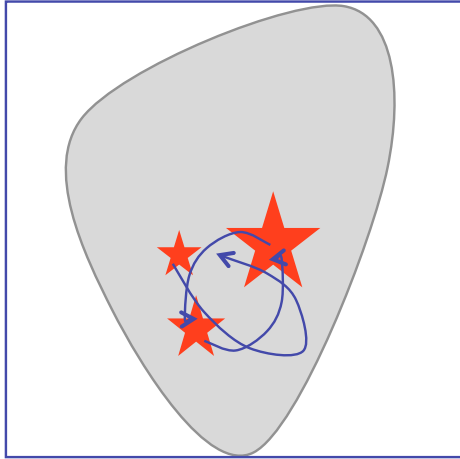
in *dense clusters*, competitive mass growth
becomes important



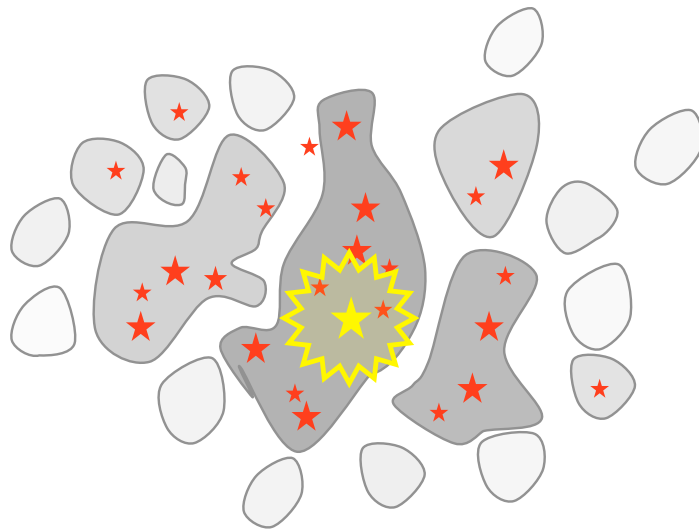
in *dense clusters*, competitive mass growth
becomes important



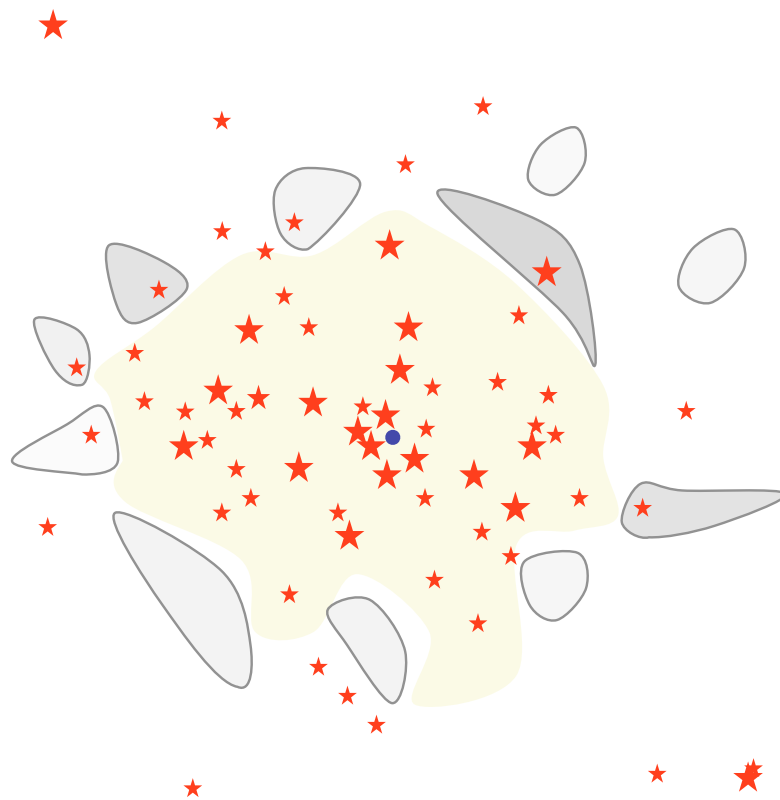
in *dense clusters*, N -body effects influence mass growth



low-mass objects may
become ejected --> accretion stops



feedback terminates star formation



result: *star cluster*, possibly with HII region



NGC 602 in the LMC: Hubble Heritage Image

result: *star cluster* with HII region



applications



two examples

- formation of molecular clouds in the disk of the Milky Way
 - timescales
 - dynamic properties
 - x-factor
- formation of star clusters inside these clouds
 - IMF



molecular cloud formation

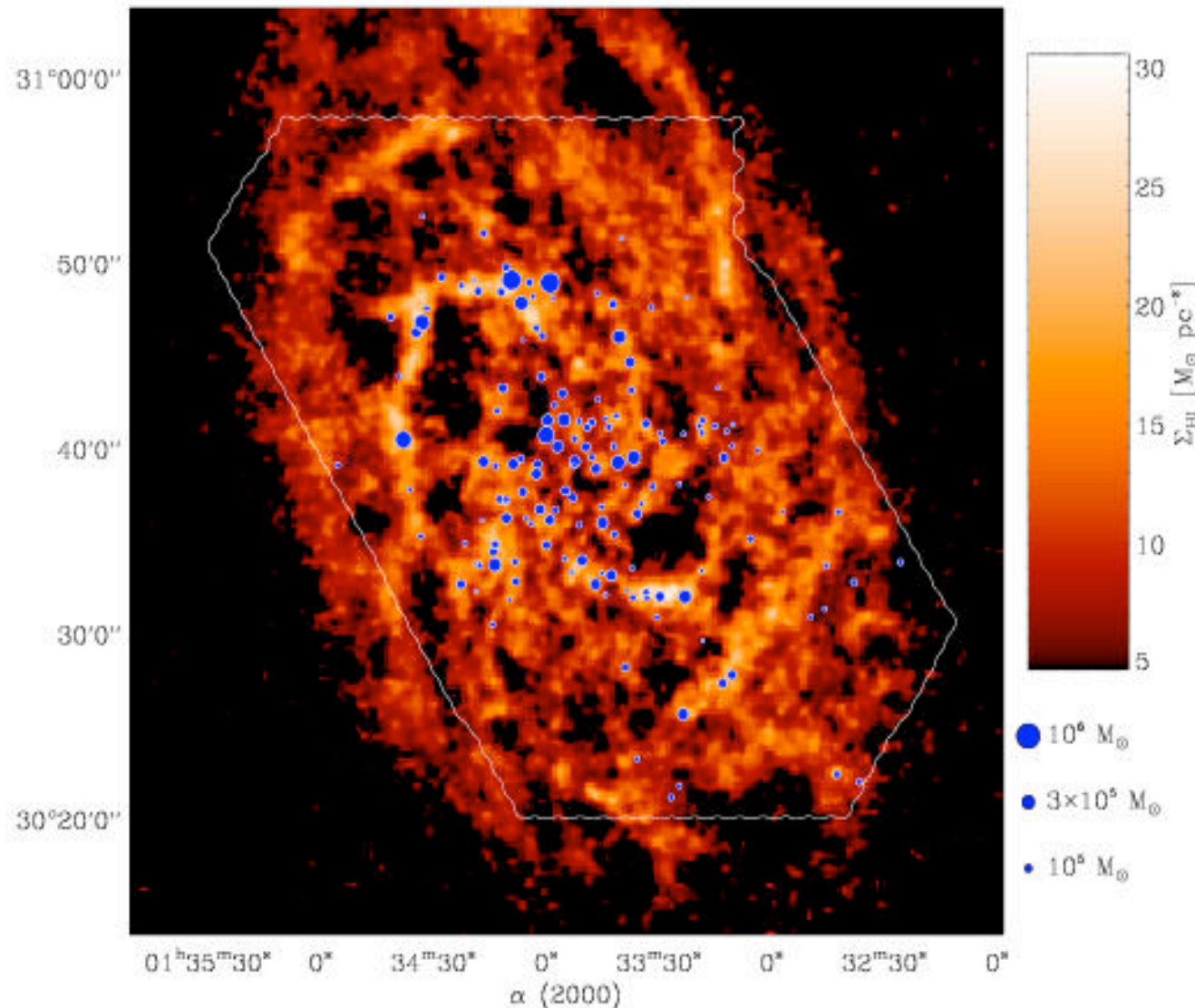


molecular cloud formation

- star formation on galactic scales
→ missing link so far:
formation of molecular clouds
- questions
 - *where* and *when* do molecular clouds form?
 - *what* are their properties?
 - *how* does that correlate to star formation?
 - global correlations? → *Schmidt law*



molecular cloud formation



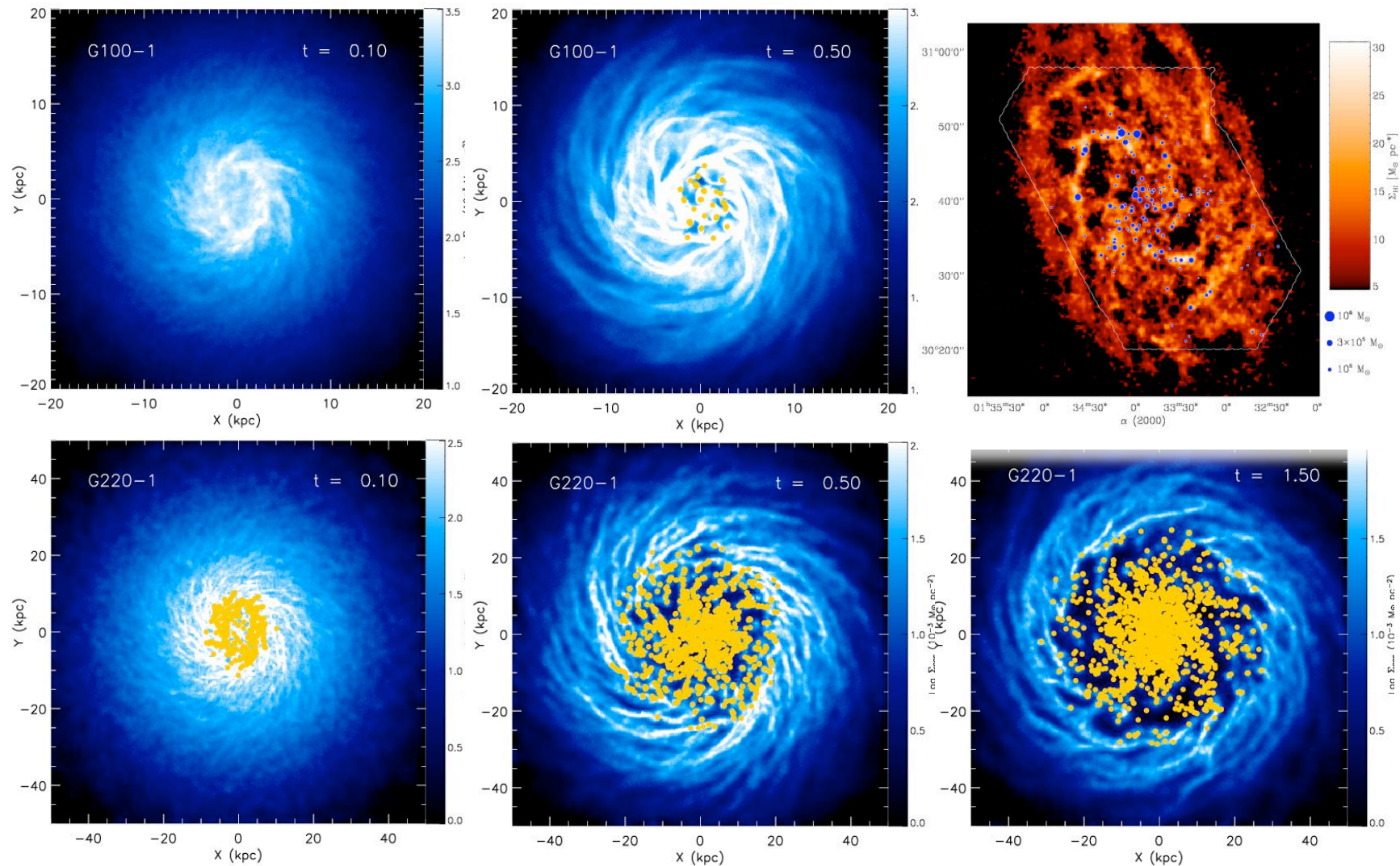
Thesis:

Molecular clouds form at *stagnation points* of large-scale convergent flows, mostly triggered by global (or external) perturbations.



modeling galactic SF

SPH calculations of self-gravitating disks of stars and (isothermal) gas in dark-matter potential, sink particles measure local collapse --> star formation



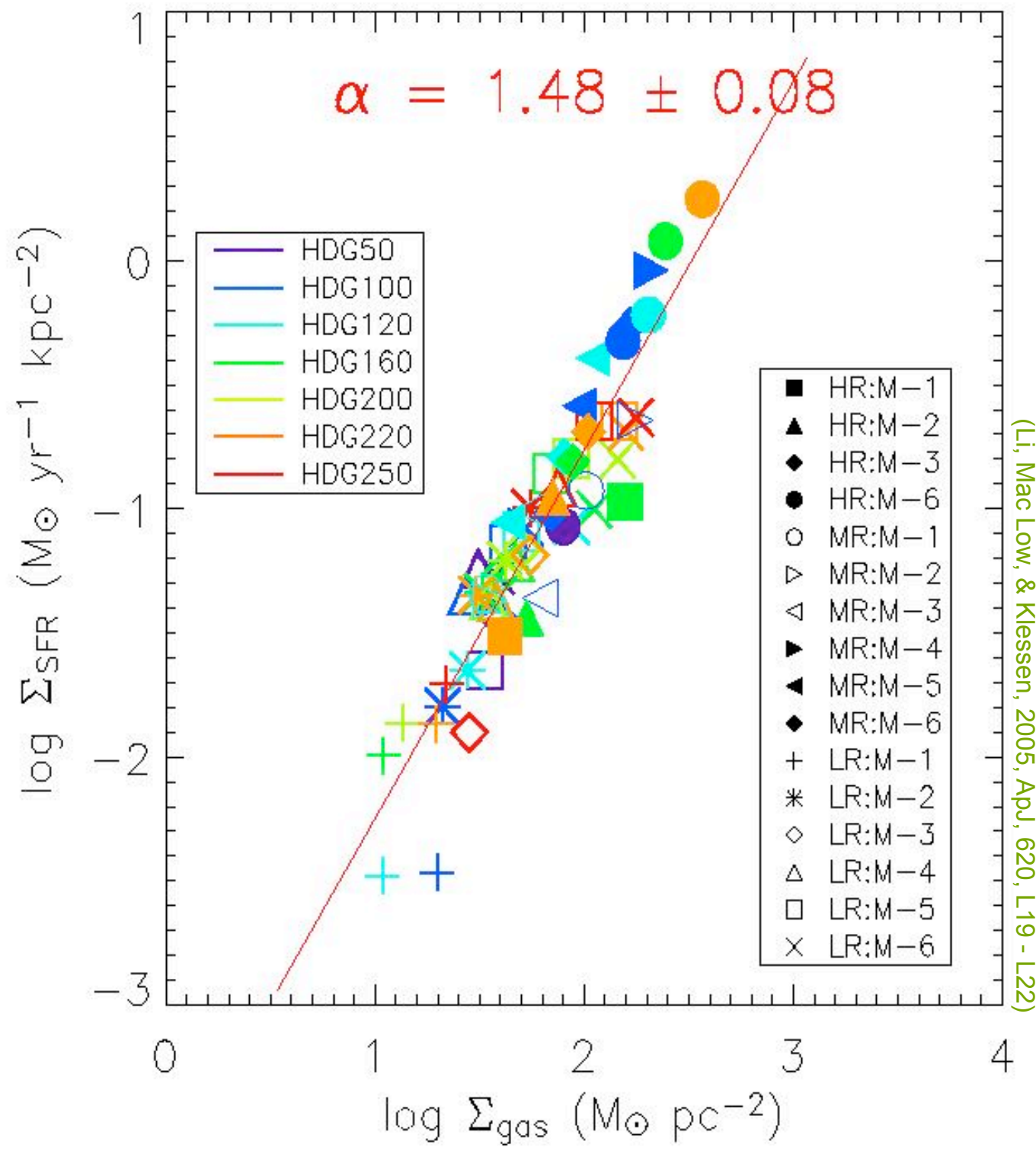
(Li, Mac Low, & Klessen, 2005, ApJ, 620, L19 - L22)



We find correlation between *star formation rate* and *gas surface density*:

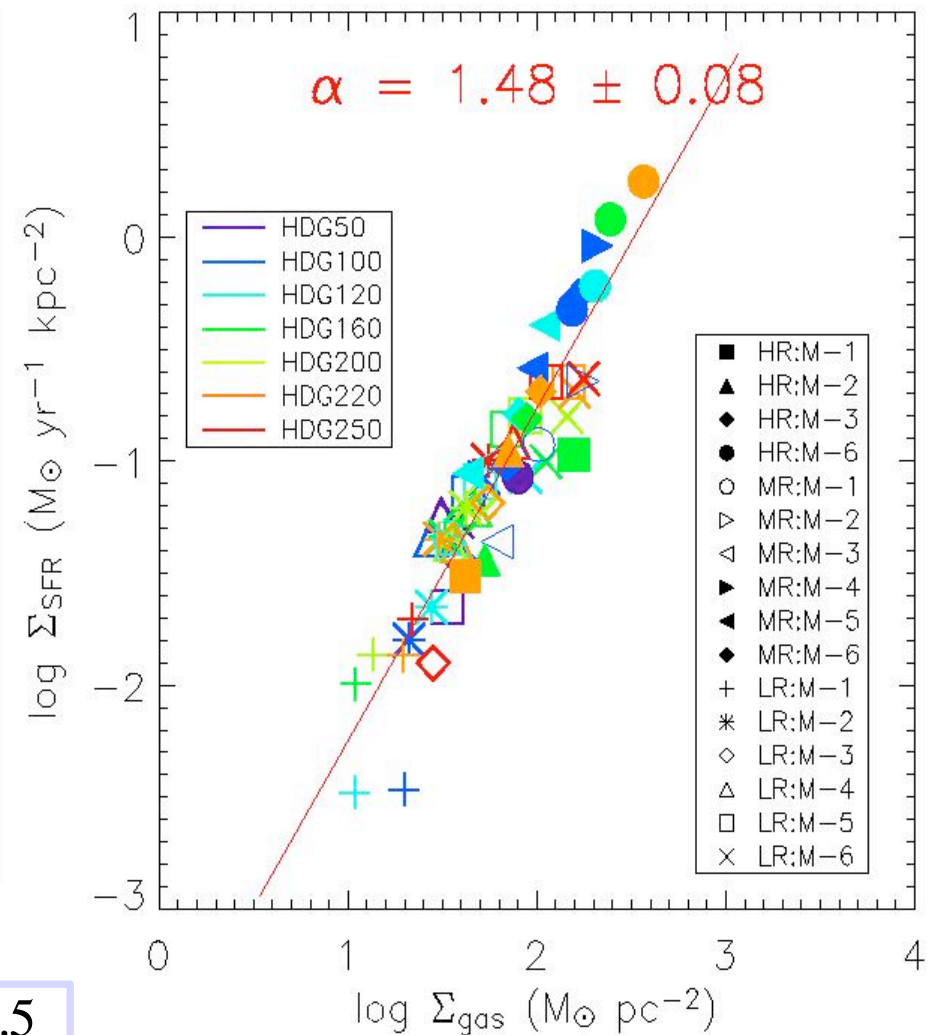
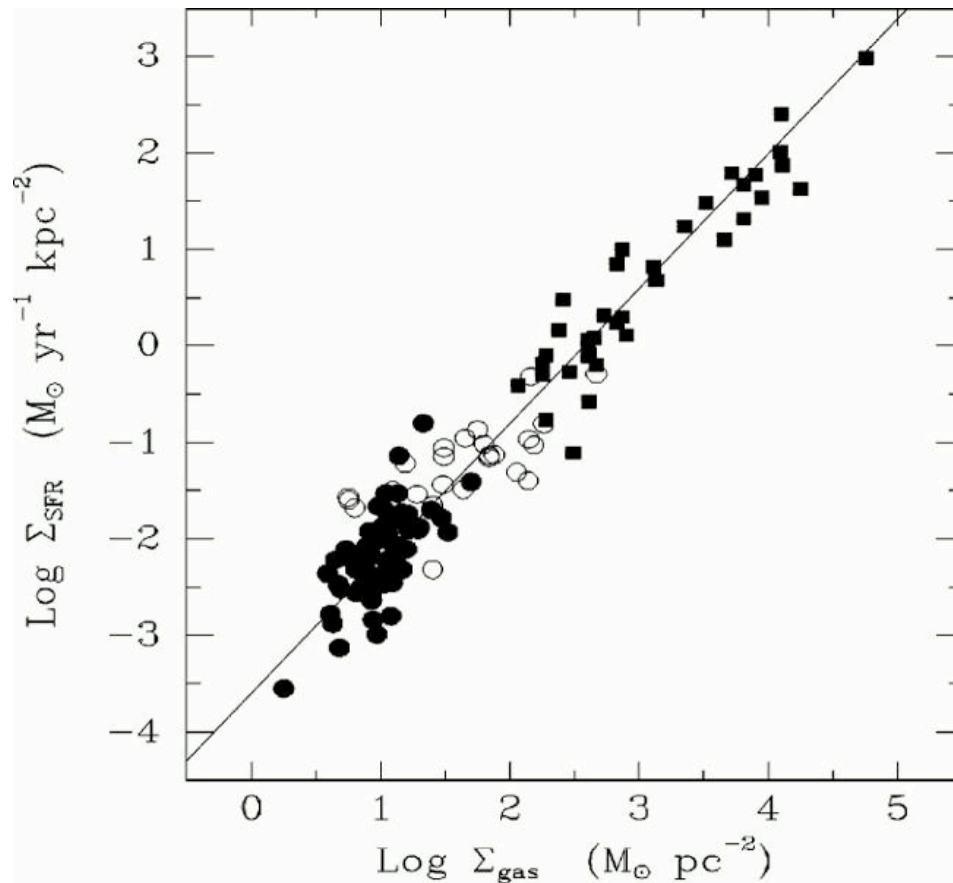
$$\Sigma_{\text{SFR}} \propto \Sigma_{\text{gas}}^{1.5}$$

global Schmidt law





observed Schmidt law



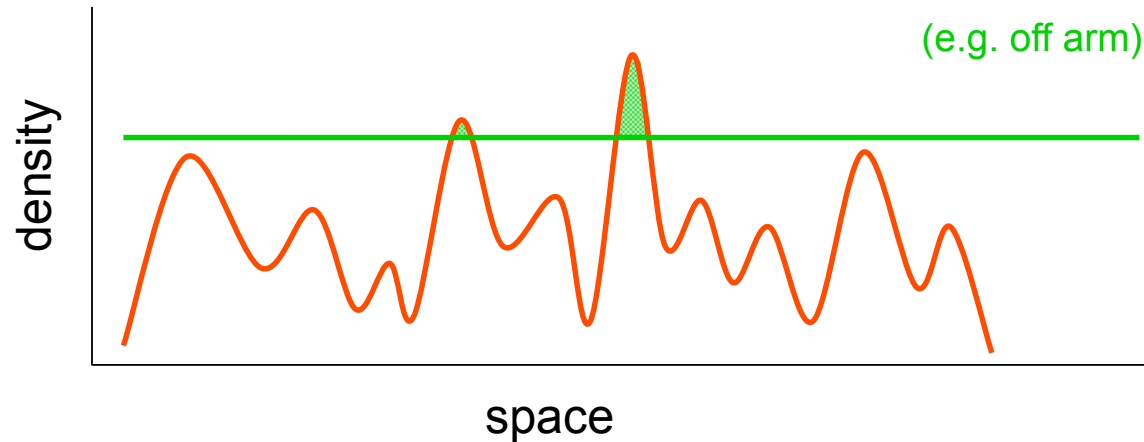
in both cases:

$$\Sigma_{\text{SFR}} \propto \Sigma_{\text{gas}}^{1.5}$$

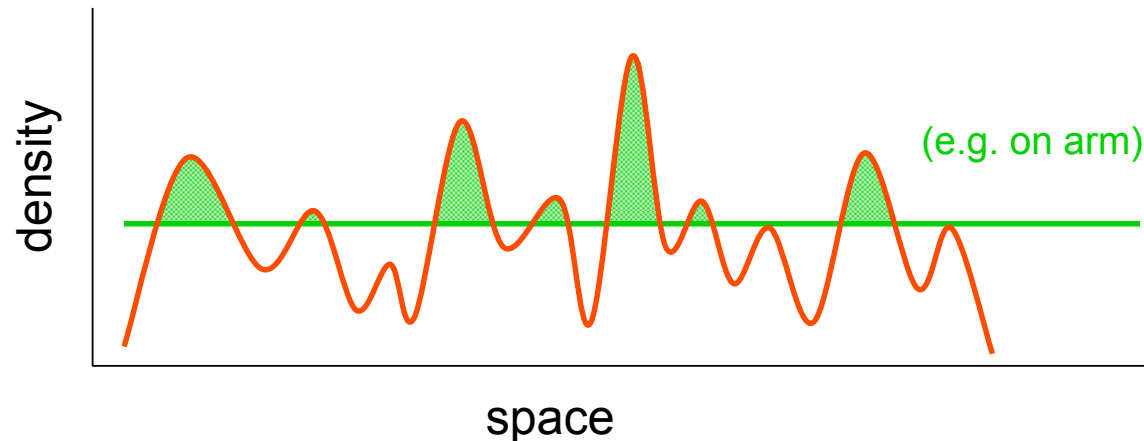
(from Kennicutt 1998)



correlation with large-scale perturbations



density/temperature fluctuations in warm atomic ISM are caused by *thermal/gravitational instability* and/or *supersonic turbulence*

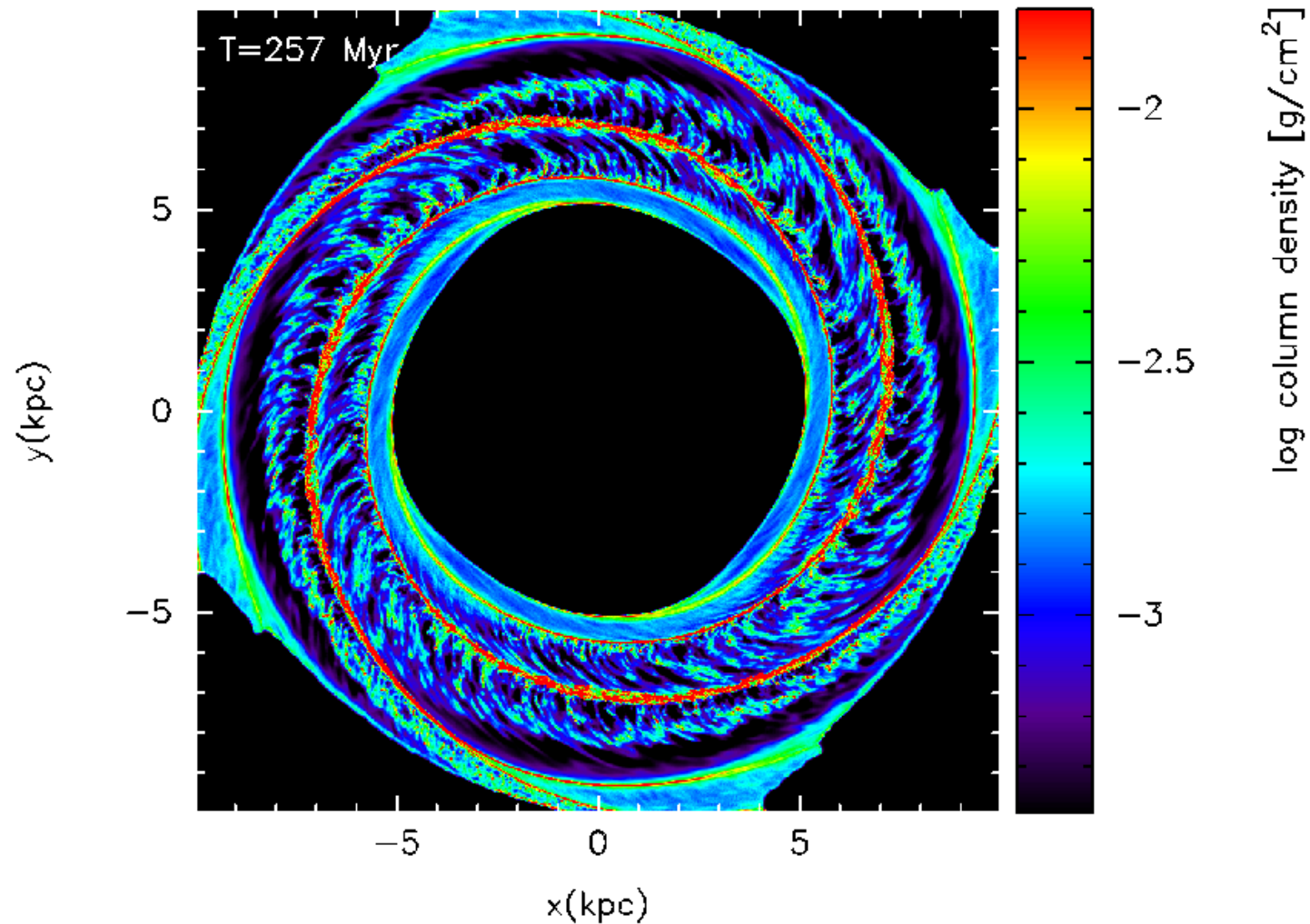


some fluctuations are *dense* enough to *form H_2* within “reasonable time”
→ *molecular cloud*

(Glover & Mac Low 2007a,b)
external perturbations (i.e. potential changes) *increase* likelihood



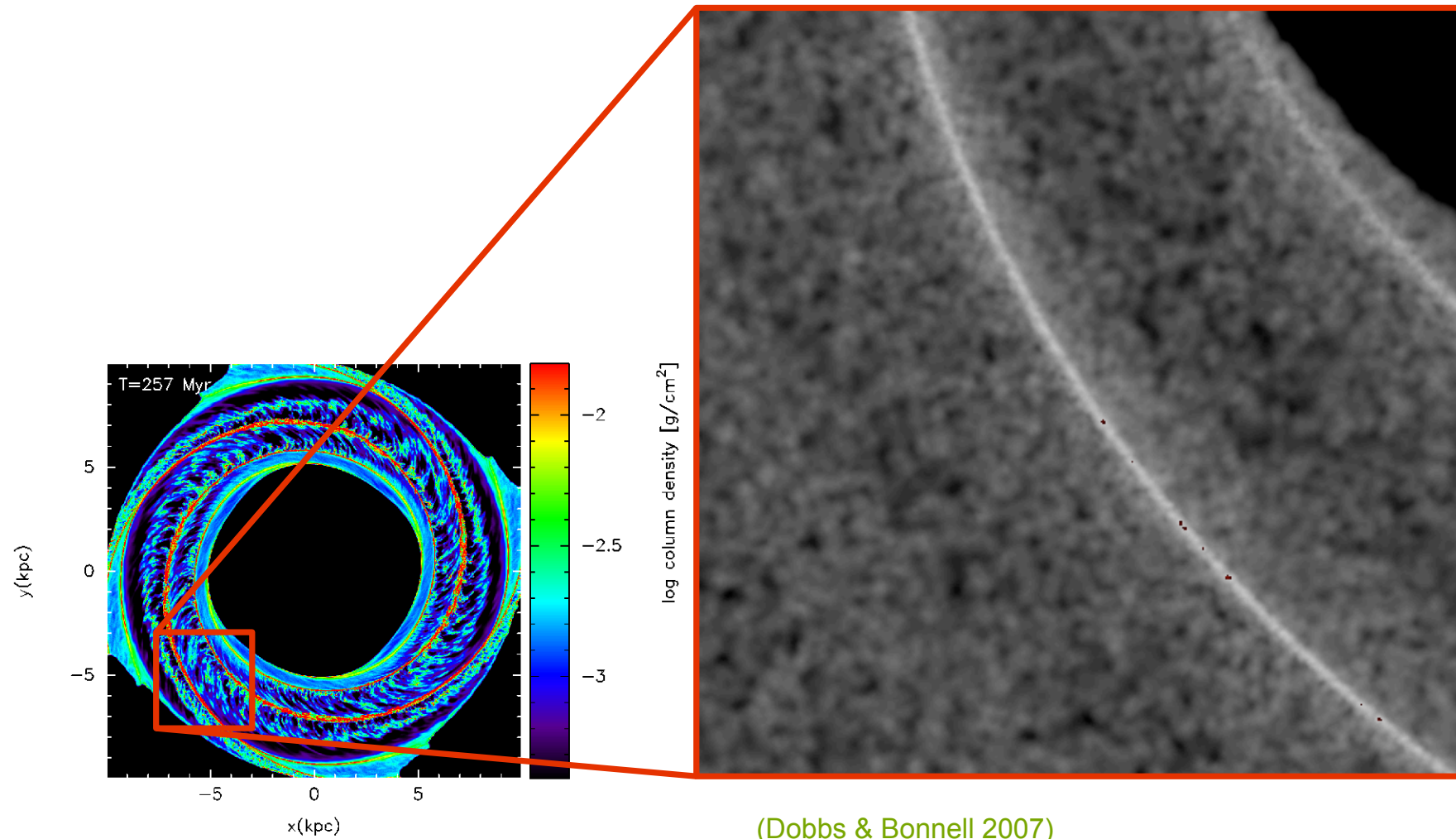
molecular cloud formation



(from Dobbs, Glover, Clark, Klessen 2008)



molecular cloud formation

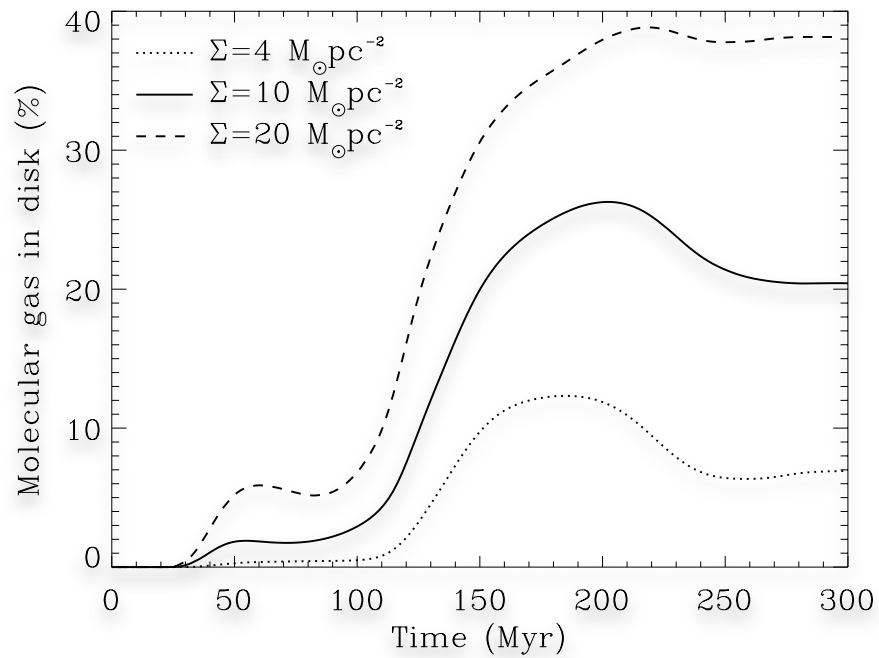


(Dobbs & Bonnell 2007)

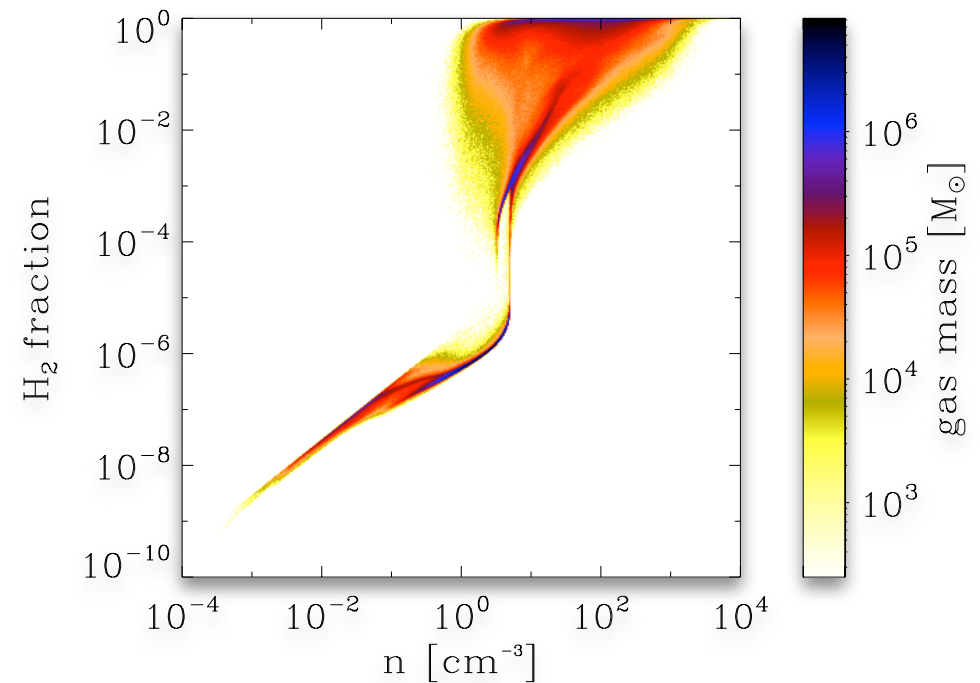


molecular cloud formation

molecular gas fraction as function of time



molecular gas fraction as function of density

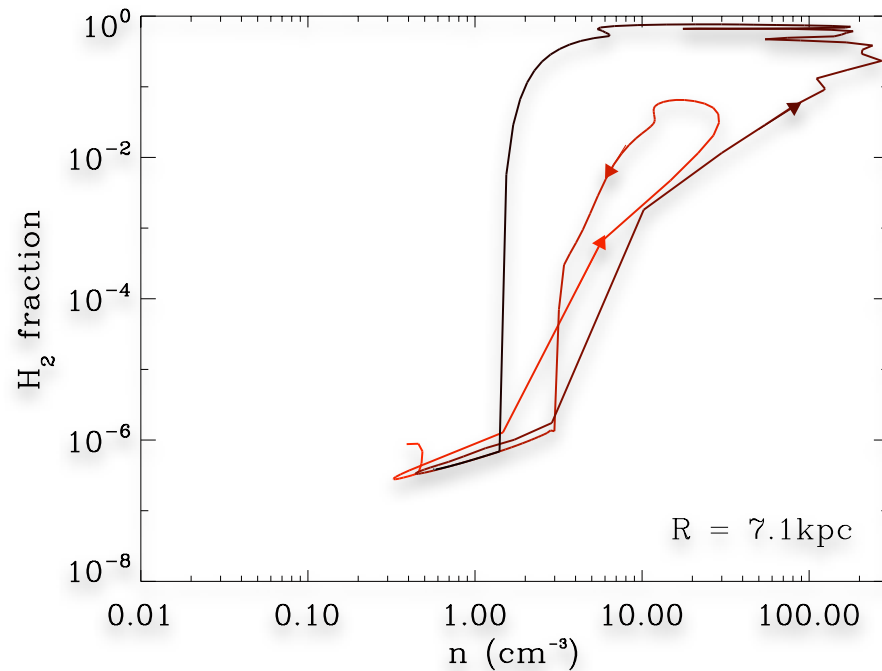


(Dobbs et al. 2008)

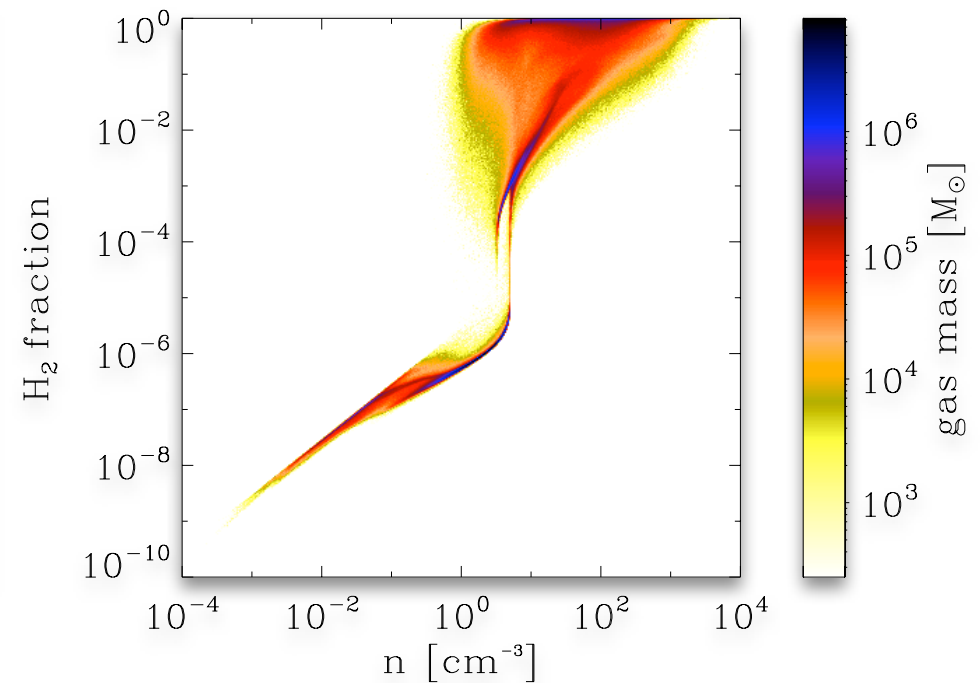


molecular cloud formation

molecular gas fraction of fluid
element as function of time



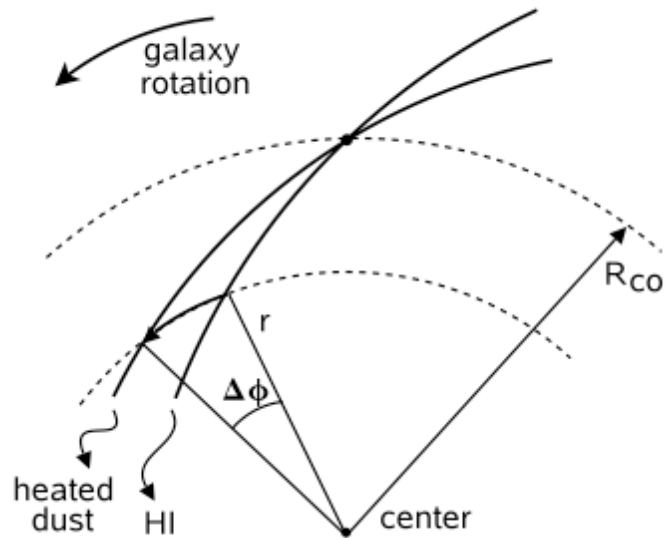
molecular gas fraction as function of density



(Dobbs et al. 2008)



observed timescales



Tamburro et al. (2008)

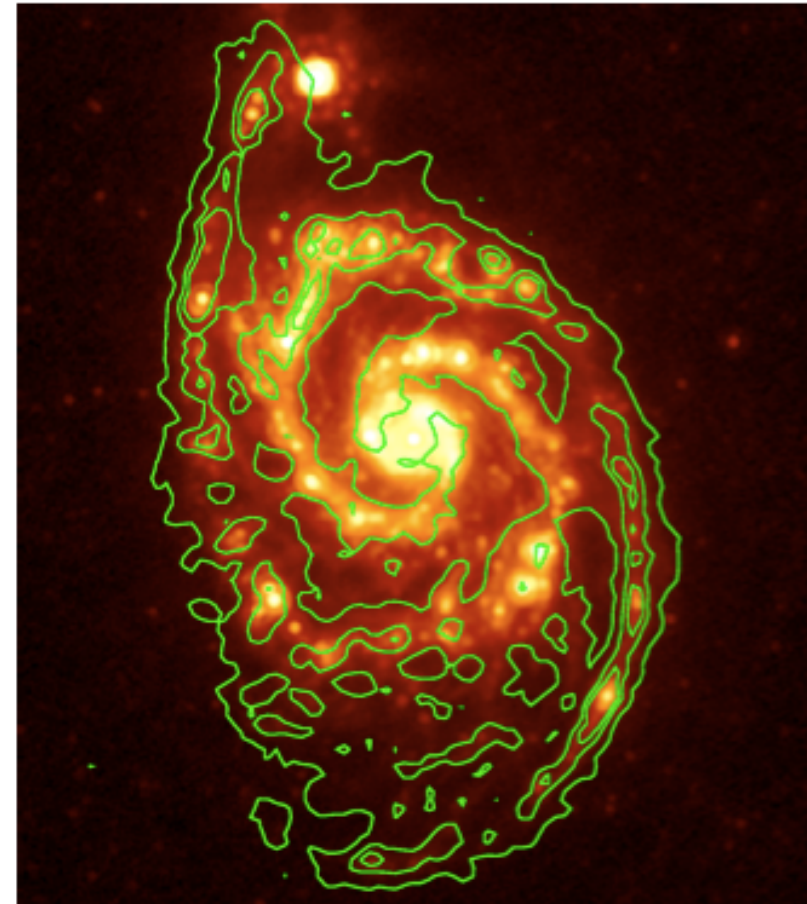


Fig. 1.— NGC 5194: the 24 μm band image is plotted in color scale; the HI emission map is overlaid with green contours.



observed timescales

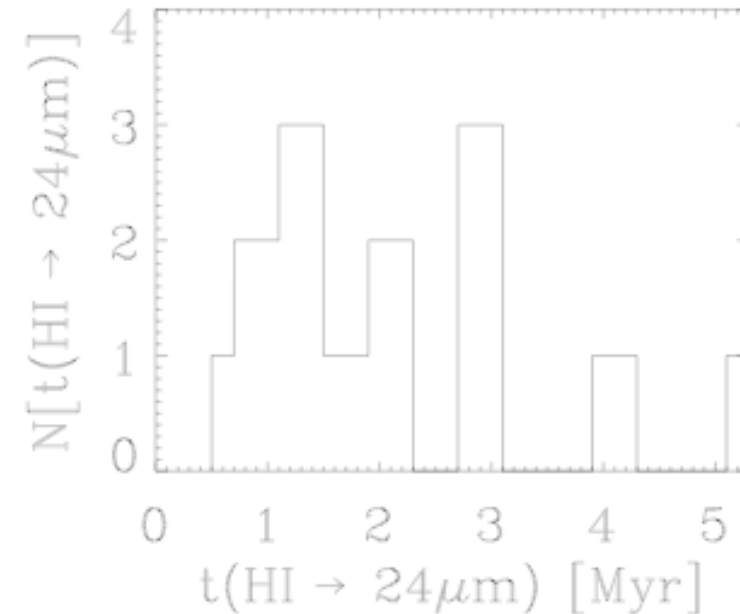
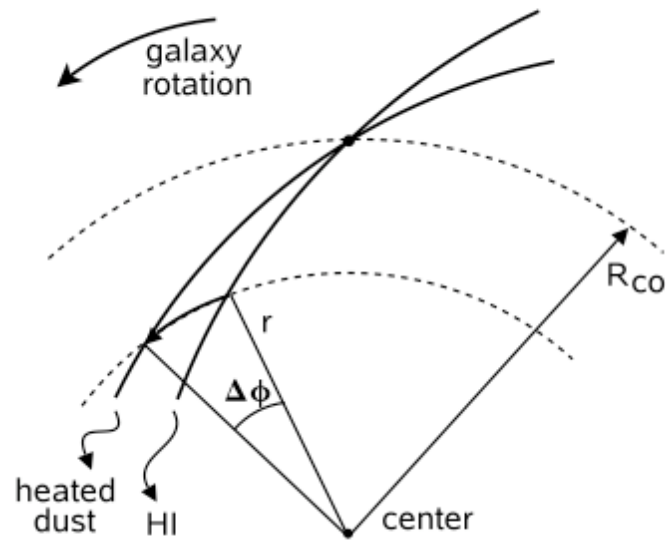
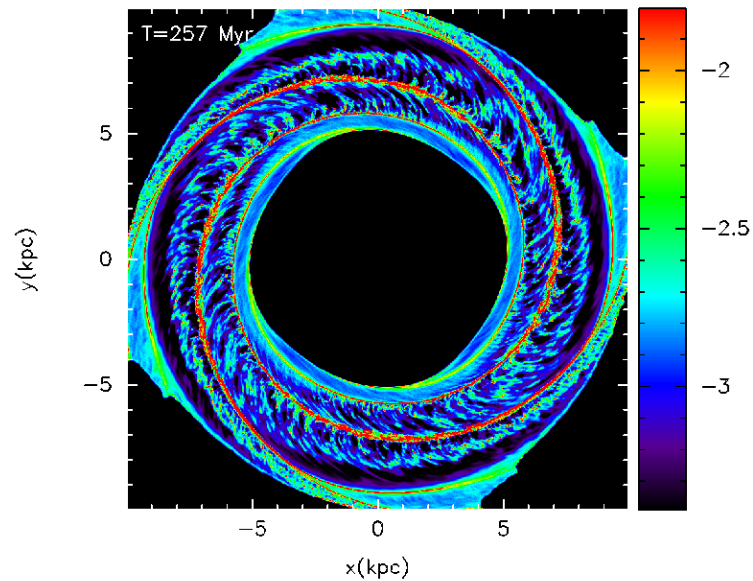


Fig. 5.— Histogram of the time scales $t_{\text{HI} \rightarrow 24 \mu\text{m}}$ derived from the fits in Figure 4 and listed in Table. 2 for the 14 sample galaxies listed in Table. 1. The timescales range between 1 and 4 Myr for almost all galaxies.

Tamburro et al. (2008)



calculated timescales



Dobbs et al. (2008)

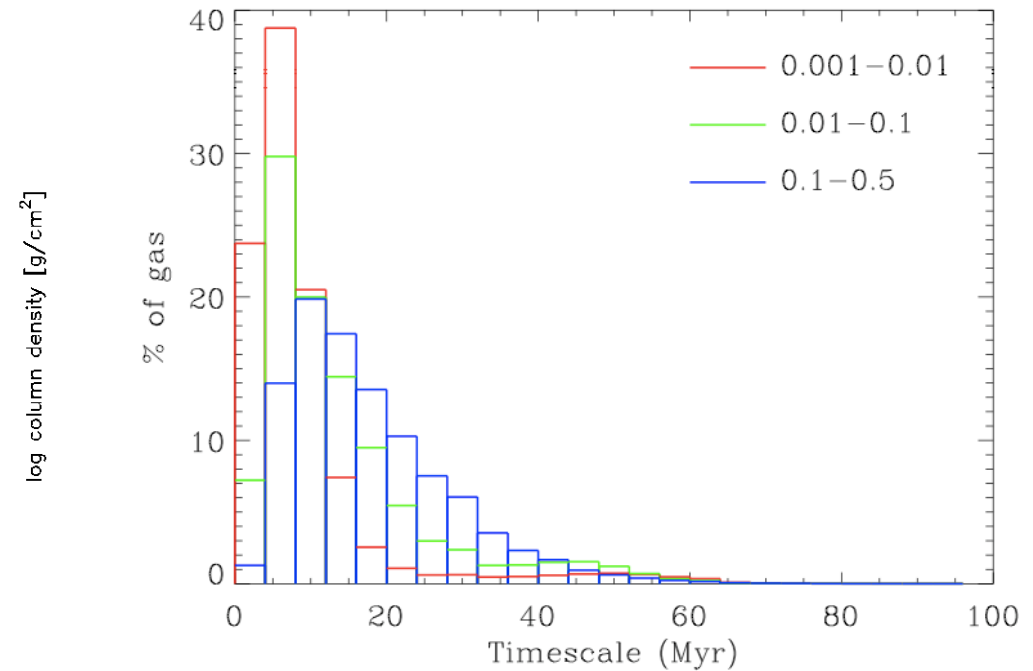


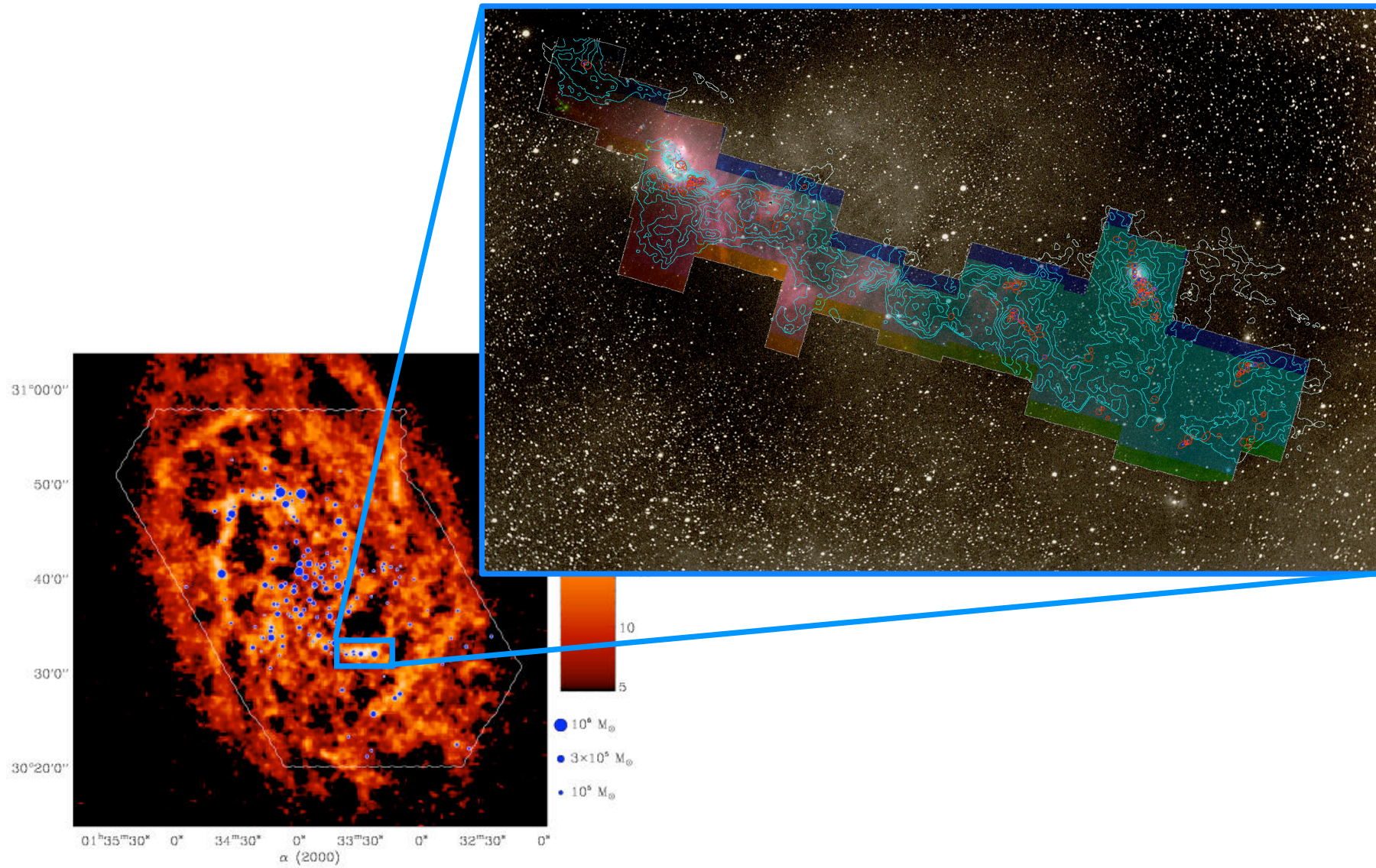
Figure 16. This histogram gives the distribution of timescales over which the gas reaches certain molecular gas fractions. The timescales denote the time for the H_2 fraction of a particle to increase from 0.001 to 0.01, 0.01 to 0.1 and 0.1 to 0.5, as indicated.



molecular cloud formation



zooming in ...



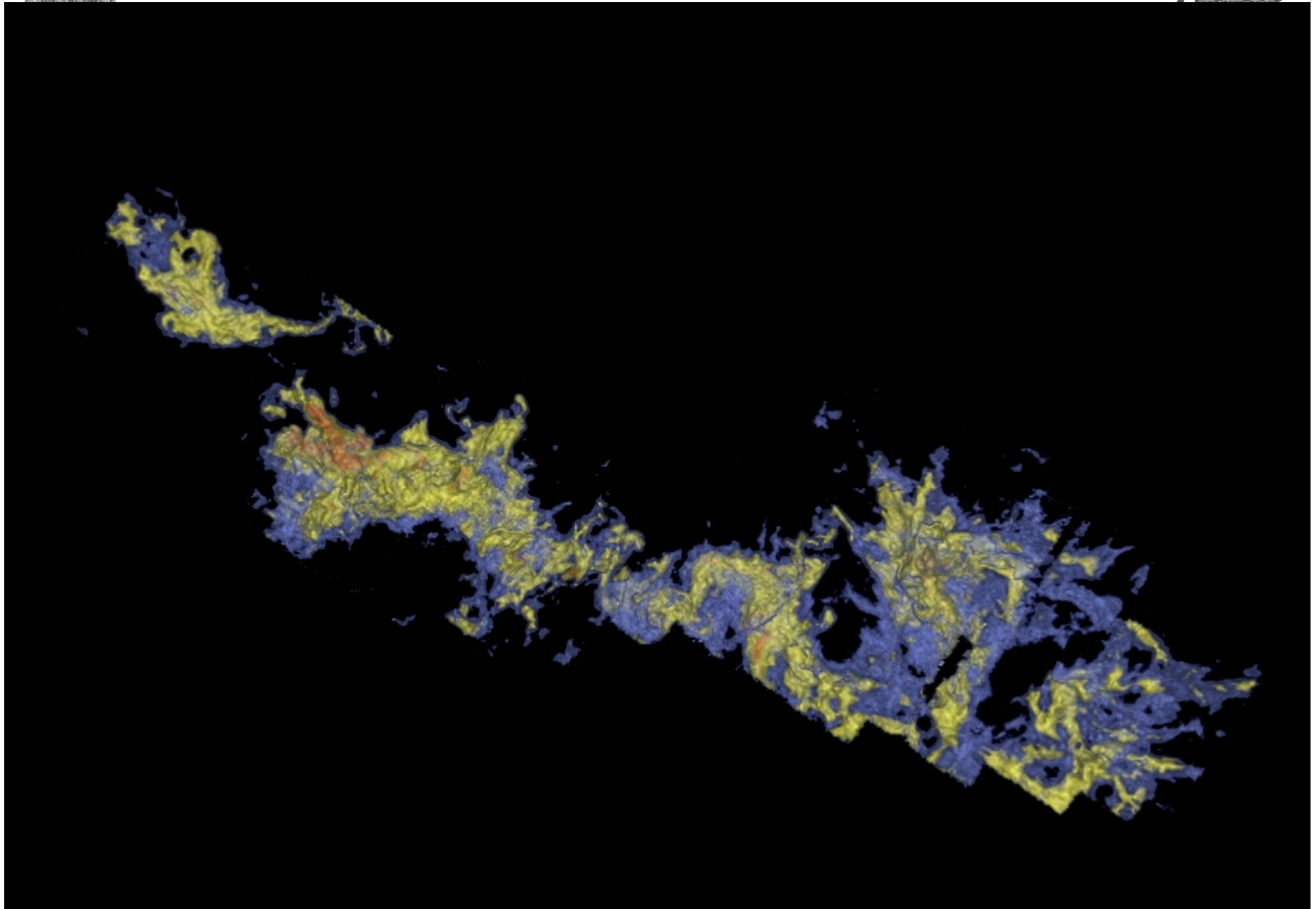
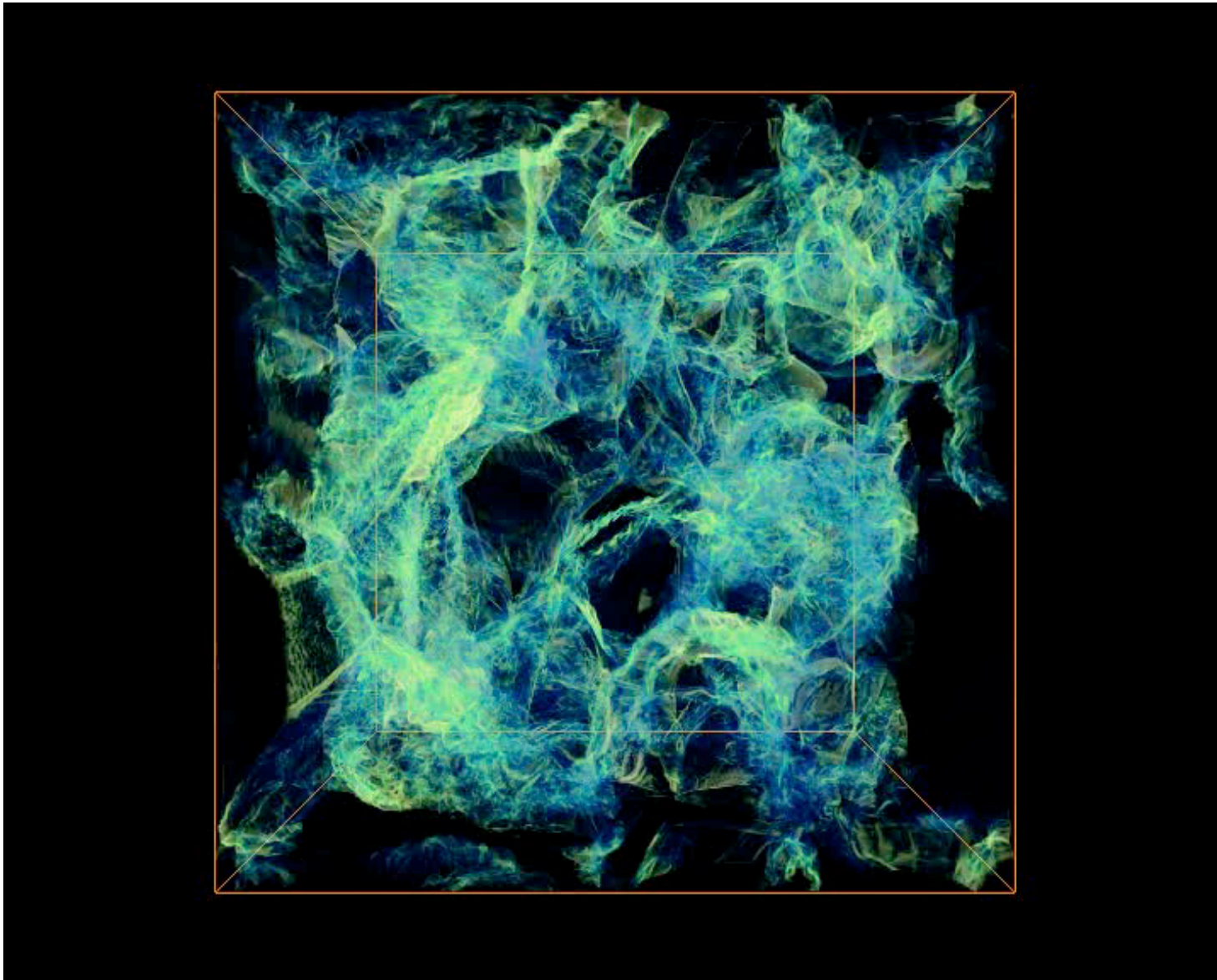


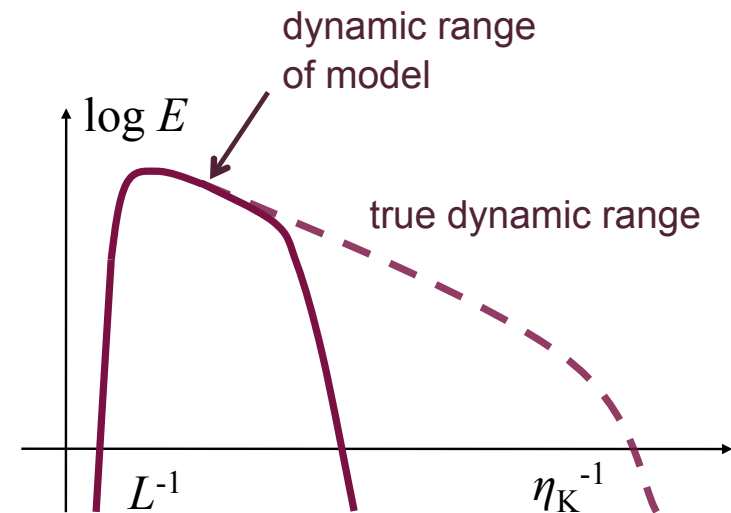
image from Alyssa Goodman: COMPLETE survey



(movie from Christoph Federrath)

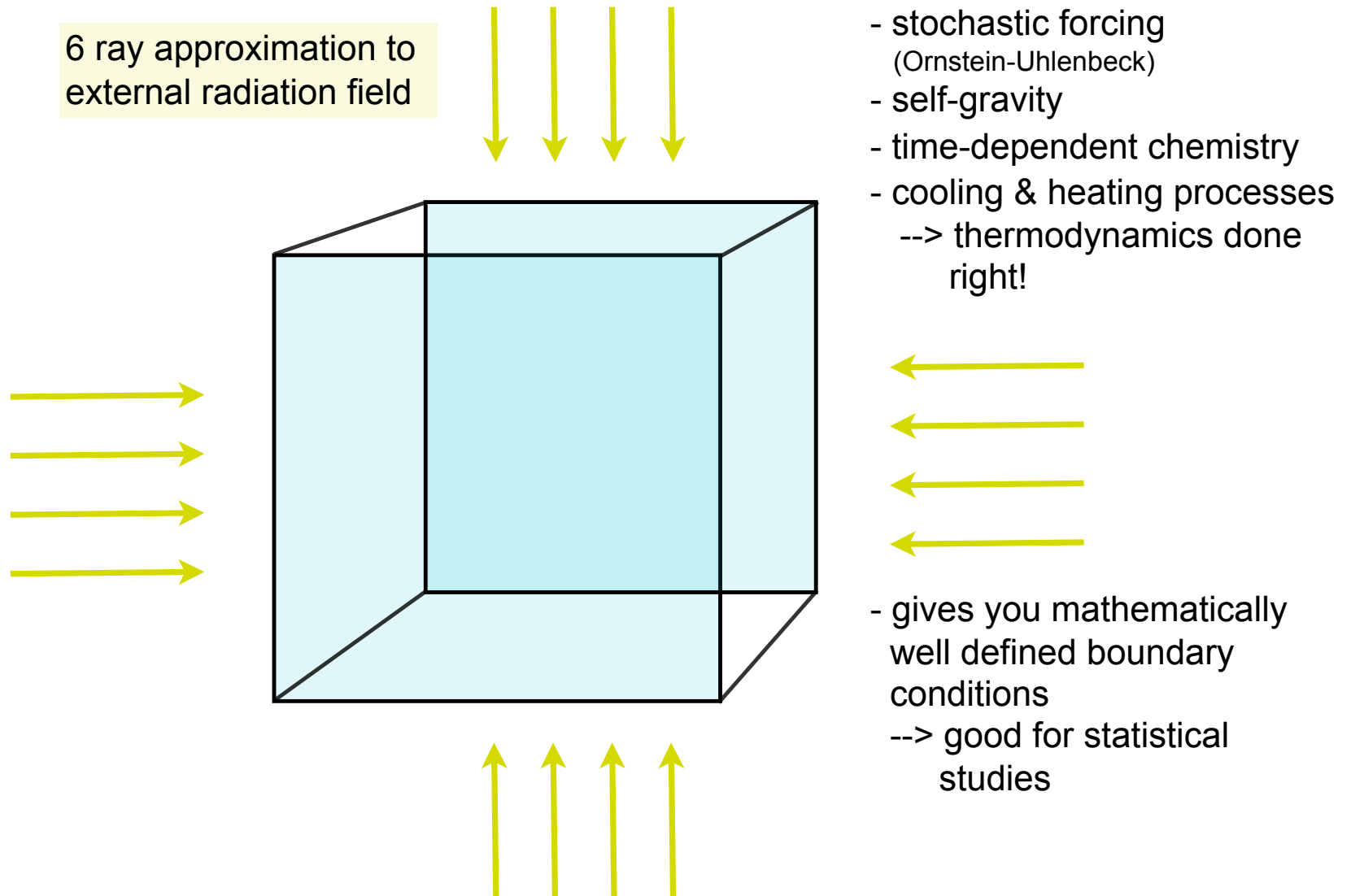
Large-eddy simulations

- We use **LES** to model the large-scale dynamics
- Principal problem: only large scale flow properties
 - Reynolds number: $Re = LV/\nu$ ($Re_{nature} \gg Re_{model}$)
 - dynamic range much smaller than true physical one
 - need **subgrid model** (in our case simple: only dissipation)
 - but what to do for more complex when processes on subgrid scale determine large-scale dynamics (chemical reactions, nuclear burning, etc)
 - Turbulence is “space filling” --> difficulty for AMR (don't know what criterion to use for refinement)
- How **large** a Reynolds number do we need to catch basic dynamics right?





experimental set-up





chemical model 0

- 32 chemical species

- 17 in instantaneous equilibrium:

H^- , H_2^+ , H_3^+ , CH^+ , CH_2^+ , OH^+ , H_2O^+ , H_3O^+ , CO^+ , HOC^+ , O^- , C^- and O_2^+

- 19 full non-equilibrium evolution

e^- , H^+ , H , H_2 , He , He^+ , C , C^+ , O , O^+ , OH , H_2O , CO ,

C_2 , O_2 , HCO^+ , CH , CH_2 and CH_3^+

- 218 reactions

- various heating and cooling processes

(Glover, Federrath, Mac Low, Klessen, 2010)



chemical model 1

Process	Reference(s)
Cooling:	
C fine structure lines	Atomic data – Silva & Viegas (2002) Collisional rates (H) – Abrahamsson, Krems & Dalgarno (2007) Collisional rates (H ₂) – Schroder et al. (1991) Collisional rates (e ⁻) – Johnson et al. (1987) Collisional rates (H ⁺) – Roueff & Le Bourlot (1990)
C ⁺ fine structure lines	Atomic data – Silva & Viegas (2002) Collisional rates (H ₂) – Flower & Launay (1977) Collisional rates (H, T < 2000 K) – Hollenbach & McKee (1989) Collisional rates (H, T > 2000 K) – Keenan et al. (1986) Collisional rates (e ⁻) – Wilson & Bell (2002)
O fine structure lines	Atomic data – Silva & Viegas (2002) Collisional rates (H) – Abrahamsson, Krems & Dalgarno (2007) Collisional rates (H ₂) – see Glover & Jappsen (2007) Collisional rates (e ⁻) – Bell, Berrington & Thomas (1998) Collisional rates (H ⁺) – Pequignot (1990, 1996) Le Bourlot, Pineau des Forêts & Flower (1999)
H ₂ rovibrational lines	Neufeld & Kaufman (1993); Neufeld, Lepp & Melnick (1995)
CO and H ₂ O rovibrational lines	Pavlovski et al. (2002)
OH rotational lines	Hollenbach & McKee (1989)
Gas-grain energy transfer	Wolfire et al. (2003)
Recombination on grains	Sutherland & Dopita (1993)
Atomic resonance lines	Abel et al. (1997)
H collisional ionization	See Table B1
H ₂ collisional dissociation	Cen (1992)
Compton cooling	
Heating:	
Photoelectric effect	Bakes & Tielens (1994); Wolfire et al. (2003)
H ₂ photodissociation	Black & Dalgarno (1977)
UV pumping of H ₂	Burton, Hollenbach & Tielens (1990)
H ₂ formation on dust grains	Hollenbach & McKee (1989)
Cosmic ray ionization	Goldsmith & Langer (1978)

(Glover, Federrath, Mac Low, Klessen, in prep)



(Glover, Federrath, Mac Low, Klessen, in prep)

Table B1. List of collisional gas-phase reactions included in our chemical model

No.	Reaction			
1	$\text{H} + \text{e}^- \rightarrow \text{H}^- + \gamma$	$k_1 = \text{dex}[-17.845 + 0.762 \log T + 0.1523(\log T)^2]$ $= \text{dex}[-16.420 + 0.1998(\log T)^2 - 5.447 \times 10^{-3}(\log T)^4 + 4.0415 \times 10^{-5}(\log T)^6]$	$T \leq 6000 \text{ K}$	1
2	$\text{H}^- + \text{H} \rightarrow \text{H}_2 + \text{e}^-$	$k_2 = 1.5 \times 10^{-9}$ $= 4.0 \times 10^{-9} T^{-0.17}$	$T > 6000 \text{ K}$ $T \leq 300 \text{ K}$ $T > 300 \text{ K}$	2
3	$\text{H} + \text{H}^+ \rightarrow \text{H}_2^+ + \gamma$	$k_3 = \text{dex}[-19.38 - 1.523 \log T + 1.118(\log T)^2 - 0.1269(\log T)^3]$		3
4	$\text{H} + \text{H}_2^+ \rightarrow \text{H}_2 + \text{H}^+$	$k_4 = 6.4 \times 10^{-10}$		4
5	$\text{H}^- + \text{H}^+ \rightarrow \text{H} + \text{H}$	$k_5 = 2.4 \times 10^{-6} T^{-1/2} (1.0 + T/20000)$		5
6	$\text{H}_2^+ + \text{e}^- \rightarrow \text{H} + \text{H}$	$k_6 = 1.0 \times 10^{-8}$ $= 1.32 \times 10^{-6} T^{-0.76}$	$T \leq 617 \text{ K}$ $T > 617 \text{ K}$	6
7	$\text{H}_2 + \text{H}^+ \rightarrow \text{H}_2^+ + \text{H}$	$k_7 = [-3.3232183 \times 10^{-7} + 3.3735382 \times 10^{-7} \ln T - 1.4491368 \times 10^{-7} (\ln T)^2 + 3.4172805 \times 10^{-8} (\ln T)^3 - 4.7813720 \times 10^{-9} (\ln T)^4 + 3.9731542 \times 10^{-10} (\ln T)^5 - 1.8171411 \times 10^{-11} (\ln T)^6 + 3.5311932 \times 10^{-13} (\ln T)^7] \times \exp\left(\frac{-21237.15}{T}\right)$		7
8	$\text{H}_2 + \text{e}^- \rightarrow \text{H} + \text{H} + \text{e}^-$	$k_8 = 3.73 \times 10^{-9} T^{0.1121} \exp\left(\frac{-99430}{T}\right)$		8
9	$\text{H}_2 + \text{H} \rightarrow \text{H} + \text{H} + \text{H}$	$k_{9,l} = 6.67 \times 10^{-12} T^{1/2} \exp\left[-\left(1 + \frac{63590}{T}\right)\right]$ $k_{9,h} = 3.52 \times 10^{-9} \exp\left(-\frac{43900}{T}\right)$ $n_{\text{cr,H}} = \text{dex}\left[3.0 - 0.416 \log\left(\frac{T}{10000}\right) - 0.327 \left\{\log\left(\frac{T}{10000}\right)\right\}^2\right]$		9 10 10
10	$\text{H}_2 + \text{H}_2 \rightarrow \text{H}_2 + \text{H} + \text{H}$	$k_{10,l} = \frac{5.996 \times 10^{-30} T^{4.1881}}{(1.0 + 6.761 \times 10^{-6} T)^{5.6881}} \exp\left(-\frac{54657.4}{T}\right)$ $k_{10,h} = 1.3 \times 10^{-9} \exp\left(-\frac{53300}{T}\right)$ $n_{\text{cr,H}_2} = \text{dex}\left[4.845 - 1.3 \log\left(\frac{T}{10000}\right) + 1.62 \left\{\log\left(\frac{T}{10000}\right)\right\}^2\right]$		11 12 12
11	$\text{H} + \text{e}^- \rightarrow \text{H}^+ + \text{e}^- + \text{e}^-$	$k_{11} = \exp[-3.271396786 \times 10^1 + 1.35365560 \times 10^1 \ln T_e - 5.73932875 \times 10^0 (\ln T_e)^2 + 1.56315498 \times 10^0 (\ln T_e)^3 - 2.87705600 \times 10^{-1} (\ln T_e)^4 + 3.48255977 \times 10^{-2} (\ln T_e)^5 - 2.63197617 \times 10^{-3} (\ln T_e)^6 + 1.11954395 \times 10^{-4} (\ln T_e)^7 - 2.03914985 \times 10^{-6} (\ln T_e)^8]$		13
12	$\text{H}^+ + \text{e}^- \rightarrow \text{H} + \gamma$	$k_{12,A} = 1.269 \times 10^{-13} \left(\frac{315614}{T}\right)^{1.503} \times [1.0 + \left(\frac{604625}{T}\right)^{0.470}]^{-1.923}$ $k_{12,B} = 2.753 \times 10^{-14} \left(\frac{315614}{T}\right)^{1.500} \times [1.0 + \left(\frac{115188}{T}\right)^{0.407}]^{-2.242}$	Case A Case B	14 14
13	$\text{H}^- + \text{e}^- \rightarrow \text{H} + \text{e}^- + \text{e}^-$	$k_{13} = \exp[-1.801849334 \times 10^1 + 2.36085220 \times 10^0 \ln T_e - 2.82744300 \times 10^{-1} (\ln T_e)^2 + 1.62331664 \times 10^{-2} (\ln T_e)^3 - 3.36501203 \times 10^{-2} (\ln T_e)^4 + 1.17832978 \times 10^{-2} (\ln T_e)^5 - 1.65619470 \times 10^{-3} (\ln T_e)^6 + 1.06827520 \times 10^{-4} (\ln T_e)^7 - 2.63128581 \times 10^{-6} (\ln T_e)^8]$		13



Table B1.

No. Rea

1 H+

2 H⁻

3 H+

4 H+

5 H⁻

6 H₂⁺

7 H₂

8 H₂

9 H₂

10 H₂

11 H+

12 H⁺

13 H⁻

14	$H^- + H \rightarrow H + H + e^-$	$k_{14} = 2.5634 \times 10^{-9} T_e^{1.78186}$ $= \exp[-2.0372609 \times 10^1$ $+ 1.13944933 \times 10^0 \ln T_e$ $- 1.4210135 \times 10^{-1} (\ln T_e)^2$ $- 8.4944554 \times 10^{-3} (\ln T_e)^3$ $- 1.323641 \times 10^{-3} (\ln T_e)^4$ $+ 2.11250 \times 10^{-4} (\ln T_e)^5$ $+ 8.6639632 \times 10^{-5} (\ln T_e)^6$ $- 2.5850097 \times 10^{-5} (\ln T_e)^7$ $+ 2.4555012 \times 10^{-6} (\ln T_e)^8$ $- 8.0683825 \times 10^{-8} (\ln T_e)^9]$	$T_e \leq 0.1 \text{ eV}$	13
15	$H^- + H^+ \rightarrow H_2^+ + e^-$	$k_{15} = 6.9 \times 10^{-9} T^{-0.35}$ $= 9.6 \times 10^{-7} T^{-0.90}$	$T_e > 0.1 \text{ eV}$ $T \leq 8000 \text{ K}$ $T > 8000 \text{ K}$	15
16	$He + e^- \rightarrow He^+ + e^- + e^-$	$k_{16} = \exp[-4.409864886 \times 10^1$ $+ 2.391596563 \times 10^1 \ln T_e$ $- 1.07532302 \times 10^1 (\ln T_e)^2$ $+ 3.05803875 \times 10^0 (\ln T_e)^3$ $- 5.6851189 \times 10^{-1} (\ln T_e)^4$ $+ 6.79539123 \times 10^{-2} (\ln T_e)^5$ $- 5.0090561 \times 10^{-3} (\ln T_e)^6$ $+ 2.06723616 \times 10^{-4} (\ln T_e)^7$ $- 3.64916141 \times 10^{-6} (\ln T_e)^8]$		13
17	$He^+ + e^- \rightarrow He + \gamma$	$k_{17,rr,A} = 10^{-11} T^{-0.5} [12.72 - 1.615 \log T$ $- 0.3162 (\log T)^2 + 0.0493 (\log T)^3]$ $k_{17,rr,B} = 10^{-11} T^{-0.5} [11.19 - 1.676 \log T$ $- 0.2852 (\log T)^2 + 0.04433 (\log T)^3]$ $k_{17,di} = 1.9 \times 10^{-3} T^{-1.5} \exp\left(-\frac{473421}{T}\right)$ $\times [1.0 + 0.3 \exp\left(-\frac{94684}{T}\right)]$	Case A Case B	16
18	$He^+ + H \rightarrow He + H^+$	$k_{18} = 1.25 \times 10^{-15} \left(\frac{T}{300}\right)^{0.25}$		18
19	$He + H^+ \rightarrow He^+ + H$	$k_{19} = 1.26 \times 10^{-9} T^{-0.75} \exp\left(-\frac{127500}{T}\right)$ $= 4.0 \times 10^{-37} T^{4.74}$	$T \leq 10000 \text{ K}$ $T > 10000 \text{ K}$	19
20	$C^+ + e^- \rightarrow C + \gamma$	$k_{20} = 4.67 \times 10^{-12} \left(\frac{T}{300}\right)^{-0.6}$ $= 1.23 \times 10^{-17} \left(\frac{T}{300}\right)^{2.49} \exp\left(\frac{21845.6}{T}\right)$ $= 9.62 \times 10^{-8} \left(\frac{T}{300}\right)^{-1.37} \exp\left(\frac{-115786.2}{T}\right)$	$T \leq 7950 \text{ K}$ $7950 \text{ K} < T \leq 21140 \text{ K}$ $T > 21140 \text{ K}$	20
21	$O^+ + e^- \rightarrow O + \gamma$	$k_{21} = 1.30 \times 10^{-10} T^{-0.64}$ $= 1.41 \times 10^{-10} T^{-0.66} + 7.4 \times 10^{-4} T^{-1.5}$ $\times \exp\left(-\frac{175000}{T}\right) [1.0 + 0.062 \times \exp\left(-\frac{145000}{T}\right)]$	$T \leq 400 \text{ K}$ $T > 400 \text{ K}$	21
22	$C + e^- \rightarrow C^+ + e^- + e^-$	$k_{22} = 6.85 \times 10^{-8} (0.193 + u)^{-1} u^{0.25} e^{-u}$	$u = 11.26/T_e$	22
23	$O + e^- \rightarrow O^+ + e^- + e^-$	$k_{23} = 3.59 \times 10^{-8} (0.073 + u)^{-1} u^{0.34} e^{-u}$	$u = 13.6/T_e$	22
24	$O^+ + H \rightarrow O + H^+$	$k_{24} = 4.99 \times 10^{-11} T^{0.405} + 7.54 \times 10^{-10} T^{-0.458}$		23
25	$O + H^+ \rightarrow O^+ + H$	$k_{25} = [1.08 \times 10^{-11} T^{0.517}$ $+ 4.00 \times 10^{-10} T^{0.00669}] \exp\left(-\frac{227}{T}\right)$		24
26	$O + He^+ \rightarrow O^+ + He$	$k_{26} = 4.991 \times 10^{-15} \left(\frac{T}{10000}\right)^{0.3794} \exp\left(-\frac{T}{1121000}\right)$ $+ 2.780 \times 10^{-15} \left(\frac{T}{10000}\right)^{-0.2163} \exp\left(\frac{T}{815800}\right)$		25
27	$C + H^+ \rightarrow C^+ + H$	$k_{27} = 3.9 \times 10^{-16} T^{0.213}$		24
28	$C^+ + H \rightarrow C + H^+$	$k_{28} = 6.08 \times 10^{-14} \left(\frac{T}{10000}\right)^{1.96} \exp\left(-\frac{170000}{T}\right)$		24
29	$C + He^+ \rightarrow C^+ + He$	$k_{29} = 8.58 \times 10^{-17} T^{0.751}$ $= 3.25 \times 10^{-17} T^{0.968}$ $= 2.77 \times 10^{-19} T^{1.597}$	$T \leq 200 \text{ K}$ $200 < T \leq 2000 \text{ K}$ $T > 2000 \text{ K}$	26
30	$H_2 + He \rightarrow H + H + He$	$k_{30,l} = \text{dex}[-27.029 + 3.801 \log(T) - 29487/T]$ $k_{30,h} = \text{dex}[-2.729 - 1.75 \log(T) - 23474/T]$ $n_{cr,He} = \text{dex}[5.0792(1.0 - 1.23 \times 10^{-5}(T - 2000))]$		27
31	$OH + H \rightarrow O + H + H$	$k_{31} = 6.0 \times 10^{-9} \exp\left(-\frac{50900}{T}\right)$		28
32	$HOC^+ + H_2 \rightarrow HCO^+ + H_2$	$k_{32} = 3.8 \times 10^{-10}$		29
33	$HOC^+ + CO \rightarrow HCO^+ + CO$	$k_{33} = 4.0 \times 10^{-10}$		30
34	$C + H_2 \rightarrow CH + H$	$k_{34} = 6.64 \times 10^{-10} \exp\left(-\frac{11700}{T}\right)$		31
35	$CH + H \rightarrow C + H_2$	$k_{35} = 1.31 \times 10^{-10} \exp\left(-\frac{80}{T}\right)$		32



(Glover, Federrath, Mac Low, Klessen, in prep)

Table B1.1

No. Rea

1 H+

2 H⁻

3 H⁺

4 H⁺

5 H⁻

6 H₂⁺

7 H₂

8 H₂

9 H₂

10 H₂

11 H⁺

12 H⁺

13 H⁻

14	H ⁻ + H → H + H + e ⁻	$k_{14} = 2.5634 \times 10^{-9} T_e^{1.78186}$	$T_e \leq 0.1 \text{ eV}$	13
36	CH + H ₂ → CH ₂ + H	$k_{36} = 5.46 \times 10^{-10} \exp\left(-\frac{1843}{T}\right)$		33
37	CH + C → C ₂ + H	$k_{37} = 6.59 \times 10^{-11}$		34
38	CH + C → CO + H	$k_{38} = 6.6 \times 10^{-11}$	$T \leq 1000 \text{ K}$	35
39	C + H ₂ → CH + H	$k_{39} = 1.09 \times 10^{-10} \exp\left(-\frac{2111}{T}\right)$	$T > 1000 \text{ K}$	36
40	CH ₂ + O → CO + H + H	$k_{40} = 1.33 \times 10^{-10}$		38
41	CH ₂ + O → CO + H ₂	$k_{41} = 8.0 \times 10^{-11}$		39
42	C ₂ + O → CO + C	$k_{42} = 5.0 \times 10^{-11} \left(\frac{T}{300}\right)^{0.5}$ $= 5.0 \times 10^{-11} \left(\frac{T}{300}\right)^{0.757}$	$T \leq 300 \text{ K}$ $T > 300 \text{ K}$	40 41
43	O + H ₂ → OH + H	$k_{43} = 3.14 \times 10^{-13} \left(\frac{T}{300}\right)^{2.7} \exp\left(-\frac{3150}{T}\right)$		42
44	OH + H → O + H ₂	$k_{44} = 6.99 \times 10^{-14} \left(\frac{T}{300}\right)^{2.8} \exp\left(-\frac{1950}{T}\right)$		43
45	OH + H ₂ → H ₂ O + H	$k_{45} = 2.05 \times 10^{-12} \left(\frac{T}{300}\right)^{1.52} \exp\left(-\frac{1736}{T}\right)$		44
46	OH + C → CO + H	$k_{46} = 1.0 \times 10^{-10}$		34
47	OH + O → O ₂ + H	$k_{47} = 3.50 \times 10^{-11}$ $= 1.77 \times 10^{-11} \exp\left(\frac{178}{T}\right)$	$T \leq 261 \text{ K}$ $T > 261 \text{ K}$	45 33
48	OH + OH → H ₂ O + H	$k_{48} = 1.65 \times 10^{-12} \left(\frac{T}{300}\right)^{1.14} \exp\left(-\frac{59}{T}\right)$		34
49	H ₂ O + H → H ₂ + OH	$k_{49} = 1.59 \times 10^{-11} \left(\frac{T}{300}\right)^{1.2} \exp\left(-\frac{9610}{T}\right)$		46
50	O ₂ + H → OH + O	$k_{50} = 2.61 \times 10^{-10} \exp\left(-\frac{8156}{T}\right)$		33
51	O ₂ + H ₂ → OH + OH	$k_{51} = 3.16 \times 10^{-10} \exp\left(-\frac{21890}{T}\right)$		47
52	O ₂ + C → CO + O	$k_{52} = 4.7 \times 10^{-11} \left(\frac{T}{300}\right)^{-0.34}$ $= 2.48 \times 10^{-12} \left(\frac{T}{300}\right)^{1.54} \exp\left(\frac{613}{T}\right)$	$T \leq 295 \text{ K}$ $T > 295 \text{ K}$	34 33
53	CO + H → C + OH	$k_{53} = 1.1 \times 10^{-10} \left(\frac{T}{300}\right)^{0.5} \exp\left(-\frac{72700}{T}\right)$		28
54	H ₂ ⁺ + H ₂ → H ₃ ⁺ + H	$k_{54} = 2.24 \times 10^{-9} \left(\frac{T}{300}\right)^{0.042} \exp\left(-\frac{T}{46600}\right)$		48
55	H ₃ ⁺ + H → H ₂ ⁺ + H ₂	$k_{55} = 7.7 \times 10^{-9} \exp\left(-\frac{17560}{T}\right)$		49
56	C + H ₂ ⁺ → CH ⁺ + H	$k_{56} = 2.4 \times 10^{-9}$		28
57	C + H ₃ ⁺ → CH ⁺ + H ₂	$k_{57} = 2.0 \times 10^{-9}$		28
58	C ⁺ + H ₂ → CH ⁺ + H	$k_{58} = 1.0 \times 10^{-10} \exp\left(-\frac{4640}{T}\right)$		50
59	CH ⁺ + H → C ⁺ + H ₂	$k_{59} = 7.5 \times 10^{-10}$		51
60	CH ⁺ + H ₂ → CH ₂ ⁺ + H	$k_{60} = 1.2 \times 10^{-9}$		51
61	CH ⁺ + O → CO ⁺ + H	$k_{61} = 3.5 \times 10^{-10}$		52
62	CH ₂ + H ⁺ → CH ₃ ⁺ + H	$k_{62} = 1.4 \times 10^{-9}$		28
63	CH ₂ ⁺ + H → CH ₃ ⁺ + H	$k_{63} = 1.0 \times 10^{-9} \exp\left(-\frac{7080}{T}\right)$		28
64	CH ₂ ⁺ + H ₂ → CH ₃ ⁺ + H	$k_{64} = 1.6 \times 10^{-9}$		53
65	CH ₂ ⁺ + O → HCO ⁺ + H	$k_{65} = 7.5 \times 10^{-10}$		28
66	CH ₃ ⁺ + H → CH ₂ ⁺ + H ₂	$k_{66} = 7.0 \times 10^{-10} \exp\left(-\frac{10560}{T}\right)$		28
67	CH ₃ ⁺ + O → HCO ⁺ + H ₂	$k_{67} = 4.0 \times 10^{-10}$		54
68	C ₂ + O ⁺ → CO ⁺ + C	$k_{68} = 4.8 \times 10^{-10}$		28
69	O ⁺ + H ₂ → OH ⁺ + H	$k_{69} = 1.7 \times 10^{-9}$		55
70	O + H ₂ ⁺ → OH ⁺ + H	$k_{70} = 1.5 \times 10^{-9}$		28
71	O + H ₃ ⁺ → OH ⁺ + H ₂	$k_{71} = 8.4 \times 10^{-10}$		56
72	OH + H ₃ ⁺ → H ₂ O ⁺ + H ₂	$k_{72} = 1.3 \times 10^{-9}$		28
73	OH + C ⁺ → CO ⁺ + H	$k_{73} = 7.7 \times 10^{-10}$		28
74	OH ⁺ + H ₂ → H ₂ O ⁺ + H	$k_{74} = 1.01 \times 10^{-9}$		57
75	H ₂ O ⁺ + H ₂ → H ₃ O ⁺ + H	$k_{75} = 6.4 \times 10^{-10}$		58
76	H ₂ O + H ₃ ⁺ → H ₃ O ⁺ + H ₂	$k_{76} = 5.9 \times 10^{-9}$		59
77	H ₂ O + C ⁺ → HCO ⁺ + H	$k_{77} = 9.0 \times 10^{-10}$		60
78	H ₂ O + C ⁺ → HOC ⁺ + H	$k_{78} = 1.8 \times 10^{-9}$		60
79	H ₃ O ⁺ + C → HCO ⁺ + H ₂	$k_{79} = 1.0 \times 10^{-11}$		28
80	O ₂ + C ⁺ → CO ⁺ + O	$k_{80} = 3.8 \times 10^{-10}$		53
81	O ₂ + C ⁺ → CO + O ⁺	$k_{81} = 6.2 \times 10^{-10}$		53
82	O ₂ + CH ₂ ⁺ → HCO ⁺ + OH	$k_{82} = 9.1 \times 10^{-10}$		53
83	O ₂ ⁺ + C → CO ⁺ + O	$k_{83} = 5.2 \times 10^{-11}$		28
84	CO + H ₃ ⁺ → HOC ⁺ + H ₂	$k_{84} = 2.7 \times 10^{-11}$		61
85	CO + H ₃ ⁺ → HCO ⁺ + H ₂	$k_{85} = 1.7 \times 10^{-9}$		61
86	HCO ⁺ + C → CO + CH ⁺	$k_{86} = 1.1 \times 10^{-9}$		28
87	HCO ⁺ + H ₂ O → CO + H ₃ O ⁺	$k_{87} = 2.5 \times 10^{-9}$		62





Table B1.1

No.	Reaction	Rate Coefficient
1	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{88} = 7.2 \times 10^{-15}$
2	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{89} = 3.7 \times 10^{-14} \exp\left(\frac{35}{T}\right)$
3	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{90} = 1.9 \times 10^{-9}$
4	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{91} = 1.4 \times 10^{-9}$
5	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{92} = 7.7 \times 10^{-10}$
6	$\text{H}_2^+ + \text{H} \rightarrow \text{H} + \text{H}_2^+$	$k_{93} = 1.1 \times 10^{-9}$
7	$\text{H}_2 + \text{H} \rightarrow \text{H} + \text{H}_2$	$k_{94} = 2.1 \times 10^{-10}$
8	$\text{H}_2 + \text{H} \rightarrow \text{H} + \text{H}_2$	$k_{95} = 1.1 \times 10^{-9}$
9	$\text{H}_2 + \text{H} \rightarrow \text{H} + \text{H}_2$	$k_{96} = 6.9 \times 10^{-9}$
10	$\text{H}_2 + \text{H} \rightarrow \text{H} + \text{H}_2$	$k_{97} = 2.04 \times 10^{-10}$
11	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{98} = 2.86 \times 10^{-10}$
12	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{99} = 6.05 \times 10^{-11}$
13	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{100} = 2.0 \times 10^{-9}$
14	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{101} = 3.3 \times 10^{-11}$
15	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{102} = 1.1 \times 10^{-9}$
16	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{103} = 5.2 \times 10^{-11}$
17	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{104} = 1.4 \times 10^{-9} \left(\frac{T}{300}\right)^{-0.5}$
18	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{105} = 1.4 \times 10^{-16} \left(\frac{T}{300}\right)^{-0.5}$
19	$\text{H} + \text{H} \rightarrow \text{H} + \text{H} + e^-$	$k_{106} = 7.5 \times 10^{-10}$
20	$\text{C}^+ + \text{H} \rightarrow \text{C} + \text{H}^+$	$k_{107} = 2.3 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.5}$
21	$\text{O}^+ + \text{H} \rightarrow \text{O} + \text{H}^+$	$k_{108} = 2.3 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.5}$
22	$\text{C}^+ + \text{H} \rightarrow \text{C} + \text{H}^+$	$k_{109} = 2.32 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.52} \exp\left(\frac{T}{22400}\right)$
23	$\text{O}^+ + \text{H} \rightarrow \text{O} + \text{H}^+$	$k_{110} = 2.34 \times 10^{-8} \left(\frac{T}{300}\right)^{-0.52}$
24	$\text{O}^+ + \text{H} \rightarrow \text{O} + \text{H}^+$	$k_{111} = 4.36 \times 10^{-8} \left(\frac{T}{300}\right)^{-0.52}$
25	$\text{O}^+ + \text{H} \rightarrow \text{O} + \text{H}^+$	$k_{112} = 7.0 \times 10^{-8} \left(\frac{T}{300}\right)^{-0.5}$
26	$\text{O}^+ + \text{H} \rightarrow \text{O} + \text{H}^+$	$k_{113} = 1.6 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.6}$
27	$\text{C}^+ + \text{H} \rightarrow \text{C} + \text{H}^+$	$k_{114} = 4.03 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.6}$
28	$\text{C}^+ + \text{H} \rightarrow \text{C} + \text{H}^+$	$k_{115} = 7.68 \times 10^{-8} \left(\frac{T}{300}\right)^{-0.6}$
29	$\text{C}^+ + \text{H} \rightarrow \text{C} + \text{H}^+$	$k_{116} = 7.75 \times 10^{-8} \left(\frac{T}{300}\right)^{-0.5}$
30	$\text{H}_2 + \text{H} \rightarrow \text{H} + \text{H}_2$	$k_{117} = 1.95 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.5}$
31	$\text{OH} + \text{H} \rightarrow \text{H} + \text{OH}$	$k_{118} = 2.0 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.4}$
32	$\text{HO} + \text{H} \rightarrow \text{H} + \text{OH}$	$k_{119} = 6.3 \times 10^{-9} \left(\frac{T}{300}\right)^{-0.48}$
33	$\text{HO} + \text{H} \rightarrow \text{H} + \text{OH}$	$k_{120} = 3.05 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.5}$
34	$\text{C}^+ + \text{H} \rightarrow \text{C} + \text{H}^+$	$k_{121} = 3.9 \times 10^{-8} \left(\frac{T}{300}\right)^{-0.5}$
35	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{122} = 8.6 \times 10^{-8} \left(\frac{T}{300}\right)^{-0.5}$
36	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{123} = 1.08 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.5}$
37	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{124} = 6.02 \times 10^{-8} \left(\frac{T}{300}\right)^{-0.5}$
38	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{125} = 2.58 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.5}$
39	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{126} = 5.6 \times 10^{-9} \left(\frac{T}{300}\right)^{-0.5}$
40	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{127} = 1.95 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.7}$
41	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{128} = 2.75 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.55}$
42	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{129} = 2.76 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.64}$
43	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{130} = 2.4 \times 10^{-8} \left(\frac{T}{300}\right)^{-0.64}$
44	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{131} = 1.1 \times 10^{-7} \left(\frac{T}{300}\right)^{-1.0}$
45	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{132} = 1.0 \times 10^{-9}$
46	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{133} = 1.0 \times 10^{-9}$
47	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{134} = 1.0 \times 10^{-10}$
48	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{135} = 5.0 \times 10^{-10}$
49	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{136} = 1.0 \times 10^{-13}$
50	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{137} = 5.0 \times 10^{-10}$
51	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{138} = 5.0 \times 10^{-10}$
52	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{139} = 7.0 \times 10^{-10}$
53	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{140} = 5.0 \times 10^{-10}$
54	$\text{CH} + \text{H} \rightarrow \text{C} + \text{H}_2$	$k_{87} = 2.5 \times 10^{-10}$





Table B1.

No. Rea

1 H+

2 H-

3 H+

4 H+

5 H-

6 H₂⁺

7 H₂

8 H₂

9 H₂

10 H₂

11 H+

12 H⁺

13 H-

14 H⁻ + H → H + H + e⁻

36 CH + H₂

37 CH + C

38 CH + C

39 C + H₂

40 CH₂ + O

41 CH₂ + O

42 C₂ + O →

15 H-

16 He

17 He

18 He

19 He

20 C⁺

21 O⁺

22 C+

23 O+

24 O⁺

25 O⁺

26 O⁺

27 C+

28 C⁺

29 C+

30 H₂

31 OH

32 HO

33 HO

34 C+

35 CH

88 H₂ + He⁺ → He + H₂⁺

89 H₂ + He⁺ → He + H + H⁺

90 CH + H⁺ → CH⁺ + H

91 CH₂ + H⁺ → CH₂⁺ + H

92 CH₂ + H⁺ → CH₂⁺ + H

93 C₂ + H⁺ → C₂⁺ + H

94 OH + H⁺ → OH⁺ + H

95 OH + He⁺ → O⁺ + He + H

96 H₂O + H⁺ → H₂O⁺ + H

97 H₂O + He⁺ → OH + He + H⁺

98 H₂O + He⁺ → OH + He + H⁺

99 C + e⁻ → C⁻ + γ

143 C + H → CH + γ

144 C + H₂ → CH₂ + γ

145 C + C → C₂ + γ

146 C + O → CO + γ

147 C⁺ + H → CH⁺ + γ

148 C⁺ + H₂ → CH₂⁺ + γ

149 C⁺ + O → CO⁺ + γ

150 O + e⁻ → O⁻ + γ

151 O + H → OH + γ

152 O + O → O₂ + γ

153 OH + H → H₂O + γ

154 H + H + H → H₂ + H

155 H + H + H₂ → H₂ + H₂

156 H + H + He → H₂ + He

157 C + C + M → C₂ + M

158 C + O + M → CO + M

159 C⁺ + O + M → CO⁺ + M

160 C + O⁺ + M → CO⁺ + M

161 O + H + M → OH + M

162 OH + H + M → H₂O + M

163 O + O + M → O₂ + M

164 O + CH → HCO⁺ + e⁻

165 H + H(s) → H₂

129 HCO⁺ + e⁻ → CO + H

130 HCO⁺ + e⁻ → OH + C

131 HOC⁺ + e⁻ → CO + H

132 H⁻ + C → CH + e⁻

133 H⁻ + O → OH + e⁻

134 H⁻ + OH → H₂O + e⁻

135 C⁻ + H → CH + e⁻

136 C⁻ + H₂ → CH₂ + e⁻

137 C⁻ + O → CO + e⁻

138 O⁻ + H → OH + e⁻

139 O⁻ + H₂ → H₂O + e⁻

140 O⁻ + C → CO + e⁻

87 HCO⁺ + H₂O → CO + H₃O⁺

k₈₈ = 7.2 × 10⁻¹⁵

k₈₉ = 3.7 × 10⁻¹⁴ exp (35/T)

k₉₀ = 1.9 × 10⁻⁹

k₉₁ = 1.4 × 10⁻⁹

k₉₂ = 7.7 × 10⁻¹⁰

k₉₃ = 1.1 × 10⁻⁹

k₉₄ = 2.1 × 10⁻⁹

k₉₅ = 1.1 × 10⁻⁹

k₉₆ = 6.9 × 10⁻⁹

k₉₇ = 2.04 × 10⁻¹⁰

k₁₄₂ = 2.25 × 10⁻¹⁵

k₁₄₃ = 1.0 × 10⁻¹⁷

k₁₄₄ = 1.0 × 10⁻¹⁷

k₁₄₅ = 4.36 × 10⁻¹⁸ (T/300)^{0.35} exp (-161.3/T)

k₁₄₆ = 2.1 × 10⁻¹⁹

= 3.09 × 10⁻¹⁷ (T/300)^{0.33} exp (-1629/T)

k₁₄₇ = 4.46 × 10⁻¹⁶ T^{-0.5} exp (-4.93/T^{2/3})

k₁₄₈ = 4.0 × 10⁻¹⁶ (T/300)^{-0.2}

k₁₄₉ = 2.5 × 10⁻¹⁸

= 3.14 × 10⁻¹⁸ (T/300)^{-0.15} exp (68/T)

k₁₅₀ = 1.5 × 10⁻¹⁵

k₁₅₁ = 9.9 × 10⁻¹⁹ (T/300)^{-0.38}

k₁₅₂ = 4.9 × 10⁻²⁰ (T/300)^{1.58}

k₁₅₃ = 5.26 × 10⁻¹⁸ (T/300)^{-5.22} exp (-90/T)

k₁₅₄ = 1.32 × 10⁻³² (T/300)^{-0.38}

= 1.32 × 10⁻³² (T/300)^{-1.0}

k₁₅₅ = 2.8 × 10⁻³¹ T^{-0.6}

k₁₅₆ = 6.9 × 10⁻³² T^{-0.4}

k₁₅₇ = 5.99 × 10⁻³³ (T/5000)^{-1.6}

= 5.99 × 10⁻³³ (T/5000)^{-0.64} exp (5255/T)

k₁₅₈ = 6.16 × 10⁻²⁹ (T/300)^{-3.08}

= 2.14 × 10⁻²⁹ (T/300)^{-3.08} exp (2114/T)

k₁₅₉ = 100 × k₂₁₀

k₁₆₀ = 100 × k₂₁₀

k₁₆₁ = 4.33 × 10⁻³² (T/300)^{-1.0}

k₁₆₂ = 2.56 × 10⁻³¹ (T/300)^{-2.0}

k₁₆₃ = 9.2 × 10⁻³⁴ (T/300)^{-1.0}

k₁₆₄ = 2.0 × 10⁻¹¹ (T/300)^{0.44}

k₁₆₅ = 3.0 × 10⁻¹⁸ T^{0.5} f_A [1.0 + 0.04(T + T_d)^{0.5}]

+ 0.002 T + 8 × 10⁻⁶ T²]⁻¹

k₁₂₉ = 2.76 × 10⁻⁷ (T/300)^{-0.64}

k₁₃₀ = 2.4 × 10⁻⁸ (T/300)^{-0.64}

k₁₃₁ = 1.1 × 10⁻⁷ (T/300)^{-1.0}

k₁₃₂ = 1.0 × 10⁻⁹

k₁₃₃ = 1.0 × 10⁻⁹

k₁₃₄ = 1.0 × 10⁻¹⁰

k₁₃₅ = 5.0 × 10⁻¹⁰

k₁₃₆ = 1.0 × 10⁻¹³

k₁₃₇ = 5.0 × 10⁻¹⁰

k₁₃₈ = 5.0 × 10⁻¹⁰

k₁₃₉ = 7.0 × 10⁻¹⁰

k₁₄₀ = 5.0 × 10⁻¹⁰

k₈₇ = 2.5 × 10⁻⁹

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chemical model 2

Table B1.

No.	Reaction
1	H +

14	$H^- + H \rightarrow H + H + e^-$	88	$H_2 + He^+ \rightarrow He + H_2^+$	$k_{88} = 7.2 \times 10^{-15}$	63
36	$CH + H_2$	89	$H_2 + He^+ \rightarrow He + H + H^+$	$k_{89} = 3.7 \times 10^{-14} \exp\left(\frac{35}{T}\right)$	63
37	$CH + C$	90	$CH + H^+ \rightarrow CH^+ + H$	$k_{90} = 1.9 \times 10^{-9}$	28
38	$CH + C$	91	$CH_2 + H^+ \rightarrow CH_2^+ + H$	$k_{91} = 1.4 \times 10^{-9}$	28
39	$C + H_2$	92	$CH_2 + H^+ \rightarrow CH + H_2^+$	$k_{92} = 7.2 \times 10^{-10}$	28
40	$CH_2 + O$	93	$C_2 + H^+ \rightarrow C_2^+ + H$	$k_{93} = 1.1 \times 10^{-9}$	28
41	$CH_2 + O$	94	$OH + H^+ \rightarrow OH^+ + H$	$k_{94} = 2.1 \times 10^{-9}$	28
42	$C_2 + O$	95	$OH + He^+ \rightarrow O^+ + He + H$	$k_{95} = 1.1 \times 10^{-9}$	28
		96	$H_2O + H^+ \rightarrow H_2O^+ + H$	$k_{96} = 6.9 \times 10^{-9}$	64
		97	$H_2O + He^+ \rightarrow OH + He + H^+$	$k_{97} = 2.04 \times 10^{-10}$	65
		98	$H_2O + He^+ \rightarrow OH^+ + He + H$	$k_{98} = 2.04 \times 10^{-10}$	65

Table B2. List of photochemical reactions included in our chemical model

No.	Reaction	Optically thin rate (s^{-1})	γ	Ref.
166	$H^- + \gamma \rightarrow H + e^-$	$R_{166} = 7.1 \times 10^{-7}$	0.5	1
167	$H_2^+ + \gamma \rightarrow H + H^+$	$R_{167} = 1.1 \times 10^{-9}$	1.9	2
168	$H_2 + \gamma \rightarrow H + H$	$R_{168} = 5.6 \times 10^{-11}$	See §2.2	3
169	$H_3^+ + \gamma \rightarrow H_2 + H^+$	$R_{169} = 4.9 \times 10^{-13}$	1.8	4
170	$H_3^+ + \gamma \rightarrow H_2^+ + H$	$R_{170} = 4.9 \times 10^{-13}$	2.3	4
171	$C + \gamma \rightarrow C^+ + e^-$	$R_{171} = 3.1 \times 10^{-10}$	3.0	5
172	$C^- + \gamma \rightarrow C + e^-$	$R_{172} = 2.4 \times 10^{-7}$	0.9	6
173	$CH + \gamma \rightarrow C + H$	$R_{173} = 8.7 \times 10^{-10}$	1.2	7
174	$CH + \gamma \rightarrow CH^+ + e^-$	$R_{174} = 7.7 \times 10^{-10}$	2.8	8
175	$CH^+ + \gamma \rightarrow C + H^+$	$R_{175} = 2.6 \times 10^{-10}$	2.5	7
176	$CH_2 + \gamma \rightarrow CH + H$	$R_{176} = 7.1 \times 10^{-10}$	1.7	7
177	$CH_2 + \gamma \rightarrow CH_2^+ + e^-$	$R_{177} = 5.9 \times 10^{-10}$	2.3	6
178	$CH_2^+ + \gamma \rightarrow CH^+ + H$	$R_{178} = 4.6 \times 10^{-10}$	1.7	9
179	$CH_3^+ + \gamma \rightarrow CH_2^+ + H$	$R_{179} = 1.0 \times 10^{-9}$	1.7	6
180	$CH_3^+ + \gamma \rightarrow CH^+ + H_2$	$R_{180} = 1.0 \times 10^{-9}$	1.7	6
181	$C_2 + \gamma \rightarrow C + C$	$R_{181} = 1.5 \times 10^{-10}$	2.1	7
182	$O^- + \gamma \rightarrow O + e^-$	$R_{182} = 2.4 \times 10^{-7}$	0.5	6
183	$OH + \gamma \rightarrow O + H$	$R_{183} = 3.7 \times 10^{-10}$	1.7	10
184	$OH + \gamma \rightarrow OH^+ + e^-$	$R_{184} = 1.6 \times 10^{-12}$	3.1	6
185	$OH^+ + \gamma \rightarrow O + H^+$	$R_{185} = 1.0 \times 10^{-12}$	1.8	4
186	$H_2O + \gamma \rightarrow OH + H$	$R_{186} = 6.0 \times 10^{-10}$	1.7	11
187	$H_2O + \gamma \rightarrow H_2O^+ + e^-$	$R_{187} = 3.2 \times 10^{-11}$	3.9	8
188	$H_2O^+ + \gamma \rightarrow H_2^+ + O$	$R_{188} = 5.0 \times 10^{-11}$	See §2.2	12
189	$H_2O^+ + \gamma \rightarrow H^+ + OH$	$R_{189} = 5.0 \times 10^{-11}$	See §2.2	12
190	$H_2O^+ + \gamma \rightarrow O^+ + H_2$	$R_{190} = 5.0 \times 10^{-11}$	See §2.2	12
191	$H_2O^+ + \gamma \rightarrow OH^+ + H$	$R_{191} = 1.5 \times 10^{-10}$	See §2.2	12
192	$H_3O^+ + \gamma \rightarrow H^+ + H_2O$	$R_{192} = 2.5 \times 10^{-11}$	See §2.2	12
193	$H_3O^+ + \gamma \rightarrow H_2^+ + OH$	$R_{193} = 2.5 \times 10^{-11}$	See §2.2	12
194	$H_3O^+ + \gamma \rightarrow H_2O^+ + H$	$R_{194} = 7.5 \times 10^{-12}$	See §2.2	12
195	$H_3O^+ + \gamma \rightarrow OH^+ + H_2$	$R_{195} = 2.5 \times 10^{-11}$	See §2.2	12
196	$O_2 + \gamma \rightarrow O_2^+ + e^-$	$R_{196} = 5.6 \times 10^{-11}$	3.7	7
197	$O_2 + \gamma \rightarrow O + O$	$R_{197} = 7.0 \times 10^{-10}$	1.8	7
198	$CO + \gamma \rightarrow C + O$	$R_{198} = 2.0 \times 10^{-10}$	See §2.2	13

25×10^{-15}	81
0×10^{-17}	82
0×10^{-17}	82
$36 \times 10^{-18} \left(\frac{T}{300}\right)^{0.35} \exp\left(-\frac{161.3}{T}\right)$	83
1×10^{-19}	84
$09 \times 10^{-17} \left(\frac{T}{300}\right)^{0.33} \exp\left(-\frac{1629}{T}\right)$	85
$46 \times 10^{-16} T^{-0.5} \exp\left(-\frac{4.93}{T^{2/3}}\right)$	86
$0 \times 10^{-16} \left(\frac{T}{300}\right)^{-0.2}$	87
5×10^{-18}	84
$14 \times 10^{-18} \left(\frac{T}{300}\right)^{-0.15} \exp\left(\frac{68}{T}\right)$	$T \leq 300$ K
5×10^{-15}	28
$9 \times 10^{-19} \left(\frac{T}{300}\right)^{-0.38}$	28
$9 \times 10^{-20} \left(\frac{T}{300}\right)^{1.58}$	82
$26 \times 10^{-18} \left(\frac{T}{300}\right)^{-5.22} \exp\left(-\frac{90}{T}\right)$	88
$32 \times 10^{-32} \left(\frac{T}{300}\right)^{-0.38}$	$T \leq 300$ K
$32 \times 10^{-32} \left(\frac{T}{300}\right)^{-1.0}$	$T > 300$ K
$8 \times 10^{-31} T^{-0.6}$	91
$9 \times 10^{-32} T^{-0.4}$	92
$99 \times 10^{-33} \left(\frac{T}{5000}\right)^{-1.6}$	$T \leq 5000$ K
$99 \times 10^{-33} \left(\frac{T}{5000}\right)^{-0.64} \exp\left(\frac{5255}{T}\right)$	$T > 5000$ K
$16 \times 10^{-29} \left(\frac{T}{300}\right)^{-3.08}$	$T \leq 2000$ K
$14 \times 10^{-29} \left(\frac{T}{300}\right)^{-3.08} \exp\left(\frac{2114}{T}\right)$	$T > 2000$ K
$10 \times k_{210}$	67
$10 \times k_{210}$	67
$33 \times 10^{-32} \left(\frac{T}{300}\right)^{-1.0}$	43
$56 \times 10^{-31} \left(\frac{T}{300}\right)^{-2.0}$	35
$2 \times 10^{-34} \left(\frac{T}{300}\right)^{-1.0}$	37
$0 \times 10^{-11} \left(\frac{T}{300}\right)^{0.44}$	95
$0 \times 10^{-18} T^{0.5} f_A [1.0 + 0.04(T + T_d)^{0.5}]$	$f_A = [1.0 + 10^4 \exp(-\frac{600}{T_d})]^{-1}$
$0.002 T + 8 \times 10^{-6} T^2 - 1$	96

$5 \times 10^{-7} \left(\frac{T}{300}\right)^{-0.64}$	79
$\times 10^{-8} \left(\frac{T}{300}\right)^{-0.64}$	79
$\times 10^{-7} \left(\frac{T}{300}\right)^{-1.0}$	28
$\times 10^{-9}$	28
$\times 10^{-9}$	28
$\times 10^{-10}$	28
$\times 10^{-10}$	28
$\times 10^{-13}$	28
$\times 10^{-10}$	28
$\times 10^{-10}$	28
$\times 10^{-10}$	28
$\times 10^{-10}$	28
$\times 10^{-10}$	28

86	$HCO^+ + C$	140	$O^- + C \rightarrow CO + e^-$	$k_{140} = 5.0 \times 10^{-10}$	62
87	$HCO^+ + H_2O$		$\rightarrow CO + H_3O^+$	$k_{87} = 2.5 \times 10^{-10}$	

(Glover, Federrath, Mac Low, Klessen, in prep)



chemical model 2

Table B1.

No.	Reaction
1	H +

14	$H^- + H \rightarrow H + H + e^-$	88	$H_2 + He^+ \rightarrow He + H_2^+$	$k_{88} = 7.2 \times 10^{-15}$	63
36	$CH + H_2$	89	$H_2 + He^+ \rightarrow He + H + H^+$	$k_{89} = 3.7 \times 10^{-14} \exp\left(\frac{35}{T}\right)$	63
37	$CH + C$	90	$CH + H^+ \rightarrow CH^+ + H$	$k_{90} = 1.9 \times 10^{-9}$	28
38	$CH + C$	91	$CH_2 + H^+ \rightarrow CH_2^+ + H$	$k_{91} = 1.4 \times 10^{-9}$	28
39	$C + H$	92	$CH_2 + H^+ \rightarrow CH + H_2$	$k_{92} = 7.7 \times 10^{-10}$	28
40	$CH_2 + O$	93	$C_2 + H^+ \rightarrow C_2^+ + H$	$k_{93} = 1.5 \times 10^{-9}$	28
41	$CH_2 + O$	94	$OH + H^+ \rightarrow OH^+ + H$	$k_{94} = 2.1 \times 10^{-9}$	28
42	$C_2 + O \rightarrow$	95	$OH + He^+ \rightarrow O^+ + He + H$	$k_{95} = 1.1 \times 10^{-9}$	28
		96	$H_2O + H^+ \rightarrow H_2O^+ + H$	$k_{96} = 6.9 \times 10^{-9}$	64
		97	$H_2O + He^+ \rightarrow OH + He + H^+$	$k_{97} = 2.04 \times 10^{-10}$	65
		98	$H_2O + He^+ \rightarrow OH^+ + He + H$	$k_{98} = 9.86 \times 10^{-10}$	65

Table B2. List of photochemical reactions included in our chemical model

No.	Reaction	Optically thin rate (s^{-1})	γ	Ref.		
166	$H^- + \gamma \rightarrow H + e^-$	$R_{166} = 7.1 \times 10^{-7}$	0.5	1	$36 \times 10^{-18} \left(\frac{T}{300}\right)^{0.35} \exp\left(-\frac{161.3}{T}\right)$	$T \leq 300$ K
167	$H_2^+ + \gamma \rightarrow H + H^+$	$R_{167} = 1.1 \times 10^{-9}$	1.9	2	$0.9 \times 10^{-17} \left(\frac{T}{300}\right)^{0.33} \exp\left(-\frac{1629}{T}\right)$	$T > 300$ K
168	$H_2 + \gamma \rightarrow H + H$	$R_{168} = 5.6 \times 10^{-11}$	See §2.2	3	$46 \times 10^{-16} T^{-0.5} \exp\left(-\frac{4.93}{T^{2/3}}\right)$	
169	$H_3^+ + \gamma \rightarrow H_2 + H^+$	$R_{169} = 4.9 \times 10^{-13}$	1.8	4	$0 \times 10^{-16} \left(\frac{T}{300}\right)^{-0.2}$	
170	$H_3^+ + \gamma \rightarrow H_2^+ + H$	$R_{170} = 4.9 \times 10^{-13}$	2.3	4	5×10^{-18}	$T \leq 300$ K
171	$C + \gamma \rightarrow C^+ + e^-$	$R_{171} = 2.1 \times 10^{-10}$	2.0	5	$14 \times 10^{-18} \left(\frac{T}{300}\right)^{-0.15} \exp\left(\frac{68}{T}\right)$	$T > 300$ K
172	$C^- + \gamma \rightarrow$					
173	$CH + \gamma \rightarrow$					
174	$CH + \gamma \rightarrow$					
175	$CH^+ + \gamma \rightarrow$					
176	$CH_2 + \gamma \rightarrow$					
177	$CH_2 + \gamma \rightarrow$					
178	$CH_2^+ + \gamma \rightarrow$					
179	$CH_3^+ + \gamma \rightarrow$					
180	$CH_3^+ + \gamma \rightarrow$					
181	$C_2 + \gamma \rightarrow$					
182	$O^- + \gamma \rightarrow$					
183	$OH + \gamma \rightarrow$					
184	$OH + \gamma \rightarrow$					
185	$OH^+ + \gamma \rightarrow$					
186	$H_2O + \gamma \rightarrow$					
187	$H_2O + \gamma \rightarrow$					
188	$H_2O^+ + \gamma \rightarrow$					
189	$H_2O^+ + \gamma \rightarrow$					
190	$H_2O^+ + \gamma \rightarrow$					
191	$H_2O^+ + \gamma \rightarrow$					
192	$H_3O^+ + \gamma \rightarrow$					
193	$H_3O^+ + \gamma \rightarrow$					
194	$H_3O^+ + \gamma \rightarrow$					
195	$H_3O^+ + \gamma \rightarrow$					
196	$O_2 + \gamma \rightarrow$					
197	$O_2 + \gamma \rightarrow O + O$	$R_{197} = 7.0 \times 10^{-10}$	1.8	7	$\times 10^{-13}$	28
198	$CO + \gamma \rightarrow C + O$	$R_{198} = 2.0 \times 10^{-10}$	See §2.2	13	$\times 10^{-10}$	28

Table B3. List of reactions included in our chemical model that involve cosmic rays or cosmic-ray induced UV emission

No.	Reaction	Rate ($s^{-1} \zeta_H^{-1}$)	Ref.
199	$H + c.r. \rightarrow H^+ + e^-$	$R_{199} = 1.0$	—
200	$He + c.r. \rightarrow He^+ + e^-$	$R_{200} = 1.1$	1
201	$H_2 + c.r. \rightarrow H^+ + H + e^-$	$R_{201} = 0.037$	1
202	$H_2 + c.r. \rightarrow H + H$	$R_{202} = 0.22$	1
203	$H_2 + c.r. \rightarrow H^+ + H^-$	$R_{203} = 6.5 \times 10^{-4}$	1
204	$H_2 + c.r. \rightarrow H_2^+ + e^-$	$R_{204} = 2.0$	1
205	$C + c.r. \rightarrow C^+ + e^-$	$R_{205} = 3.8$	1
206	$O + c.r. \rightarrow O^+ + e^-$	$R_{206} = 5.7$	1
207	$CO + c.r. \rightarrow CO^+ + e^-$	$R_{207} = 6.5$	1
208	$C + \gamma_{c.r.} \rightarrow C^+ + e^-$	$R_{208} = 2800$	2
209	$CH + \gamma_{c.r.} \rightarrow C + H$	$R_{209} = 4000$	3
210	$CH^+ + \gamma_{c.r.} \rightarrow C^+ + H$	$R_{210} = 960$	3
211	$CH_2 + \gamma_{c.r.} \rightarrow CH_2^+ + e^-$	$R_{211} = 2700$	1
212	$CH_2 + \gamma_{c.r.} \rightarrow CH + H$	$R_{212} = 2700$	1
213	$C_2 + \gamma_{c.r.} \rightarrow C + C$	$R_{213} = 1300$	3
214	$OH + \gamma_{c.r.} \rightarrow O + H$	$R_{214} = 2800$	3
215	$H_2O + \gamma_{c.r.} \rightarrow OH + H$	$R_{215} = 5300$	3
216	$O_2 + \gamma_{c.r.} \rightarrow O + O$	$R_{216} = 4100$	3
217	$O_2 + \gamma_{c.r.} \rightarrow O_2^+ + e^-$	$R_{217} = 640$	3
218	$CO + \gamma_{c.r.} \rightarrow C + O$	$R_{218} = 0.21 T^{1/2} x_{H_2} x_{CO}^{-1/2}$	4

86	$HCO^+ + C$	140	$O^- + C \rightarrow CO + e^-$	$k_{140} = 5.0 \times 10^{-10}$	28
87	$HCO^+ + H_2O$		$\rightarrow CO + H_3O^+$	$k_{87} = 2.5 \times 10^{-10}$	28

(Glover, Federrath, Mac Low, Klessen, 2010, MNRS, 404, 2)



HI to H₂ conversion rate

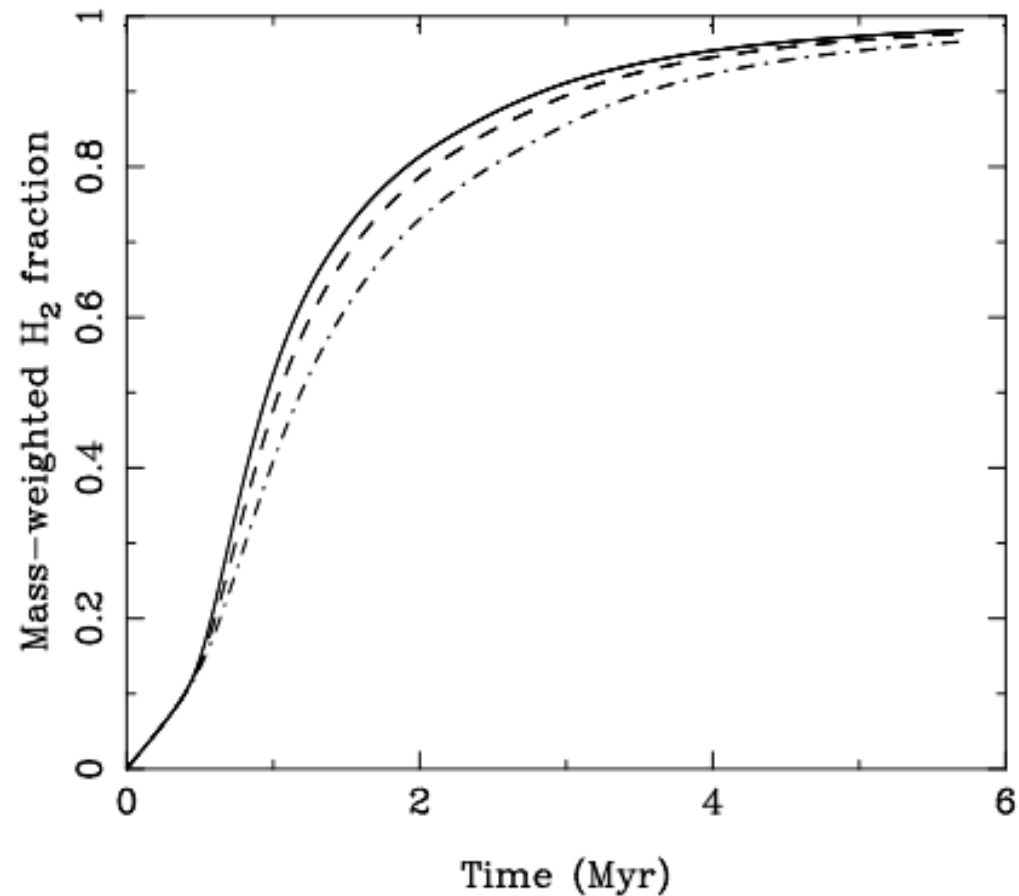
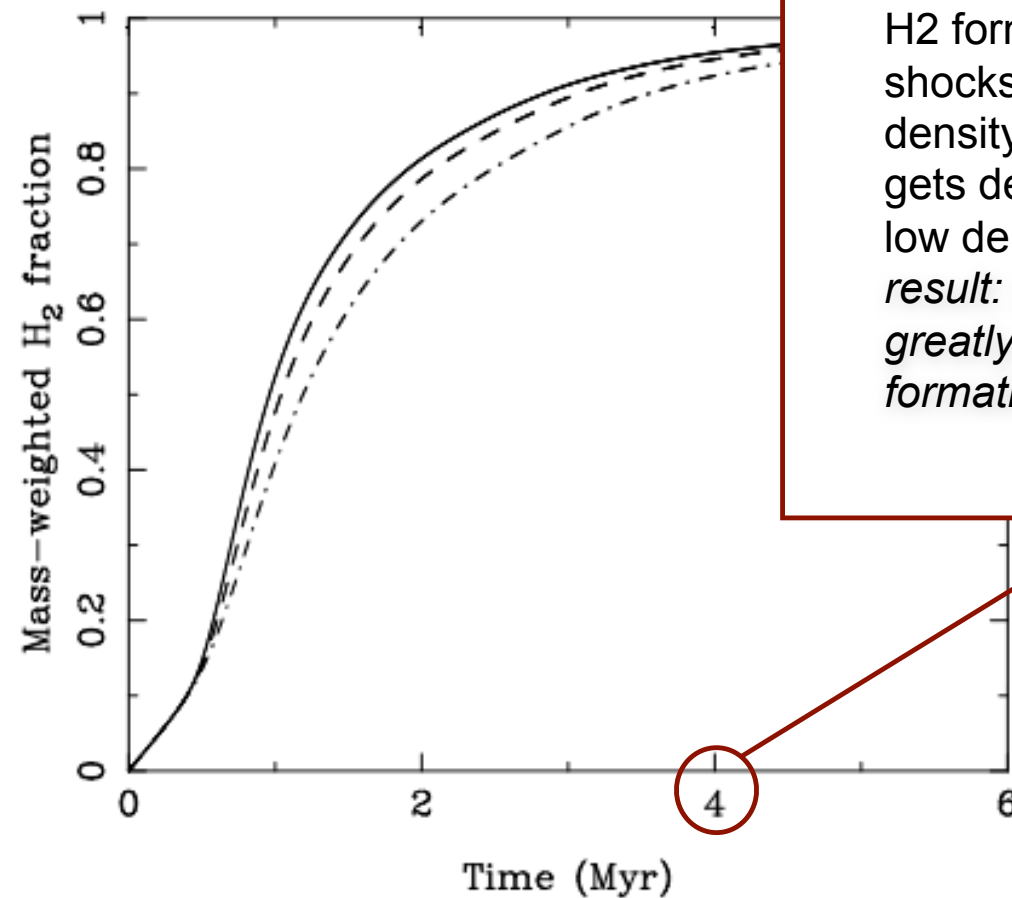


Figure 4. Time evolution of the mass-weighted H₂ abundance in simulations R1, R2 and R3, which have numerical resolutions of 64^3 zones (dot-dashed), 128^3 zones (dashed) and 256^3 zones (solid), respectively.



HI to H₂ conversion rate

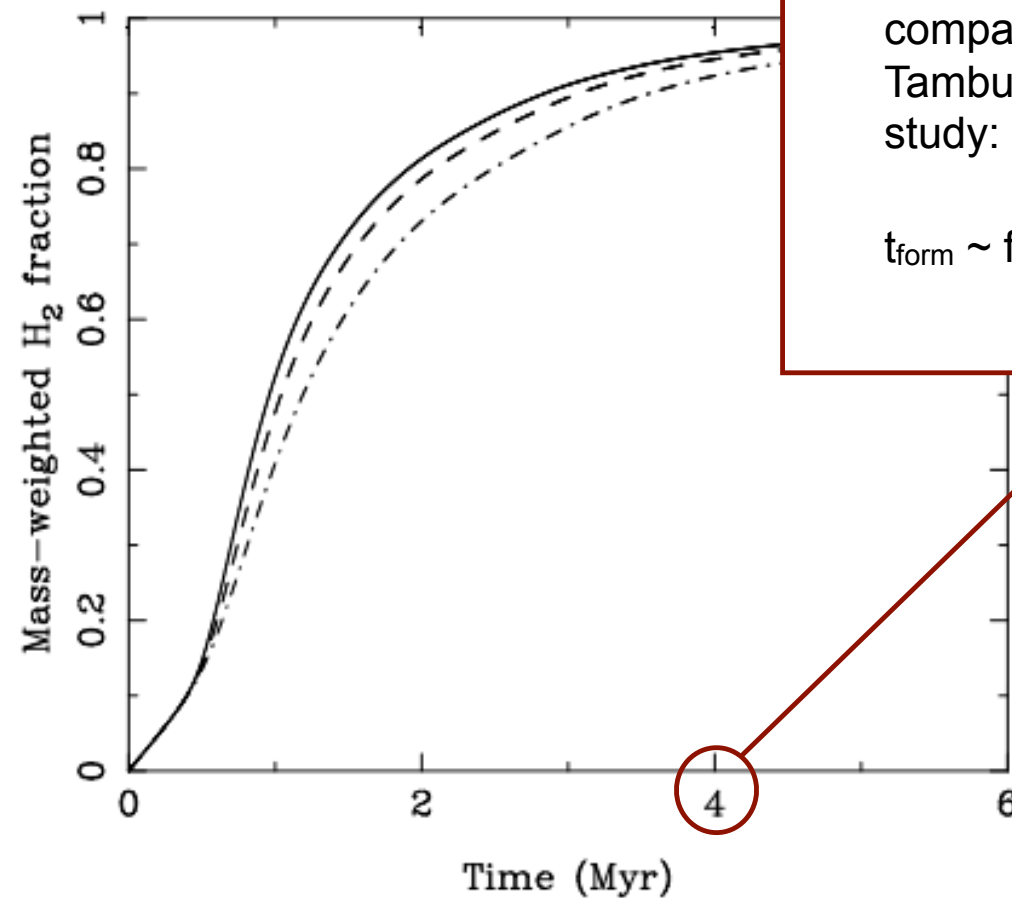


H₂ forms rapidly in shocks / transient density fluctuations / H₂ gets destroyed slowly in low density regions / *result: turbulence greatly enhances H₂-formation rate*

Figure 4. Time evolution of the mass-weighted H₂ abundance in simulations R1, R2 and R3, which have numerical resolutions of 64³ zones (dot-dashed), 128³ zones (dashed) and 256³ zones (solid), respectively.



HI to H₂ conversion rate



compare to data from
Tamburro et al. (2008)
study:

$t_{\text{form}} \sim \text{few} \times 10^6 \text{ years}$

Figure 4. Time evolution of the mass-weighted H₂ abundance in simulations R1, R2 and R3, which have numerical resolutions of 64³ zones (dot-dashed), 128³ zones (dashed) and 256³ zones (solid), respectively.



CO, C⁺ formation rates

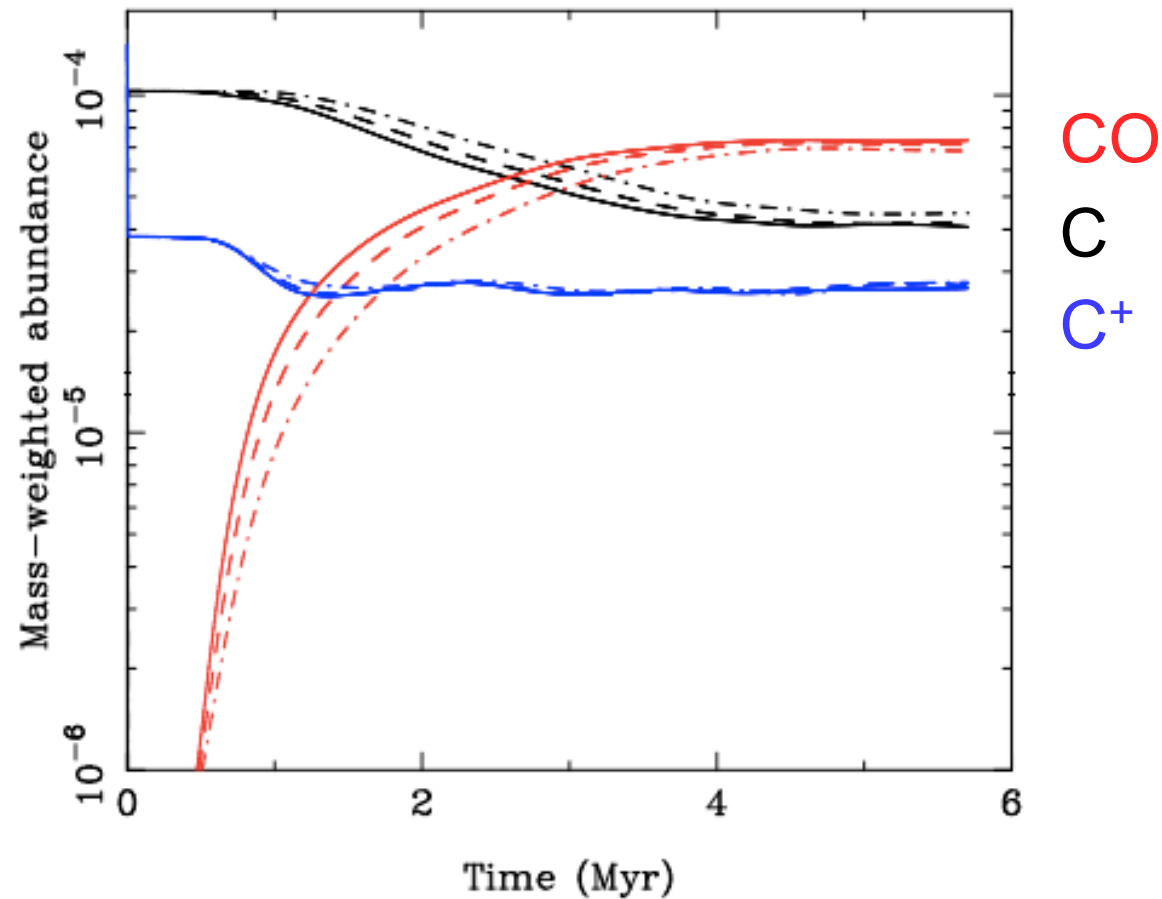
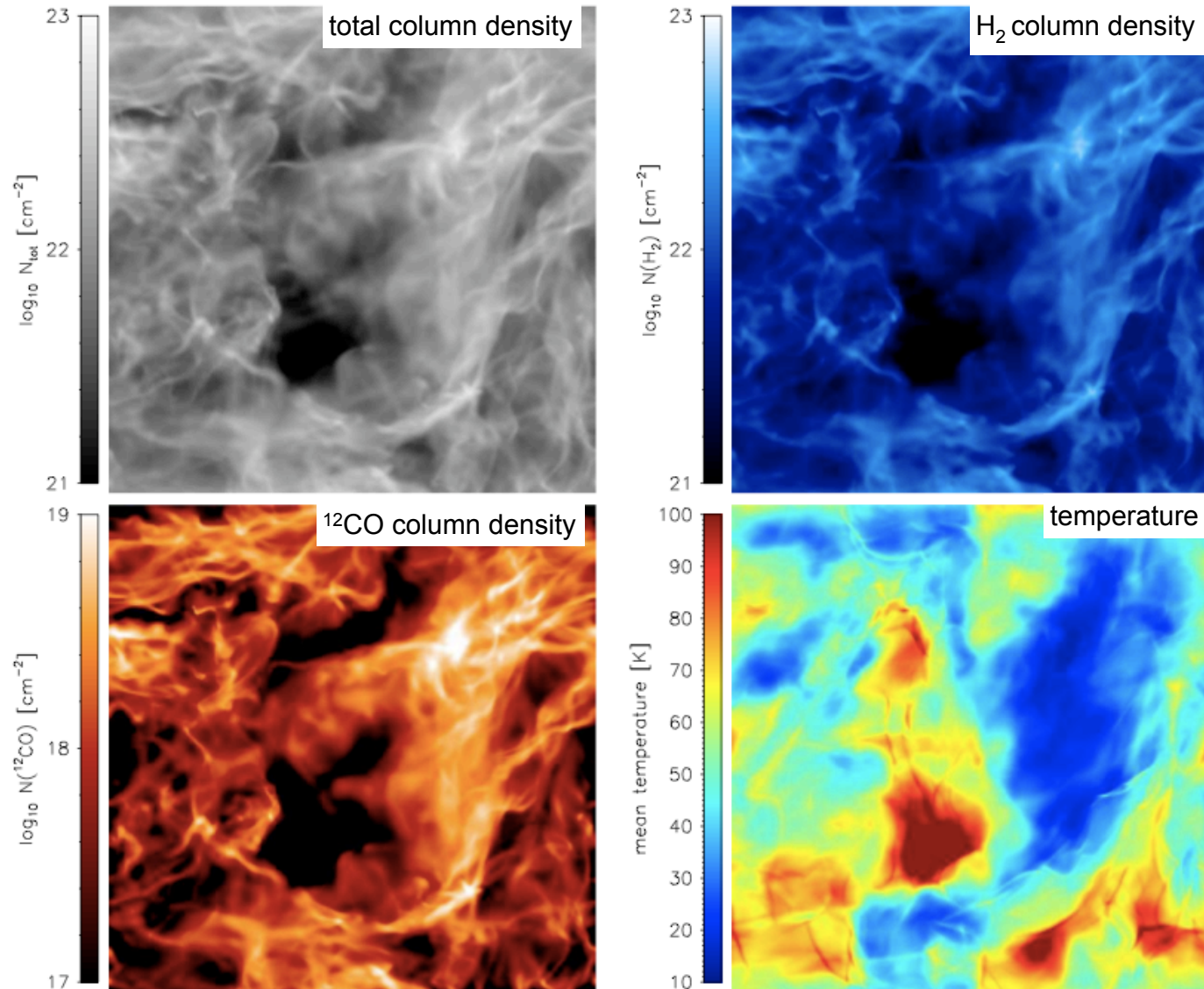


Figure 5. Time evolution of the mass-weighted abundances of atomic carbon (black lines), CO (red lines), and C⁺ (blue lines) in simulations with numerical resolutions of 64³ zones (dot-dashed), 128³ zones (dashed) and 256³ zones (solid).



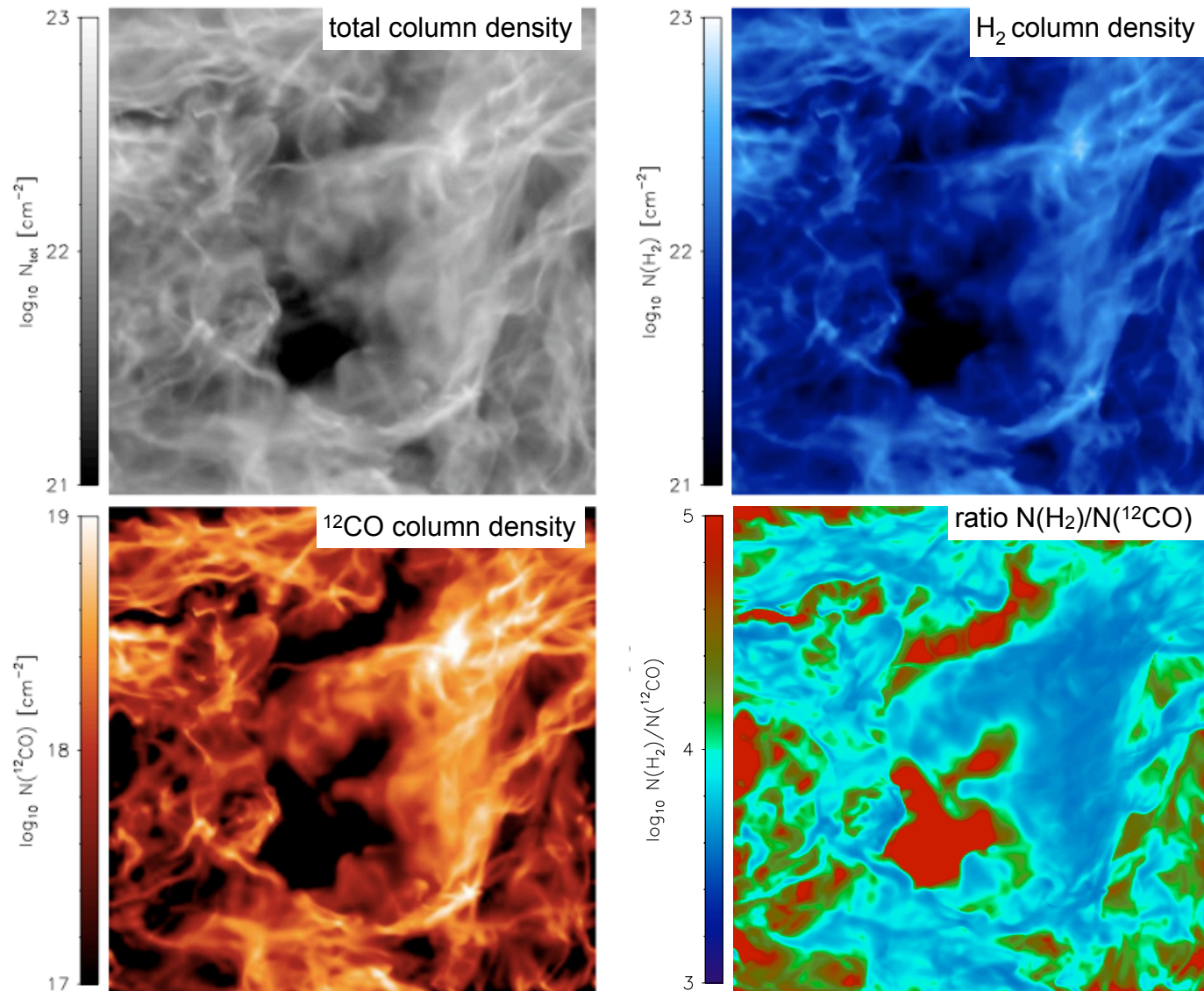
effects of chemistry 1



(Glover, Federrath, Mac Low, Klessen, 2010)



effects of chemistry 2



(Glover, Federrath, Mac Low, Klessen, 2010)



effects of chemistry 4

- deliverables / predictions:
 - x-factor estimates (as function of environmental conditions)
 - synthetic line emission maps (in combination with line transfer)
 - pdf's of density, velocity, emissivity / structure functions (to directly connect to observational regime)
 - *COMMENT:* density pdf is *NOT* lognormal!
--> implications for analytical IMF theories



X-factor

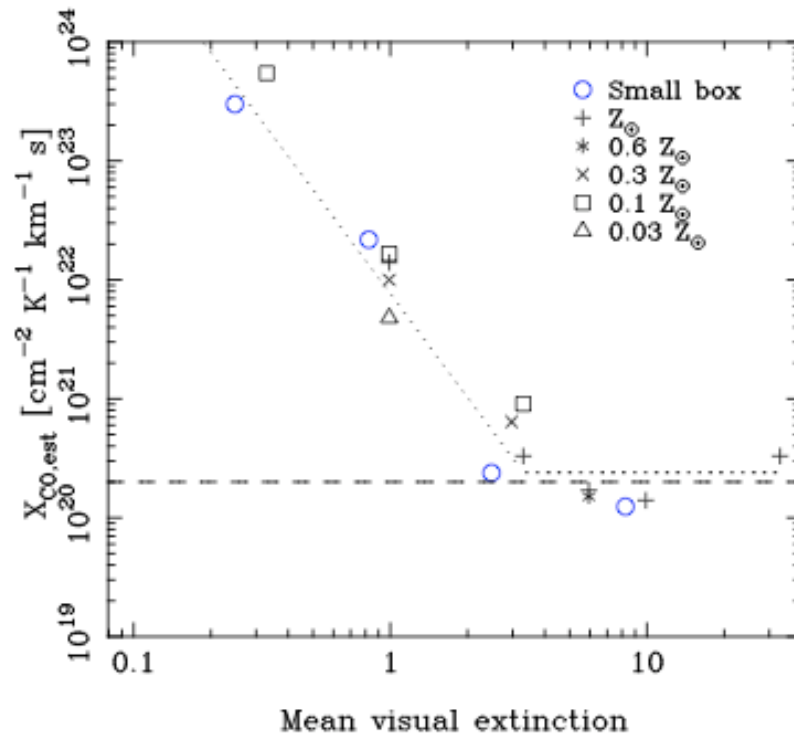


Figure 8. Estimate of the CO-to-H₂ conversion factor $X_{\text{CO,est}}$, plotted as a function of the mean visual extinction of the gas, $\langle A_V \rangle$. The simplifications made in our modelling mean that each value of $X_{\text{CO,est}}$ is uncertain by at least a factor of two. At $\langle A_V \rangle > 3$, the values we find are consistent with the value of $X_{\text{CO}} = 2 \times 10^{20} \text{cm}^{-2} \text{K}^{-1} \text{km}^{-1} \text{s}$ determined observationally for the Milky Way by Dame et al. (2001), indicated in the plot by the horizontal dashed line. At $\langle A_V \rangle < 3$, we find evidence for a strong dependence of $X_{\text{CO,est}}$ on $\langle A_V \rangle$. The empirical fit given by Equation 11 is indicated as the dotted line in the Figure, and demonstrates that at low $\langle A_V \rangle$, the CO-to-H₂ conversion factor increases roughly as $X_{\text{CO,est}} \propto A_V^{-2.8}$. It should also be noted that at any particular $\langle A_V \rangle$, the dependence of $X_{\text{CO,est}}$ on metallicity is relatively small. Previous claims of a strong metallicity dependence likely reflect the fact that there is a strong dependence on the mean extinction, which varies as $\langle A_V \rangle \propto Z$ given fixed mean cloud density and cloud size.

from Glover & Mac Low (2010, ApJ, submitted)

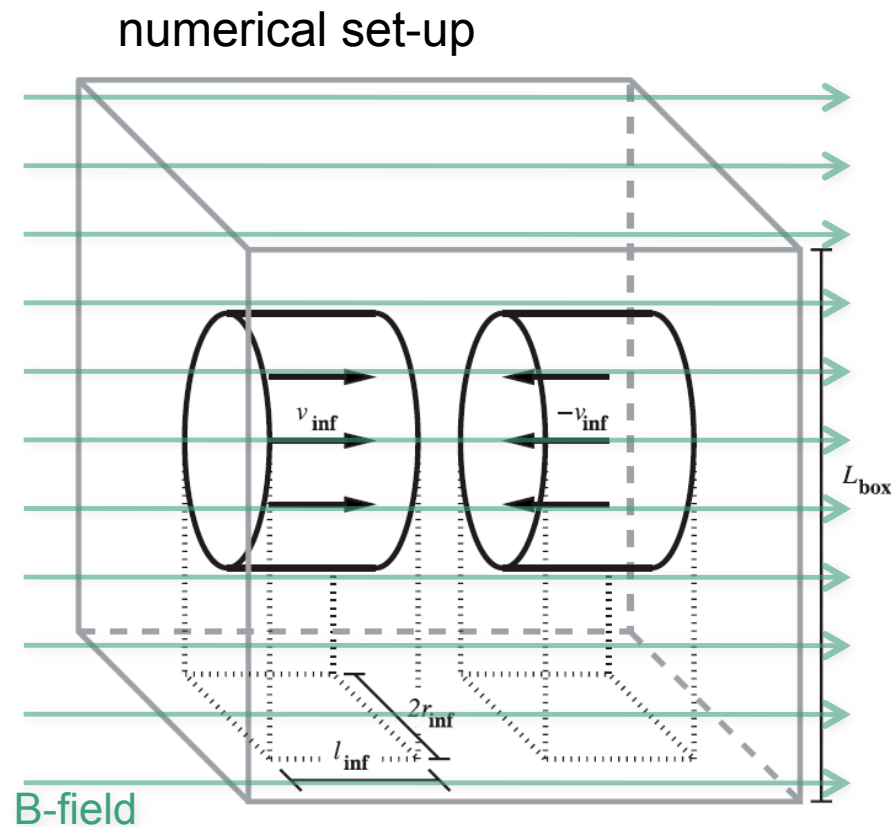


from atomic gas to molecular clouds

- let's look at the details:
 - how does molecular cloud material form in convergent flows, e.g., as triggered by spiral density waves...
 - do sequence of idealized numerical experiments
- questions
 - are molecular clouds truly “multi-phase” media?
 - turbulence? dynamical & morphological properties?
 - what is relation to initial & environmental conditions?
 - magnetic field structure?



convergent flows: set-up



from Vazquez-Semadeni et al. (2007)

● convergent flow studies

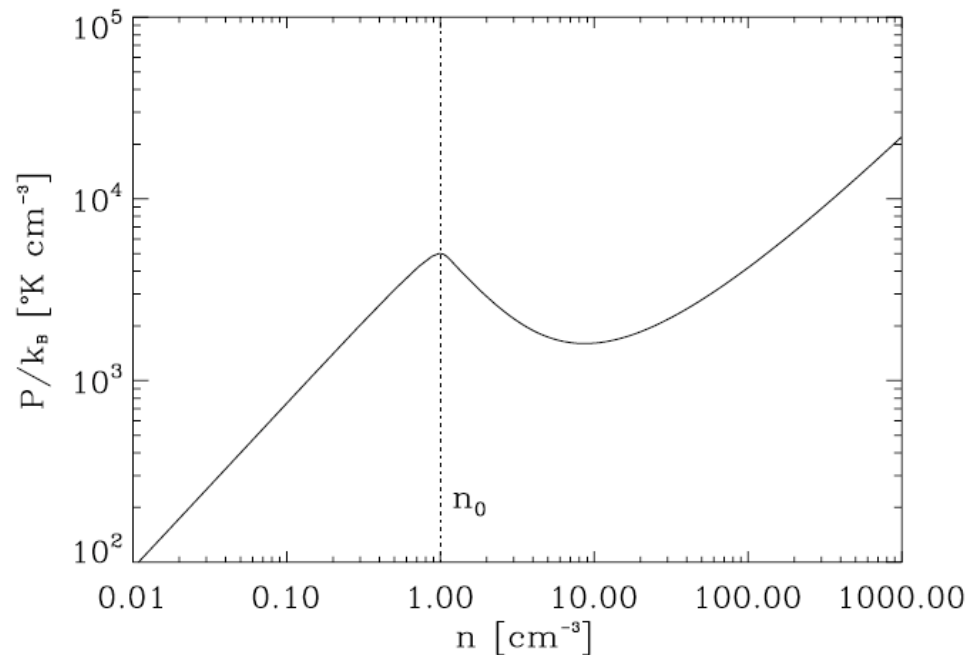
- atomic flows collide
- cooling curve (soon chemistry)
- gravity
- magnetic fields
- numerics: AMR, BGK, SPH

see studies by Banerjee et al., Heitsch et al., Hennebelle et al., Vazquez-Semadeni et al.



convergent flows: set-up

adopted cooling curve



from Vazquez-Semadeni et al. (2007)

● convergent flow studies

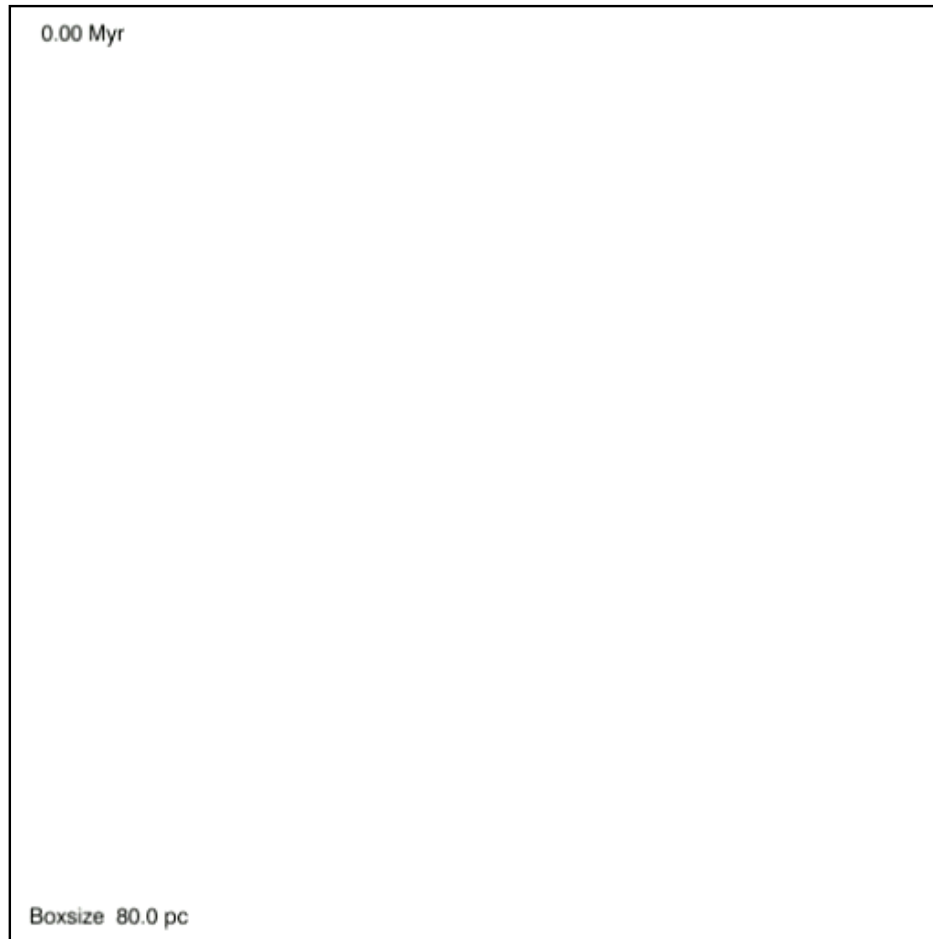
- atomic flows collide
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see studies by Banerjee et al., Heitsch et al., Hennebelle et al., Vazquez-Semadeni et al.

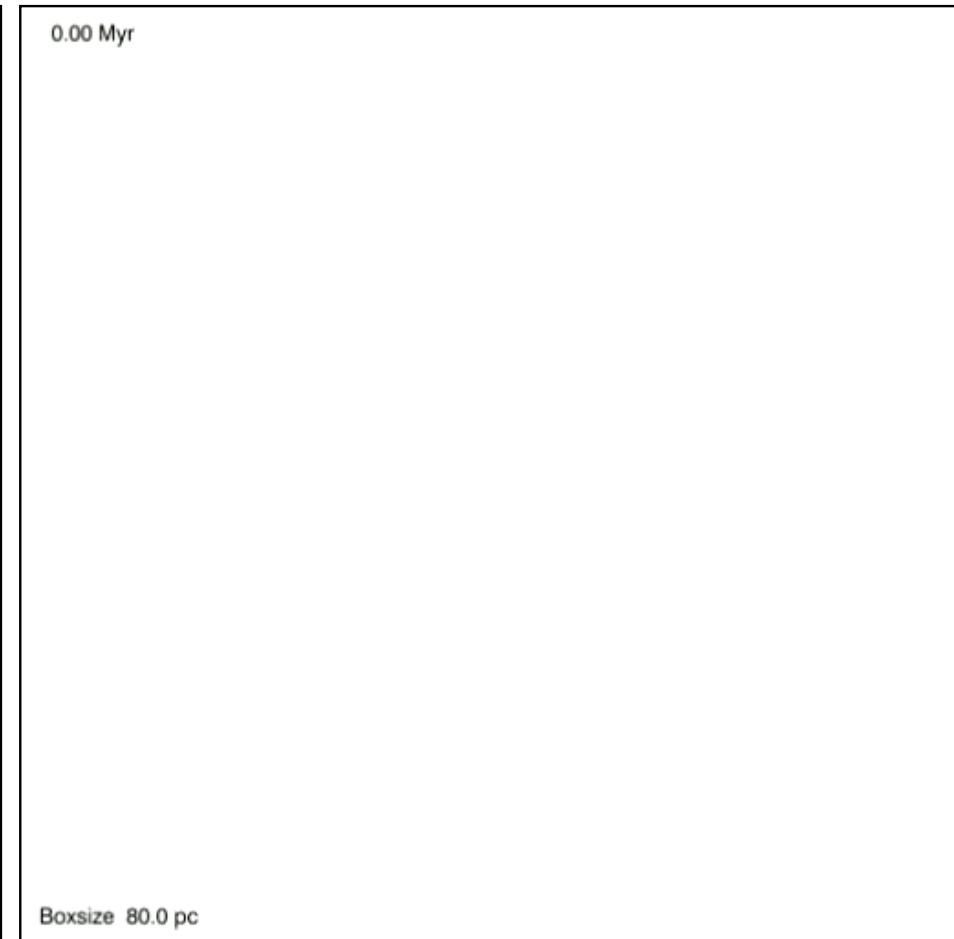


MC formation in convergent flows

the non-magnetic case



-edge-on view



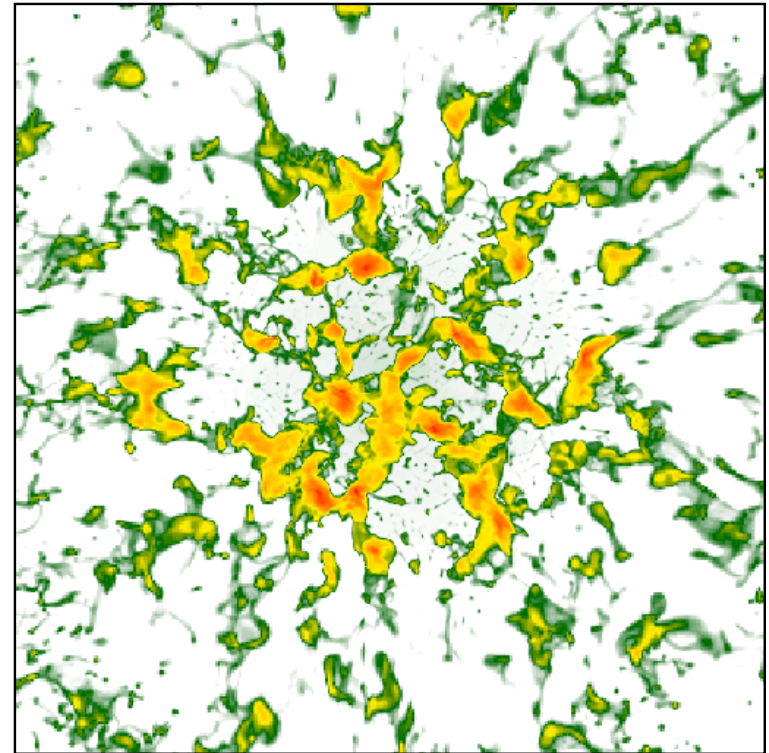
-face-on view

thermal instability + gravity creates complex molecular cloud structure:

MC formation in convergent flows

this simple set-up reproduces (and explains!) some of the main properties of MCs:

- highly **patchy** and **clumpy**
- high fraction of **substructure**
- cold dense molecular clumps **coexist** with warm atomic gas
- not a well bounded entity
- **dynamical** evolution (different star formation modes: from low mass to high mass SF?)

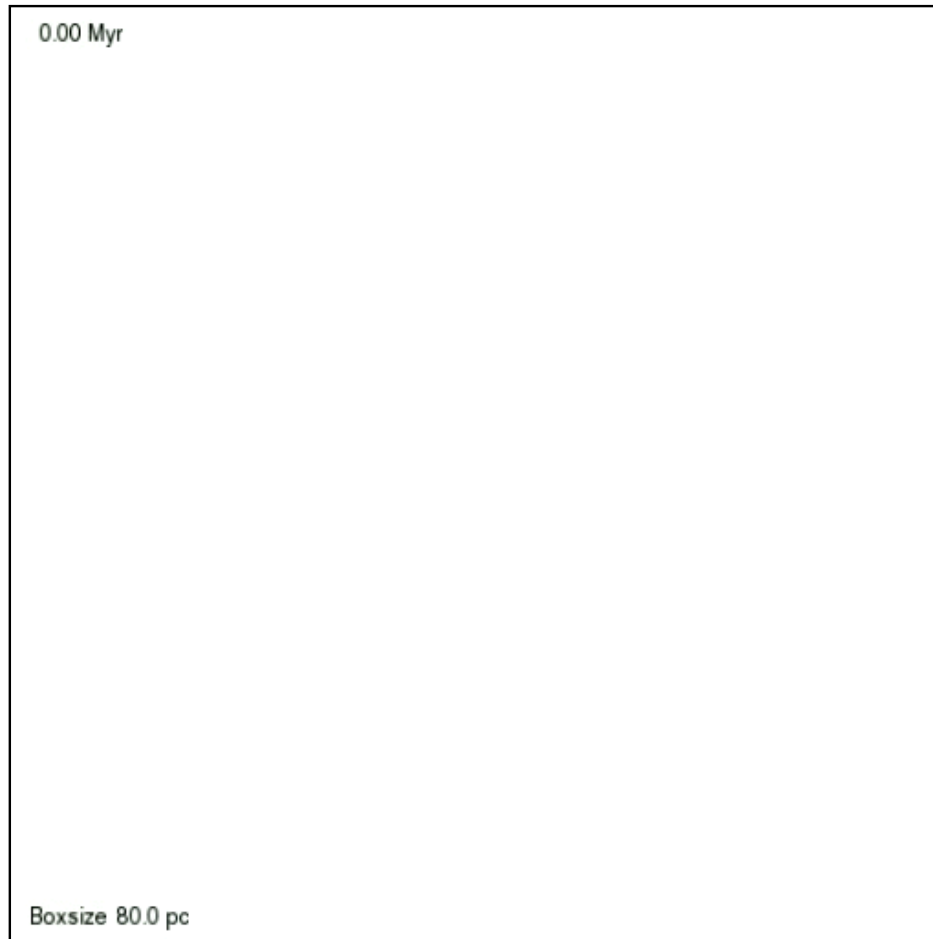


from Banerjee et al. (2008)

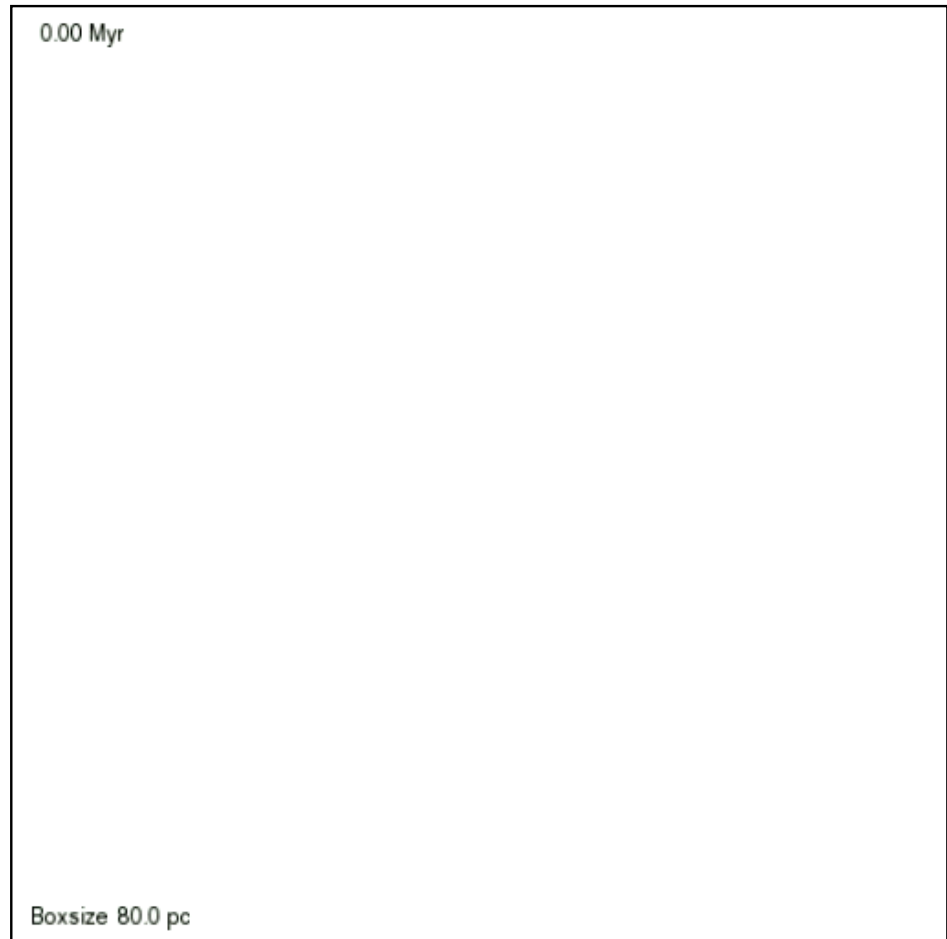
(see also studies by Hennebelle et al. and Vazquez-Semadeni et al. and Heitsch et al.)

MC formation in convergent flows

the weakly magnetized ($B_x = 1\mu\text{G}$) case



edge-on view



face-on view

from Banerjee et al. (2008)

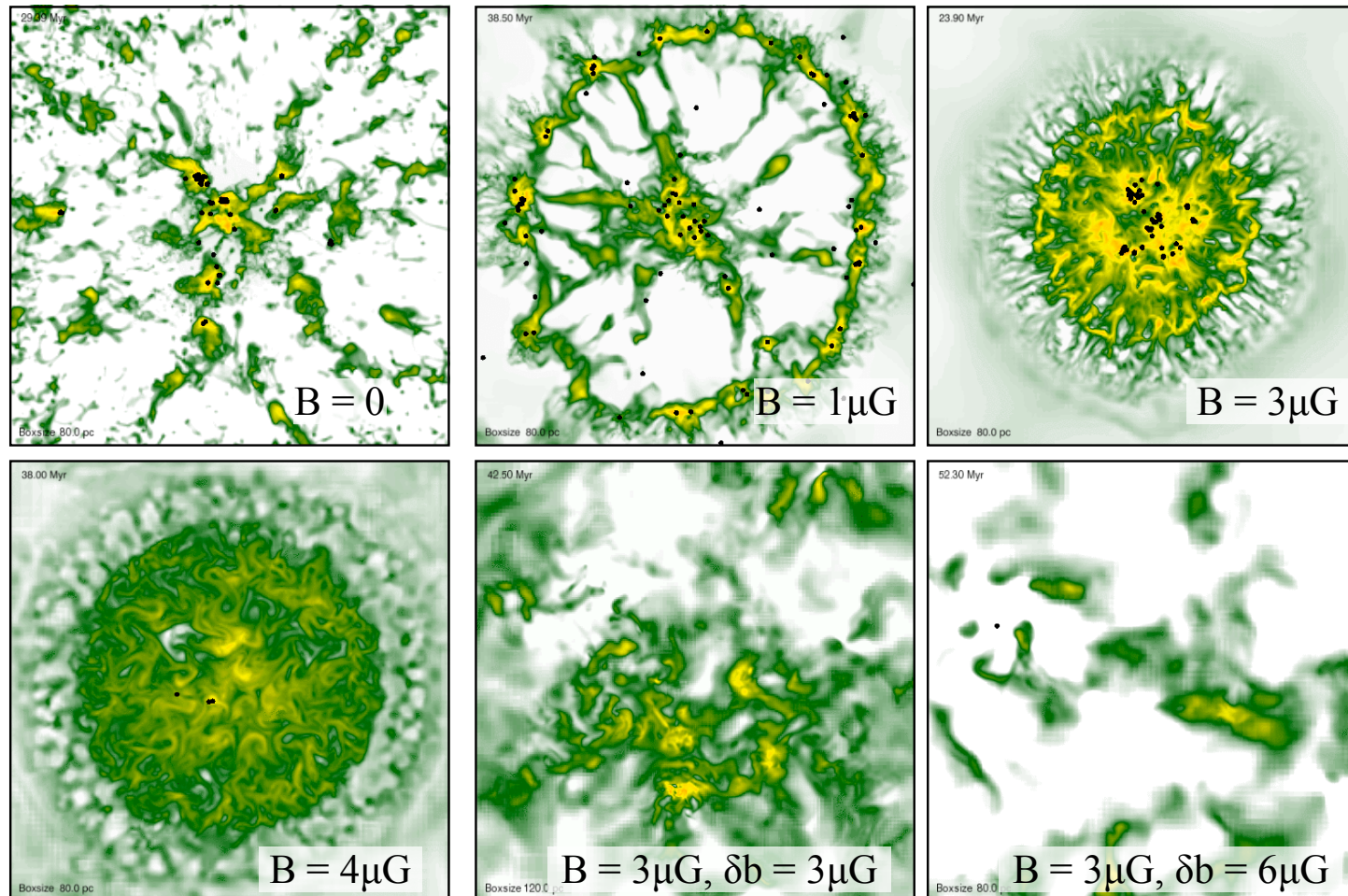
(see also studies by Hennebelle et al. and Vazquez-Semadeni et al. and Heitsch et al.)

MC formation in convergent flows

with random component: $B_x = 3\mu\text{G} + \delta b = 3\mu\text{G}$



MC formation in convergent flows

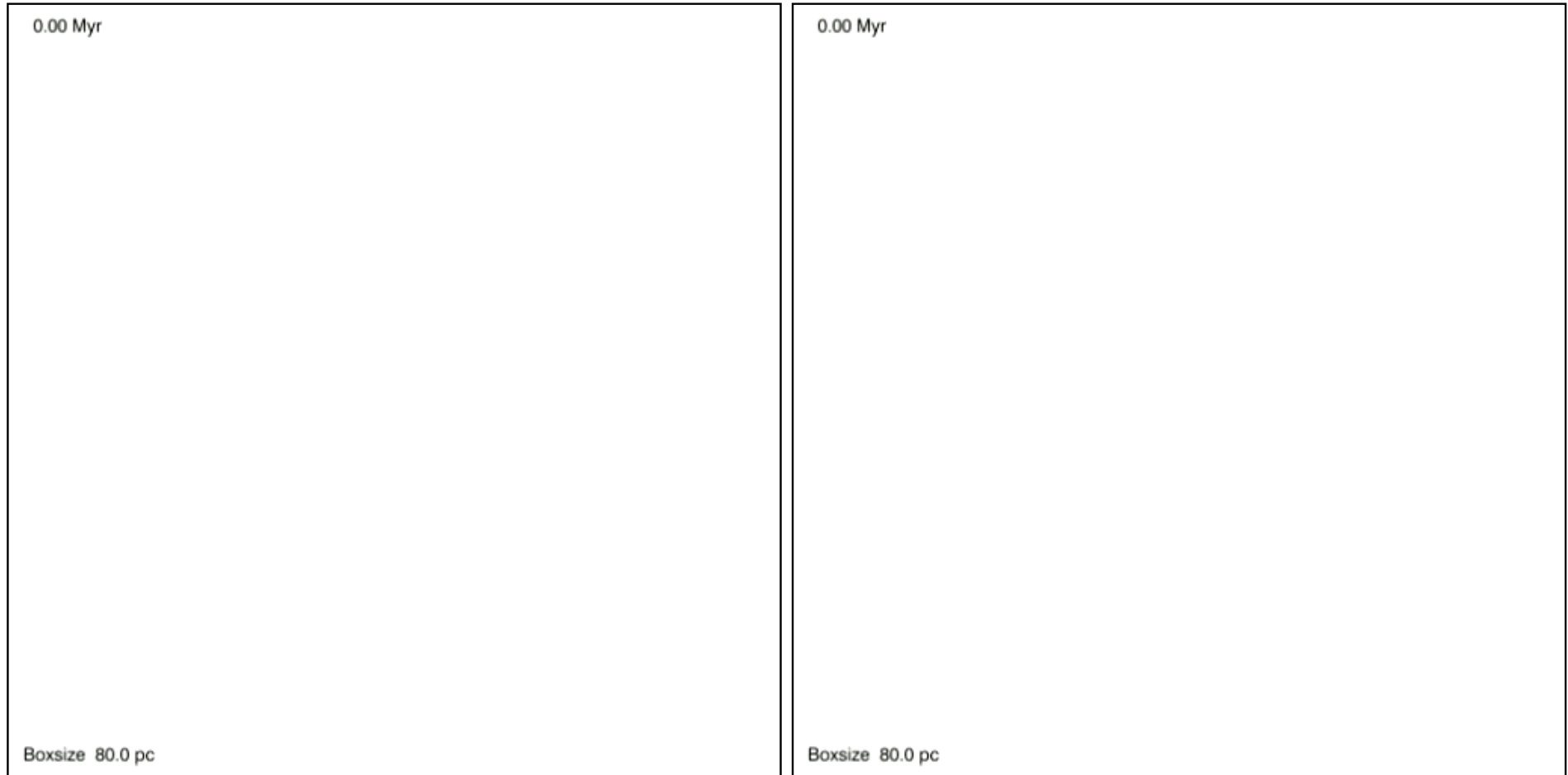


Morphology of the molecular cloud and **star formation efficiency** depends on the strength of the magnetic field

Banerjee et al. in prep.

MC formation in convergent flows

Influence of Ambipolar Diffusion: $B_x = 3\mu\text{G}$ (super-critical)

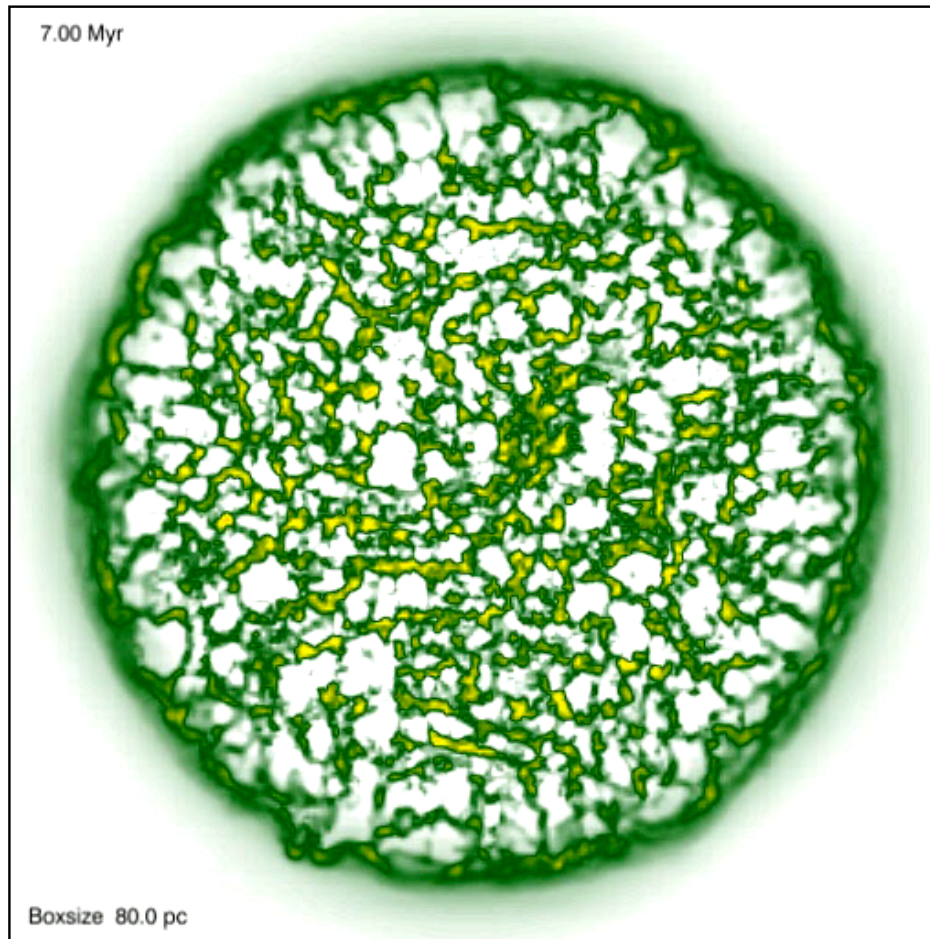


Ideal MHD

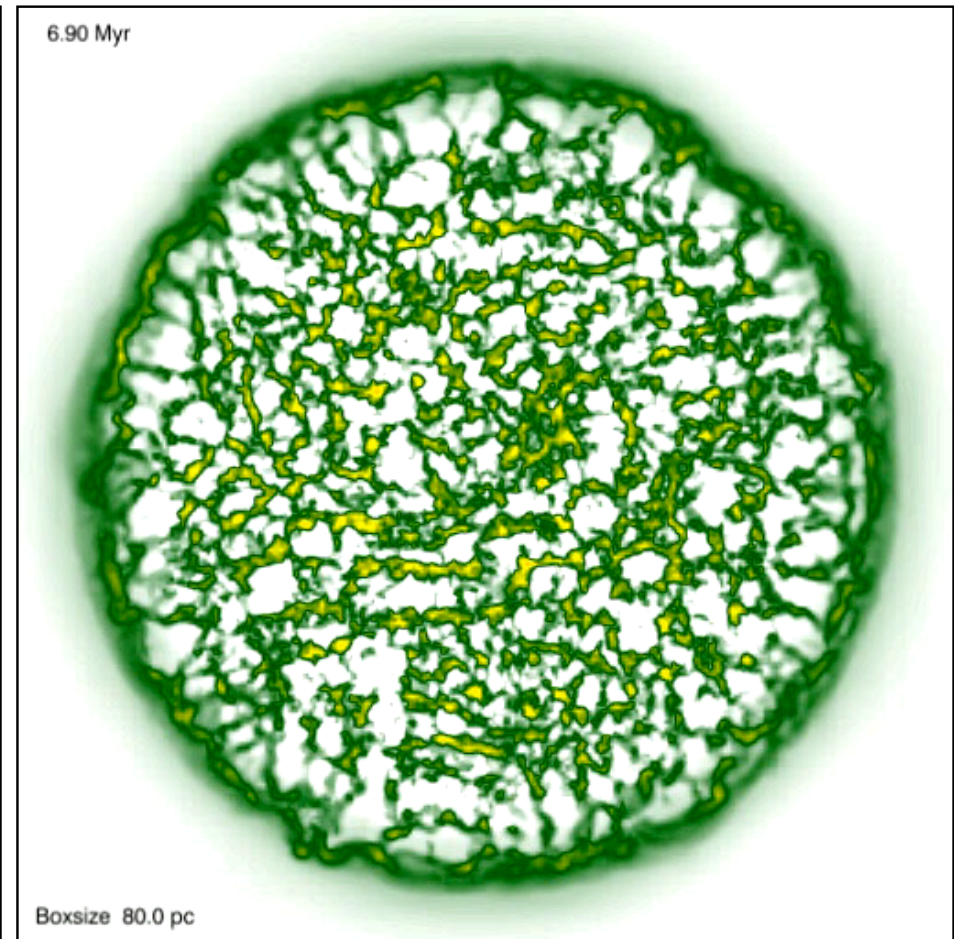
with AD

MC formation in convergent flows

Influence of Ambipolar Diffusion: $B_x = 4\mu\text{G}$ (critical)



Ideal MHD

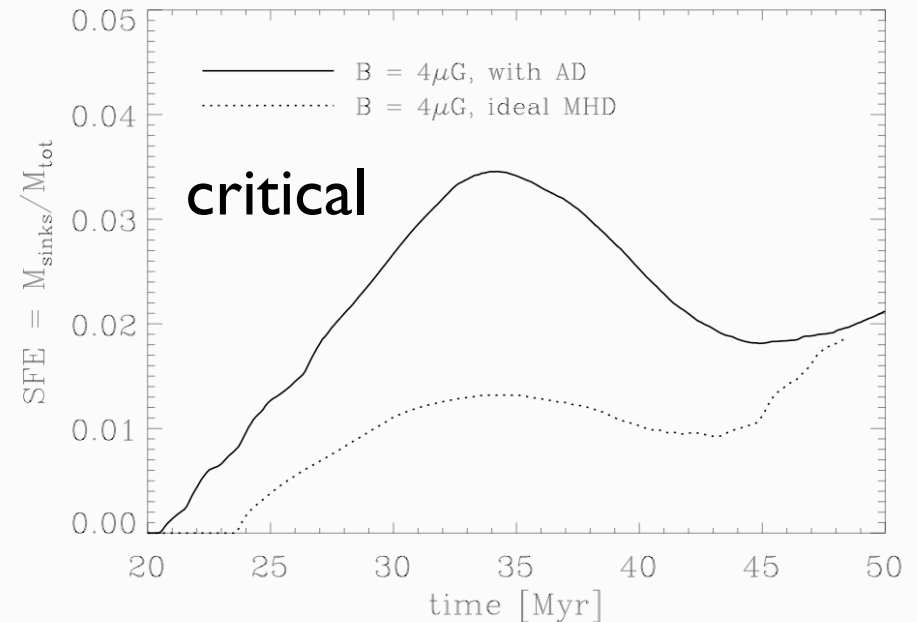
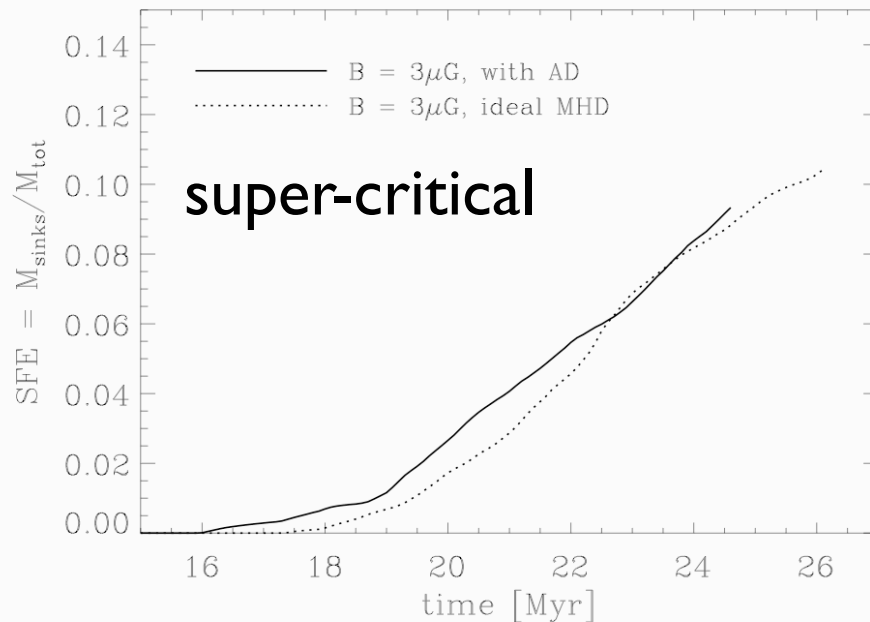


with AD

Banerjee et al. in prep.

MC formation in convergent flows

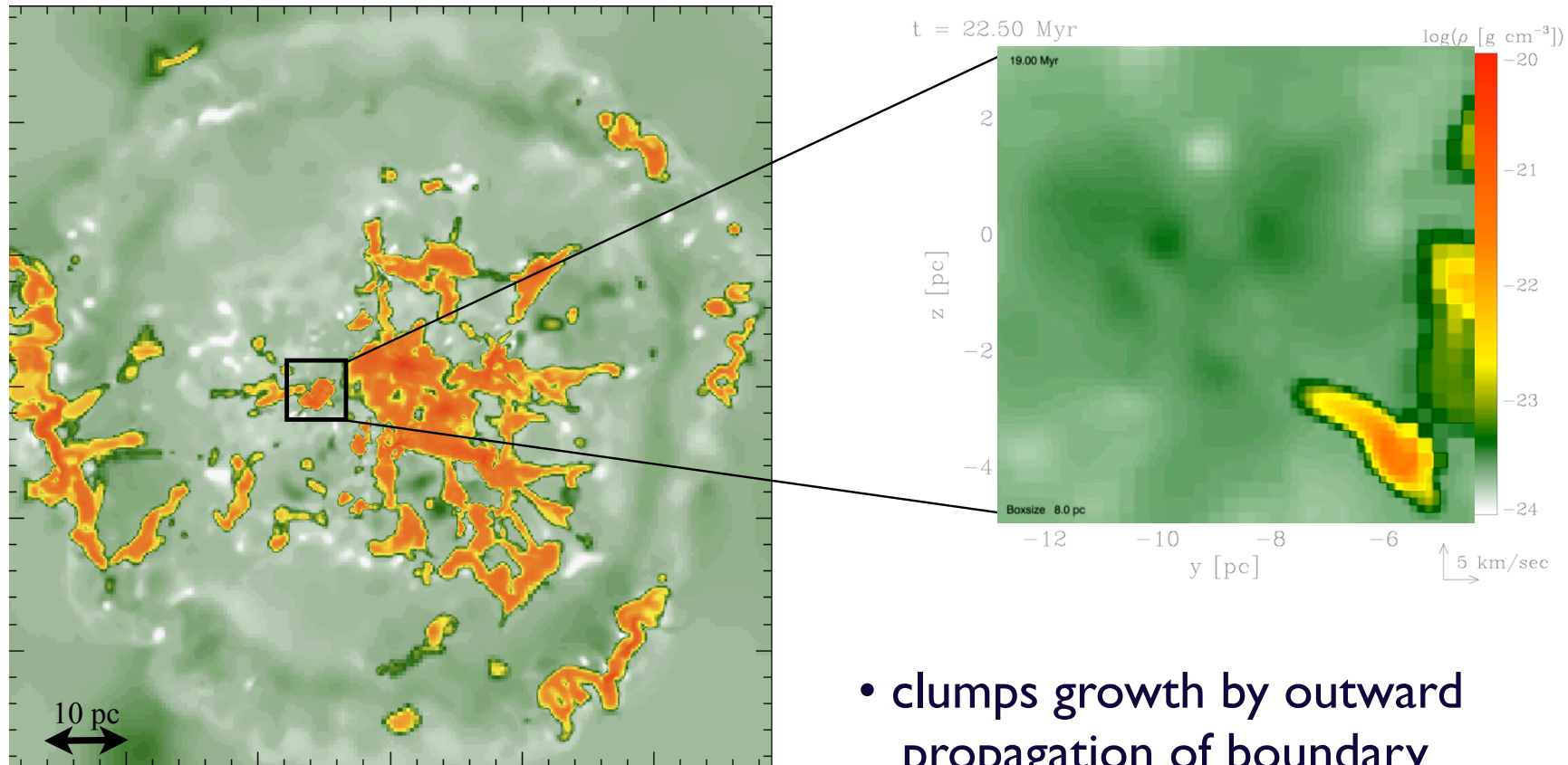
Influence of Ambipolar Diffusion



- Ambipolar diffusion is **not** a major player for star formation on molecular cloud scales
- this is different during protostellar collapse (Hennebelle et al.)

MC formation in convergent flows

morphology and clump evolution



- MCs are inhomogeneous
- cold clumps embedded in warm atomic gas

- clumps growth by outward propagation of boundary layers and
- coalescence at later times

see studies by Banerjee et al., Heitsch et al., Hennebelle et al., Vazquez-Semadeni et al.



some results: growth of cores

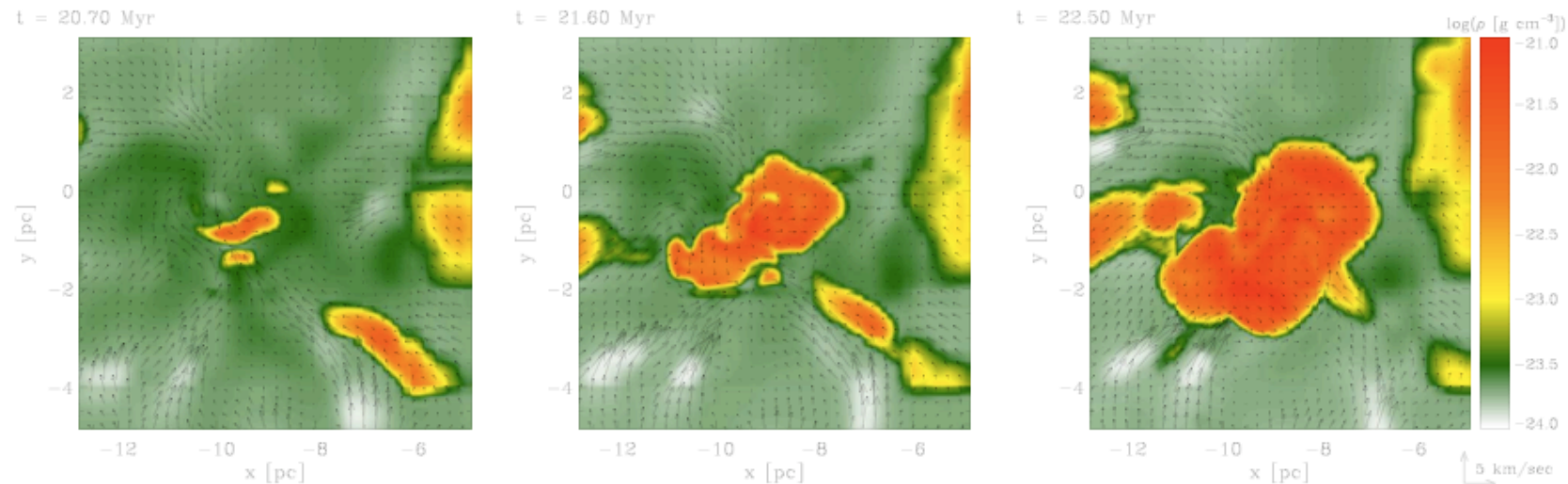


Figure 2. Shows the time evolution of a typical clump which initially develops out of the thermally unstable WNM in shock layers of turbulent flows. A small cold condensate grows by outward propagation of its boundary layer. Coalescence and merging with nearby clumps further increases the size and mass of these clumps. The global gravitational potential of the proto-cloud enhances the merging probability with time. The images show 2D slices of the density (logarithmic colour scale) and the gas velocity (indicated as arrows) in the plane perpendicular to the large scale flows.

two phases of core growth:

(1) *by outward propagation of boundary layer* → Jeans sub-critical phase

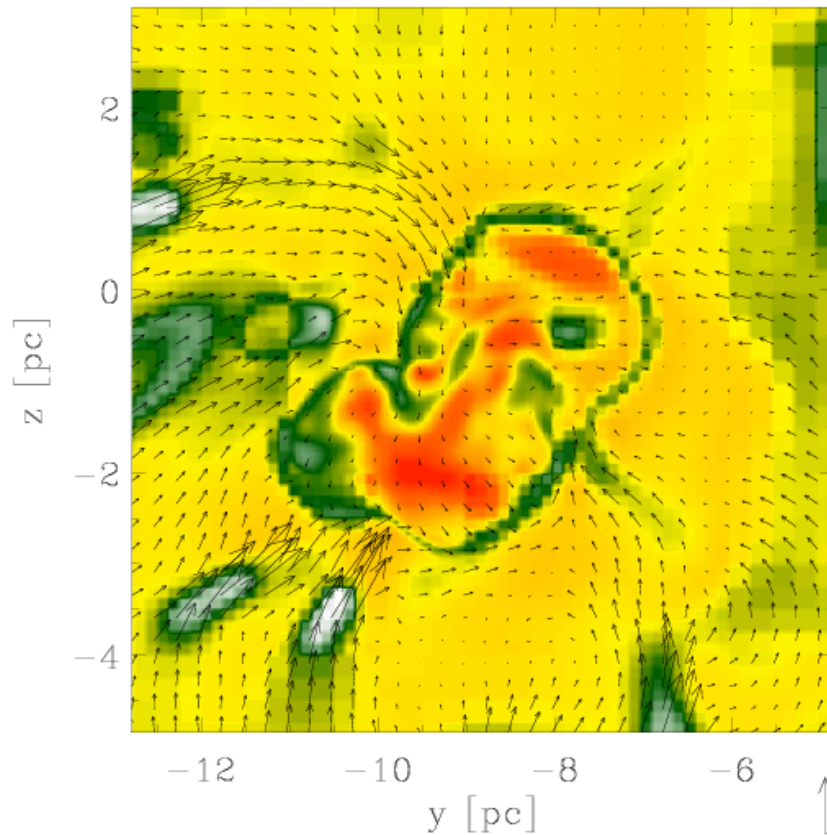
(2) *core mergers* → super-Jeans → gravitational collapse & star formation

example: *Pipe nebula* ???

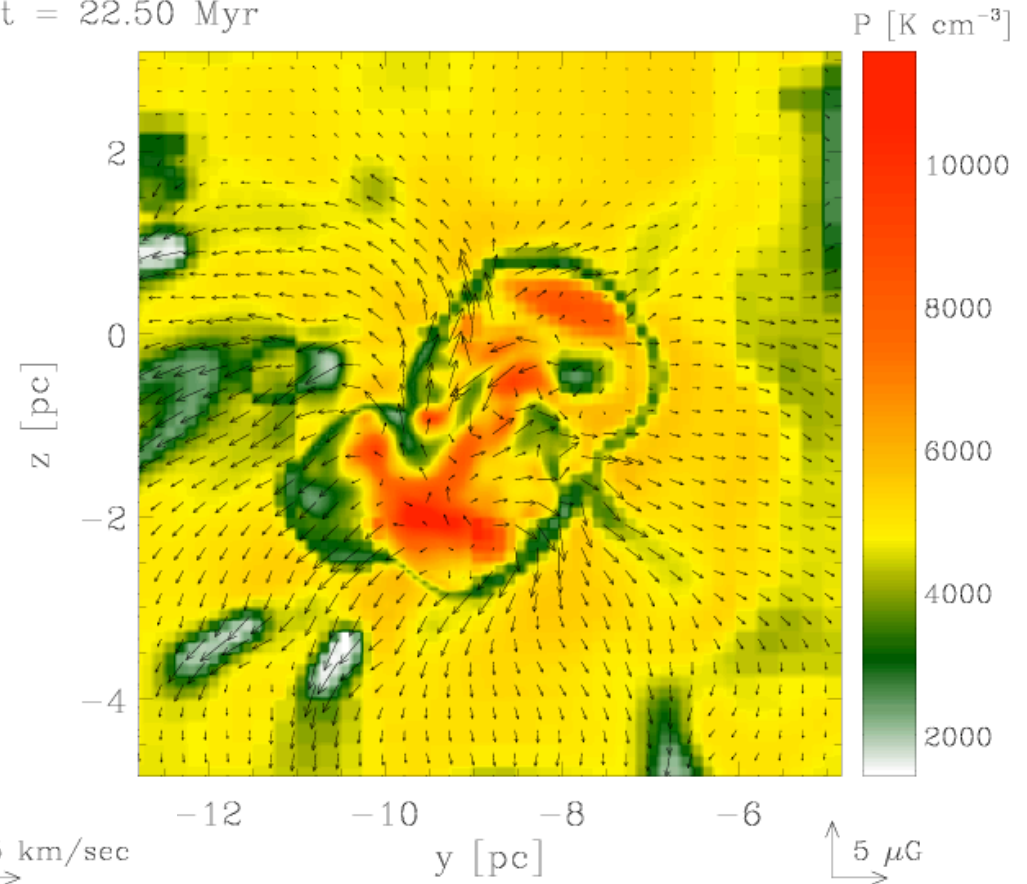
from Banerjee et al. (2008)



$t = 22.50 \text{ Myr}$



$t = 22.50 \text{ Myr}$

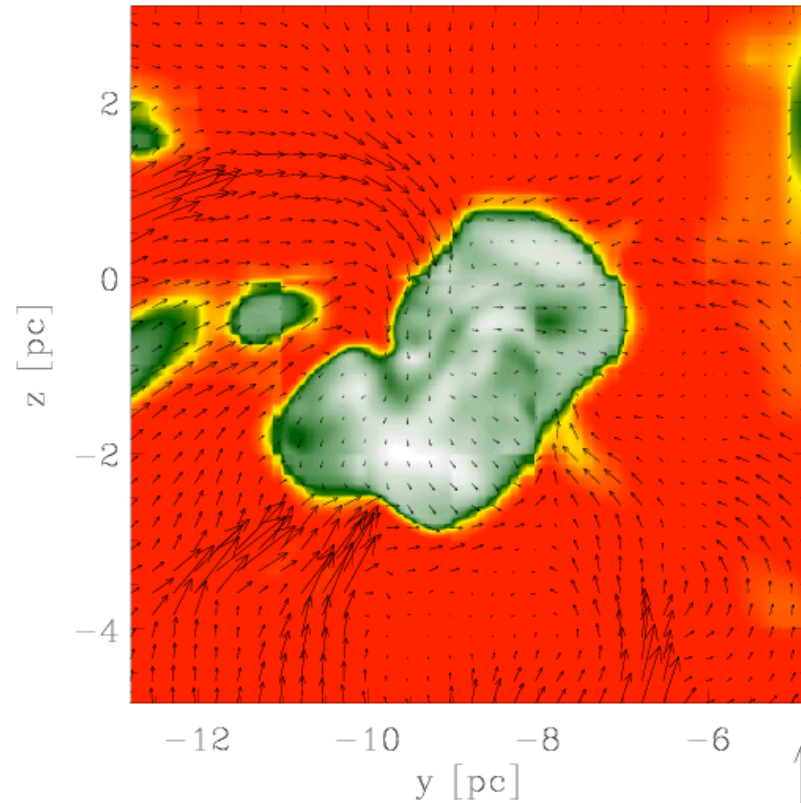


- cores roughly in pressure balance with surroundings
- relation between flow and magnetic field:
mass flow mostly along field lines

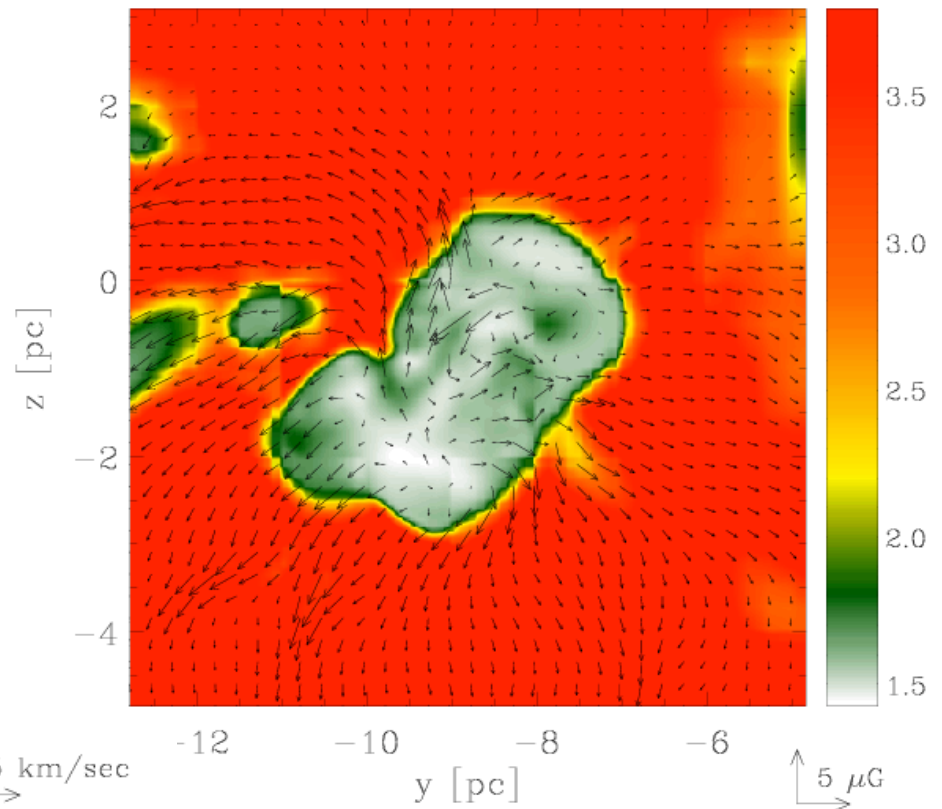
from Banerjee et al. (2008)



$t = 22.50 \text{ Myr}$



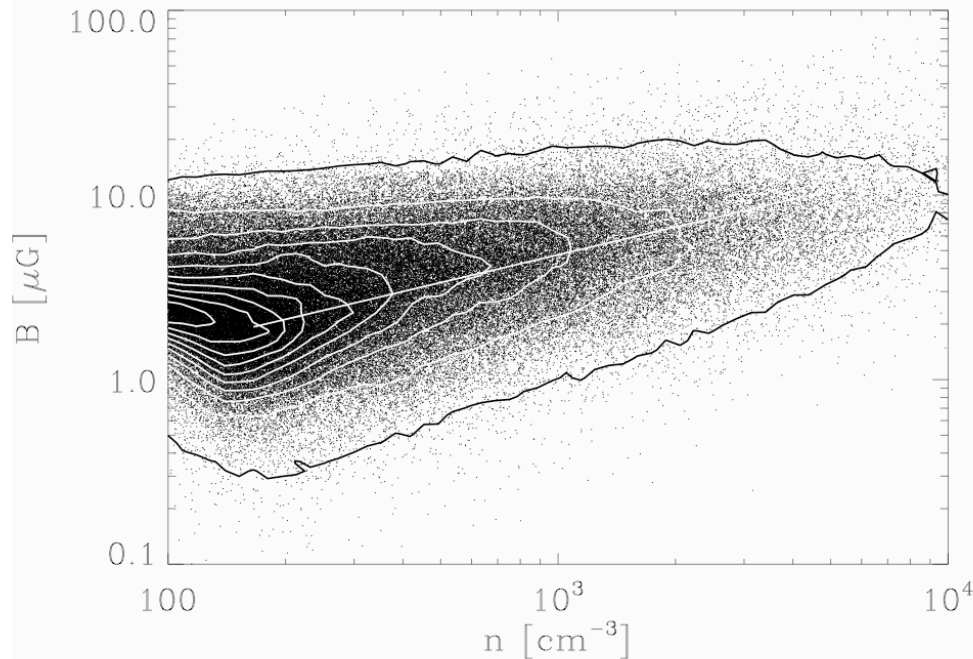
$t = 22.50 \text{ Myr}$



- typical core densities $n \sim 2 - 5 \times 10^3 \text{ cm}^{-3}$
- typical core temperatures $T \sim 30 - 50 \text{ K}$

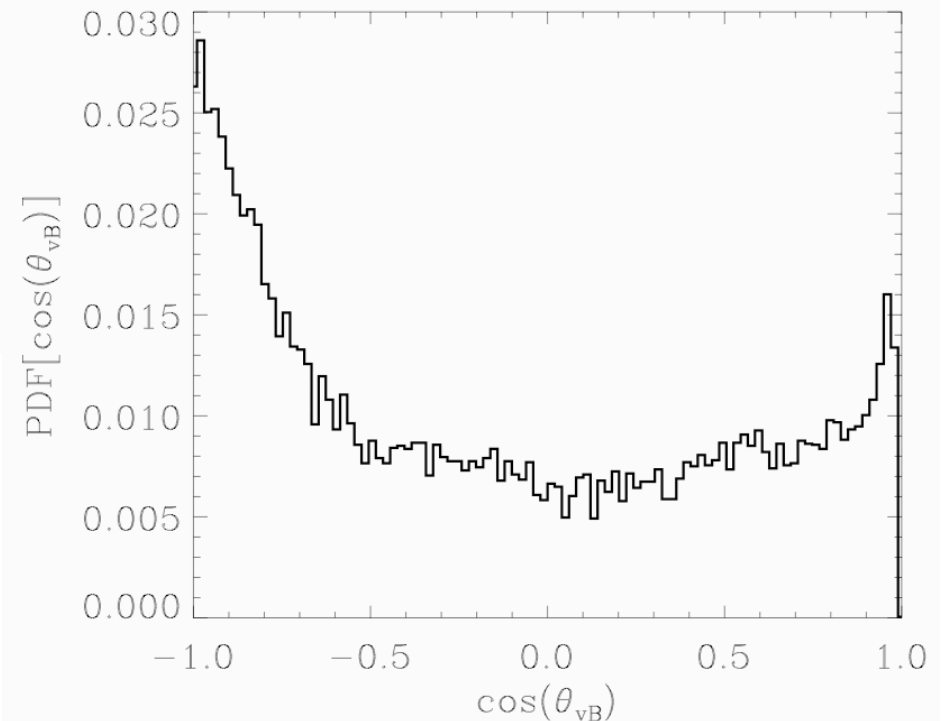
from Banerjee et al. (2008)

some results: statistical correlations



- **large** scatter of magnetic field strengths: sub- and super-critical cores exist
- median slope: $B \propto n^{0.5}$ (e.g. *Crutcher 1999*)

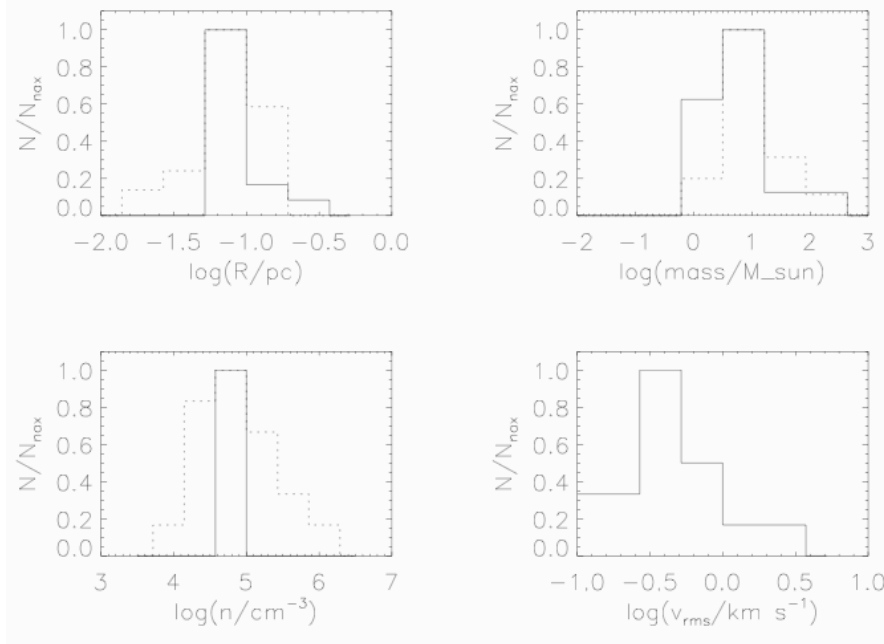
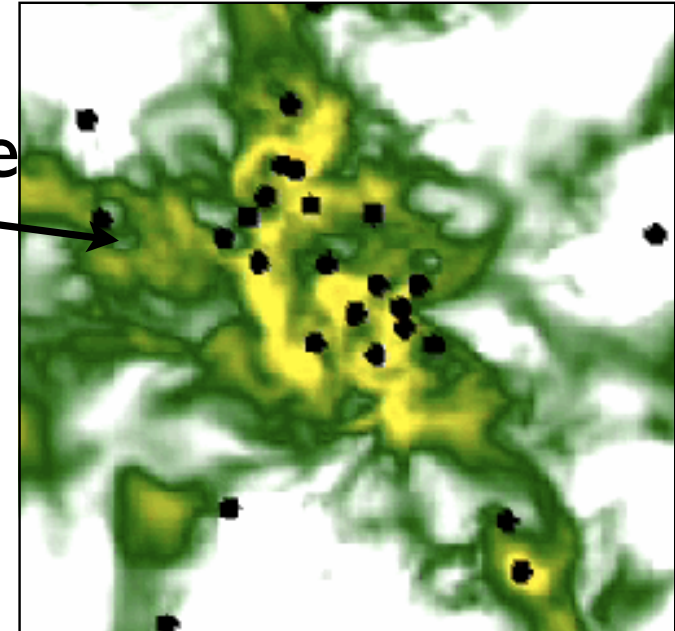
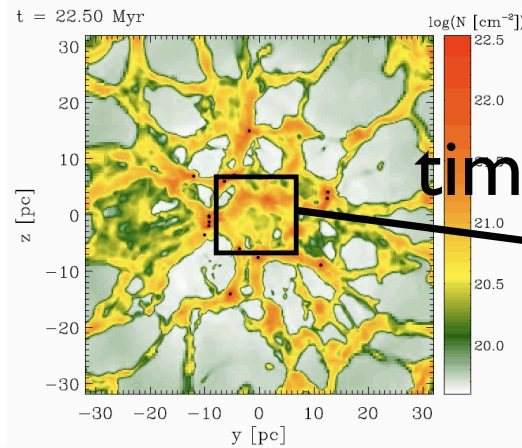
- **strong** correlation of gas streams and magnetic field lines



some results: loci of high-mass stars

global contraction phase

center of the cloud
→ birthplace for
massive stars?
(eg. Zinnecker & Yorke 2007)



comparison of core properties with
observation of Cygnus X by *Motte et al*
2007



initial mass
function



initial mass function

- what is the relation between molecular cloud fragmentation and the distribution of stars?
- important quantity: *IMF*
- equally important **CAVEAT**:
 - “everyone” gets the right *IMF*
 - better look for secondary indicators
 - *stellar multiplicity*
 - protostellar *spin* (including disk)
 - *spatial distribution + kinematics* in young clusters
 - *magnetic field strength and orientation*



IMF

- distribution of stellar masses depends on
 - turbulent initial conditions
 - > mass spectrum of prestellar cloud cores
 - collapse and interaction of prestellar cores
 - > competitive accretion and N -body effects
 - thermodynamic properties of gas
 - > balance between heating and cooling
 - > EOS (determines which cores go into collapse)
 - (proto) stellar feedback terminates star formation
 - ionizing radiation, bipolar outflows, winds, SN



IMF

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example: model of Orion cloud

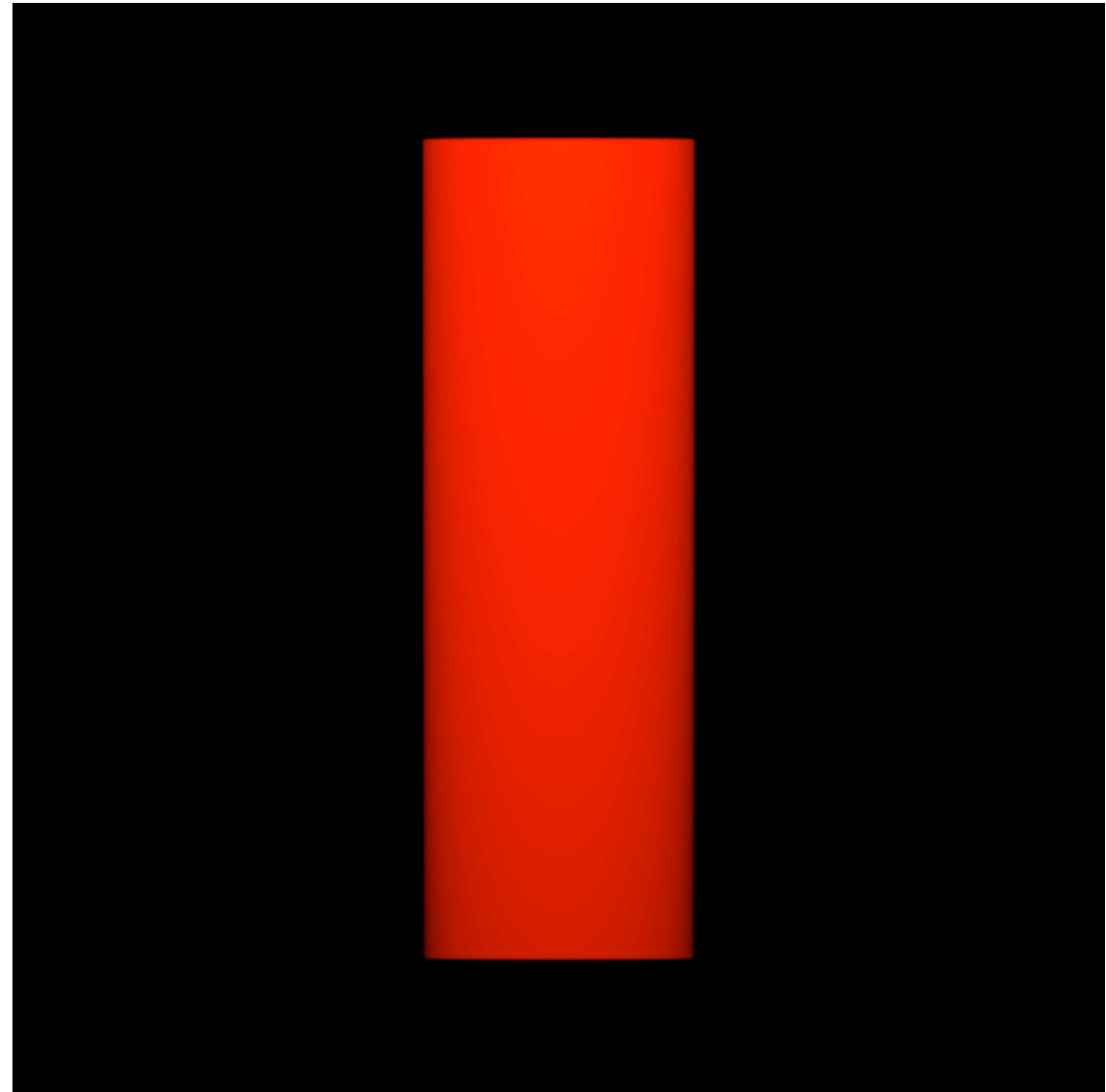
„model“ of Orion cloud:
15.000.000 SPH particles,
 $10^4 M_{\text{sun}}$ in 10 pc, mass
resolution $0,02 M_{\text{sun}}$, forms
 ~ 2.500 „stars“ (sink particles)

isothermal EOS, top bound,
bottom unbound

has clustered as well as
distributed „star“ formation

efficiency varies from 1% to
20%

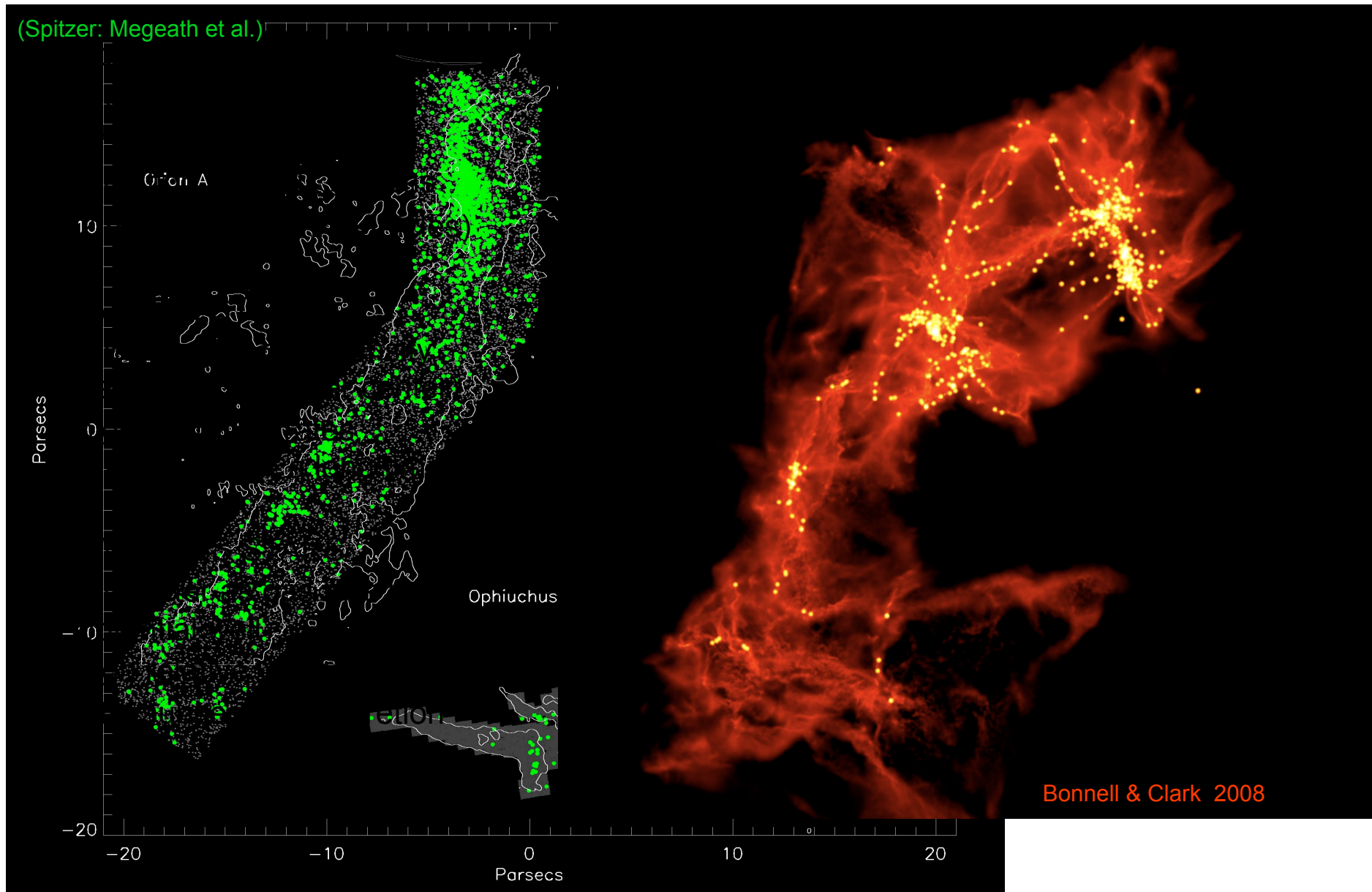
develops full IMF
(distribution of sink particle masses)



(Bonnell & Clark 2008)



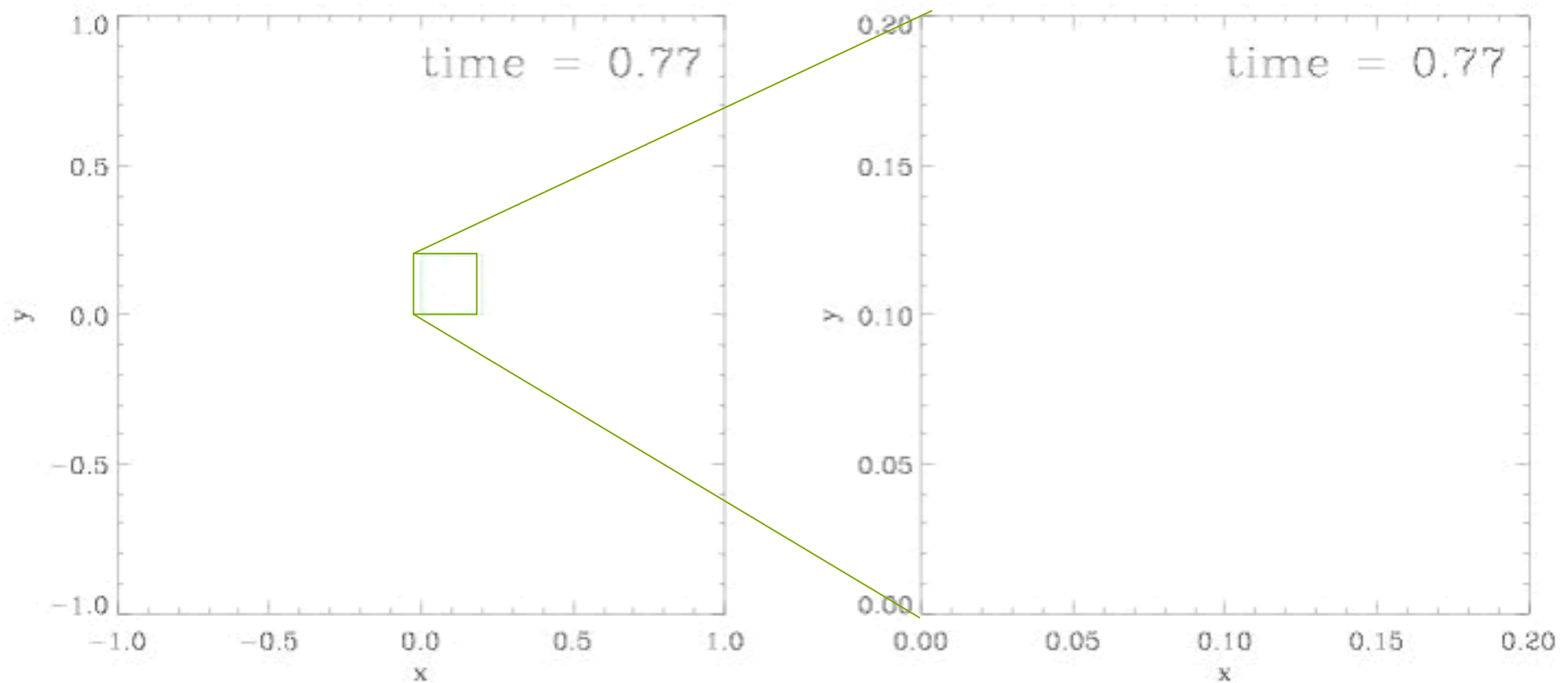
example: model of Orion cloud





dynamics of nascent star cluster

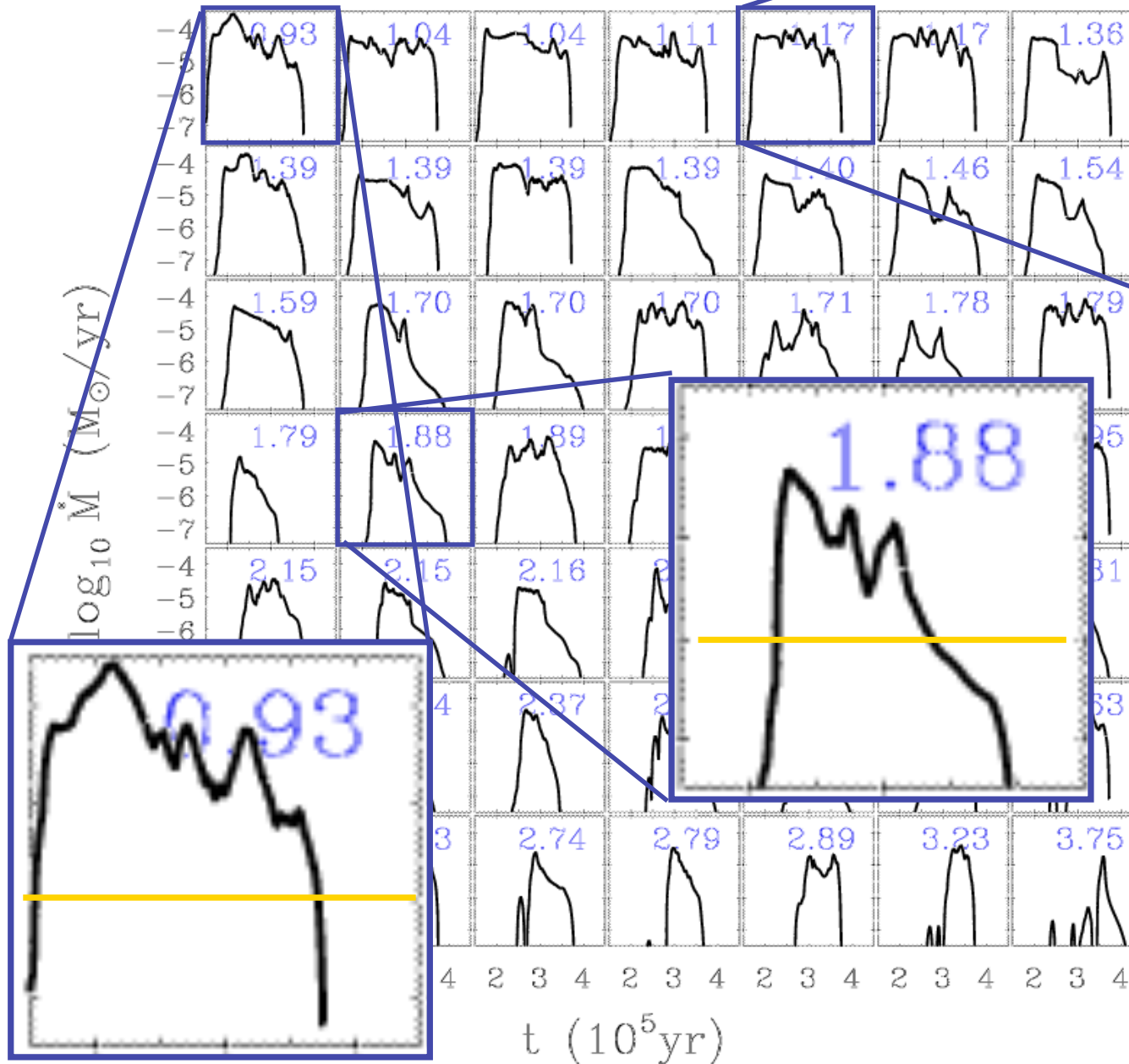
in dense clusters protostellar interaction may become important!



Trajectories of protostars in a nascent dense cluster created by gravoturbulent fragmentation
(from Klessen & Burkert 2000, *ApJS*, 128, 287)



accretion rates in clust



Mass accretion rates *vary with time* and are strongly *influenced* by the *cluster environment*.

(Klessen 2001, ApJ, 550, L77;
also Schmeja & Klessen,
2004, A&A, 419, 405)



IMF

- distribution of stellar masses depends on
 - turbulent initial conditions
 - > mass spectrum of prestellar cloud cores
 - collapse and interaction of prestellar cores
 - > competitive accretion and N -body effects
 - *thermodynamic properties of gas*
 - > *balance between heating and cooling*
 - > *EOS (determines which cores go into collapse)*
 - (proto) stellar feedback terminates star formation
 - ionizing radiation, bipolar outflows, winds, SN

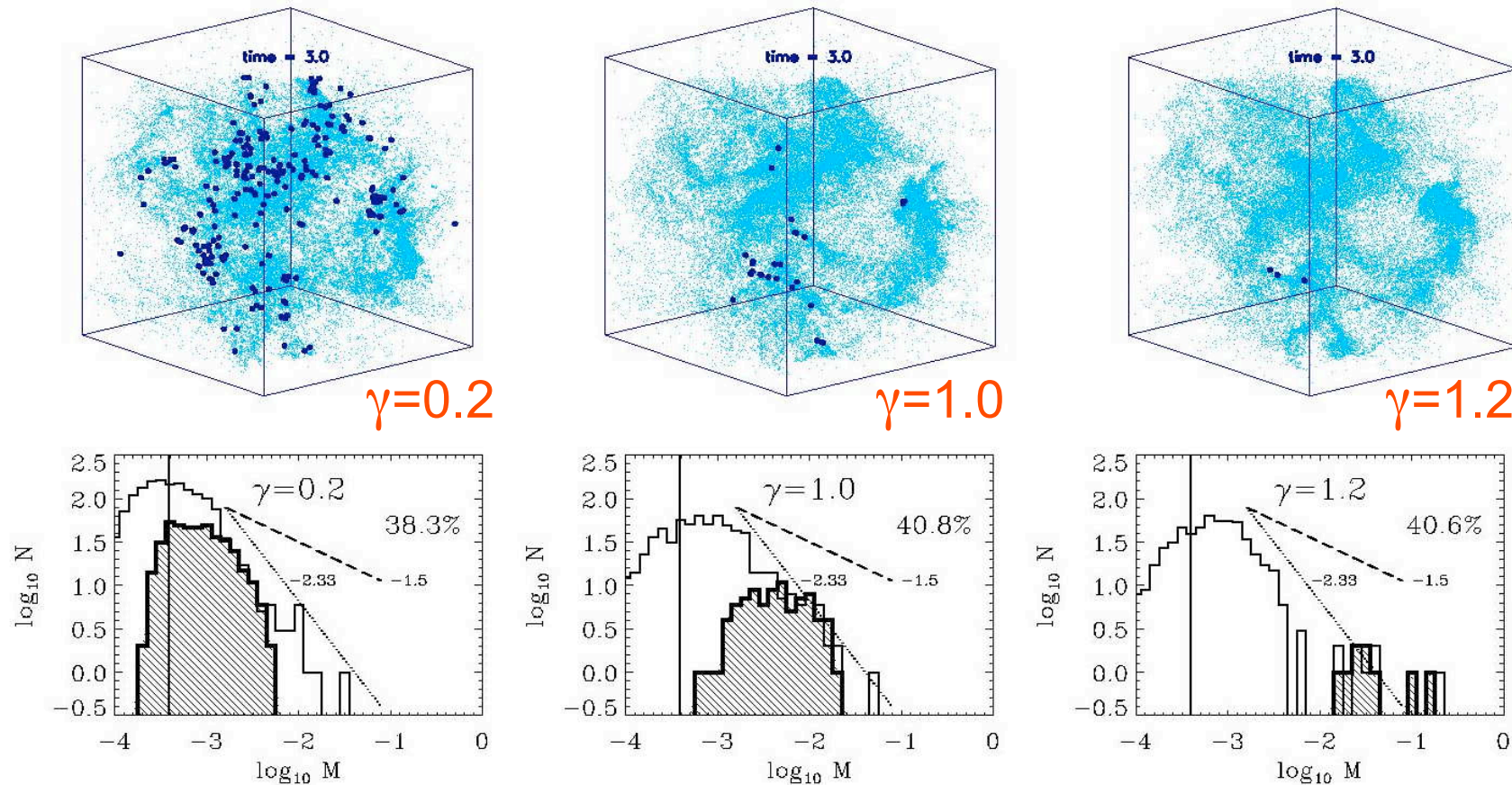


dependency on EOS

- degree of fragmentation depends on *EOS*!
- polytropic EOS: $p \propto \rho^\gamma$
- $\gamma < 1$: dense cluster of low-mass stars
- $\gamma > 1$: isolated high-mass stars
- (see Li, Klessen, & Mac Low 2003, ApJ, 592, 975; also Kawachi & Hanawa 1998, Larson 2003)



dependency on EOS



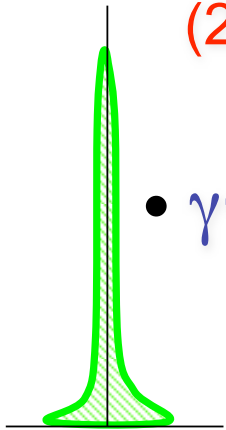
for $\gamma < 1$ fragmentation is enhanced \rightarrow *cluster of low-mass stars*
for $\gamma > 1$ it is suppressed \rightarrow formation of *isolated massive stars*



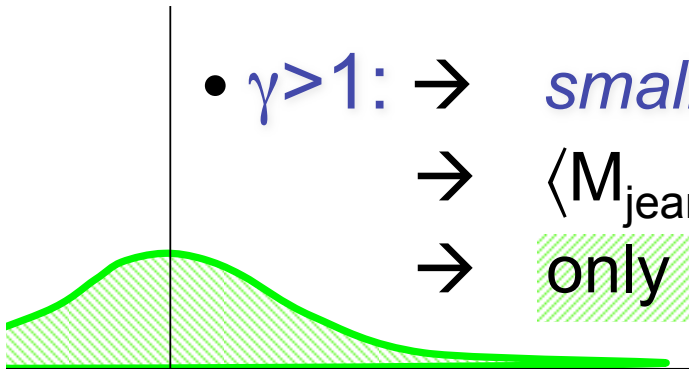
how does that work?

$$(1) \quad p \propto \rho^\gamma \quad \rightarrow \quad \rho \propto p^{1/\gamma}$$

$$(2) \quad M_{\text{jeans}} \propto \gamma^{3/2} \rho^{(3\gamma-4)/2}$$



- $\gamma < 1$: \rightarrow *large* density excursion for given pressure
 \rightarrow $\langle M_{\text{jeans}} \rangle$ becomes small
 \rightarrow number of fluctuations with $M > M_{\text{jeans}}$ is large



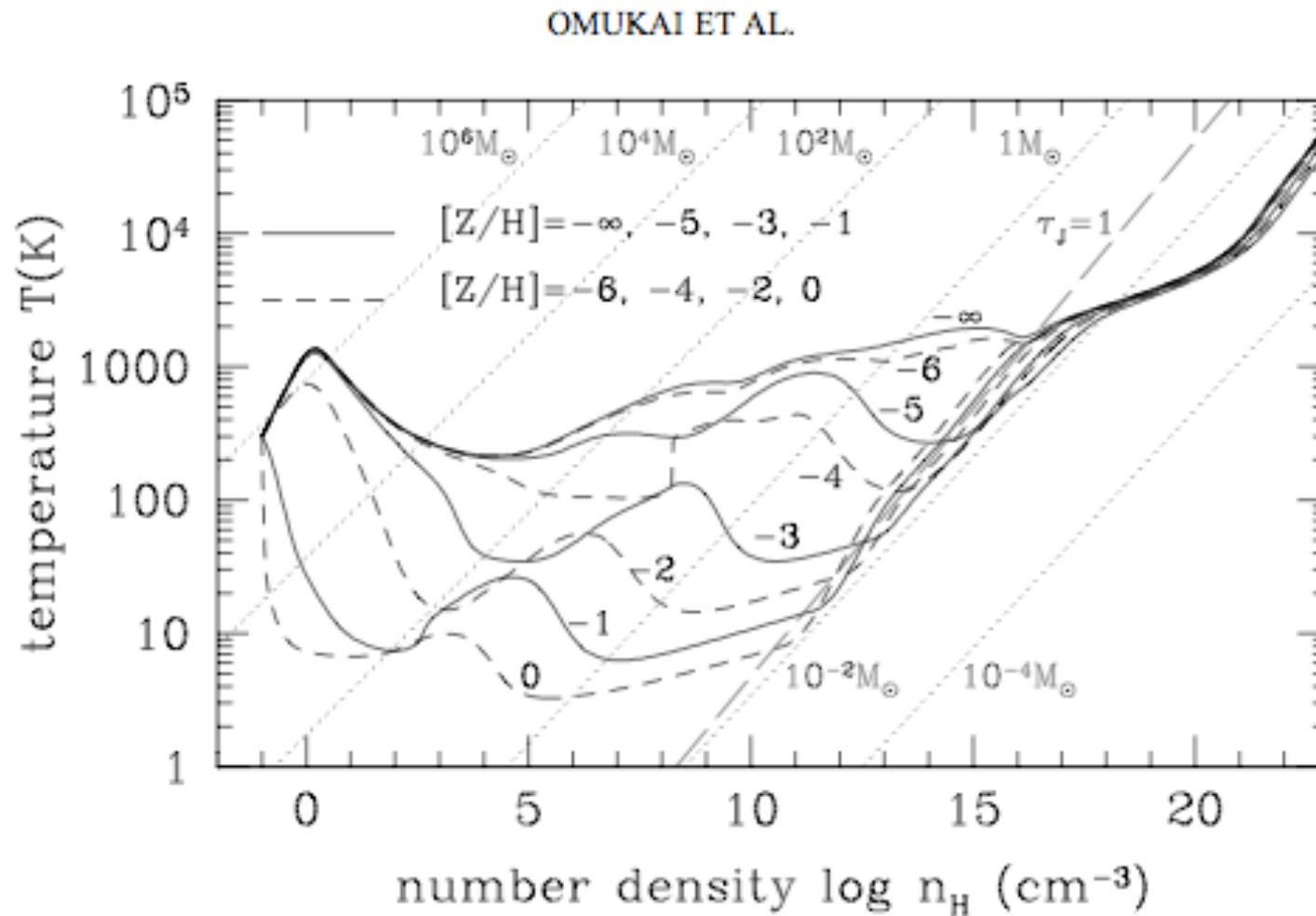
- $\gamma > 1$: \rightarrow *small* density excursion for given pressure
 \rightarrow $\langle M_{\text{jeans}} \rangle$ is large
 \rightarrow only few and massive clumps exceed M_{jeans}



EOS in different environments



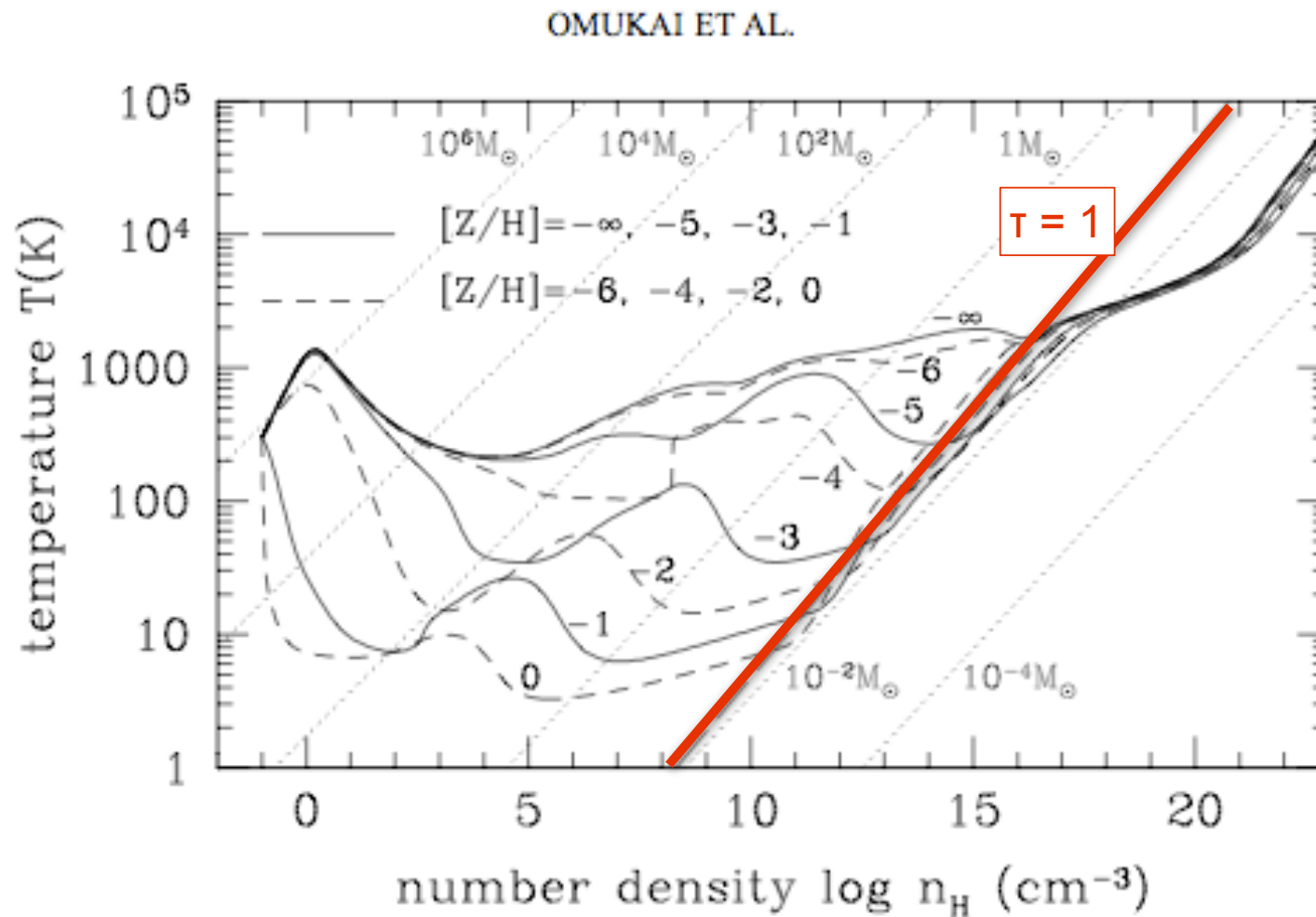
EOS as function of metallicity



(Omukai et al. 2005)



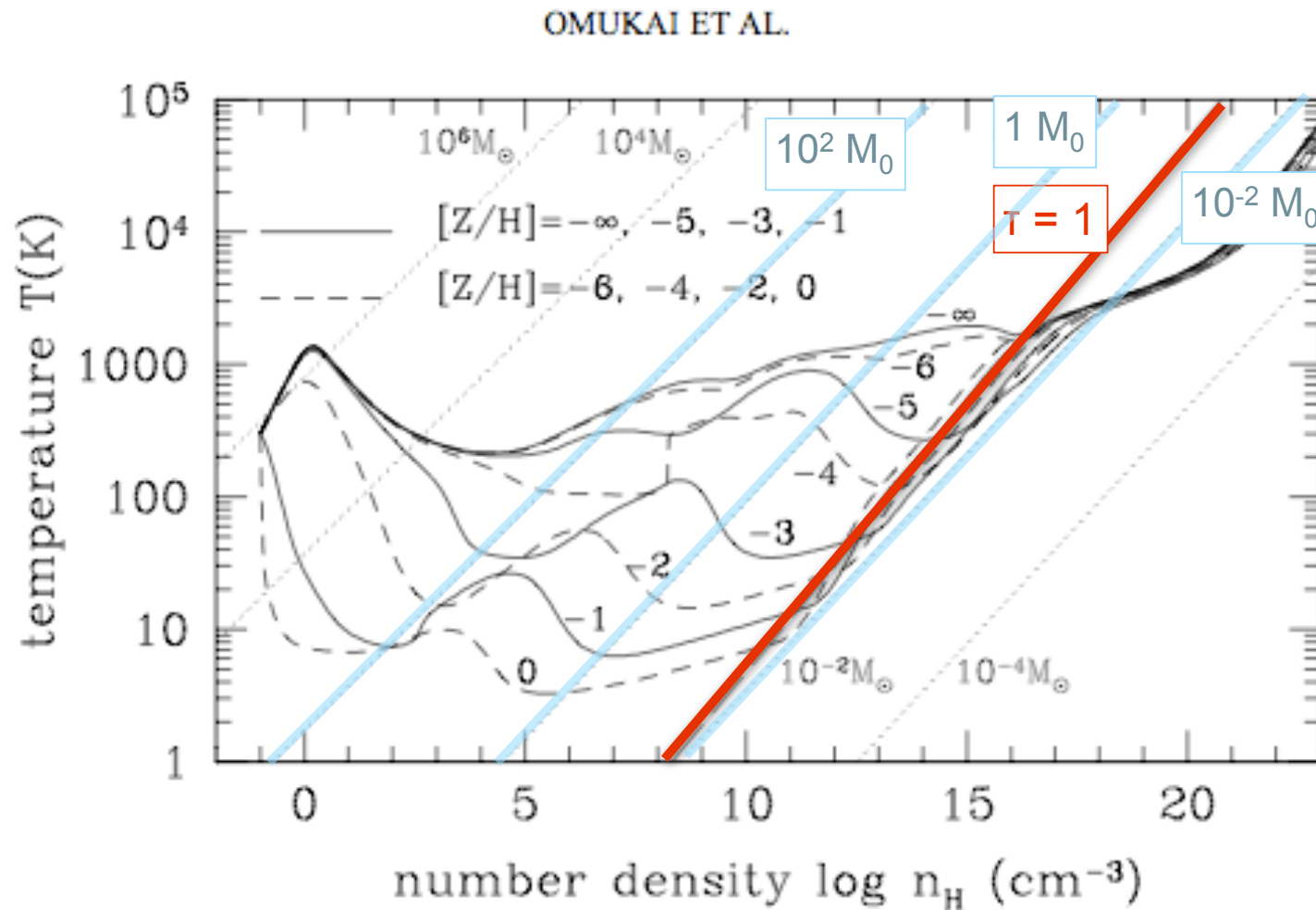
EOS as function of metallicity



(Omukai et al. 2005)



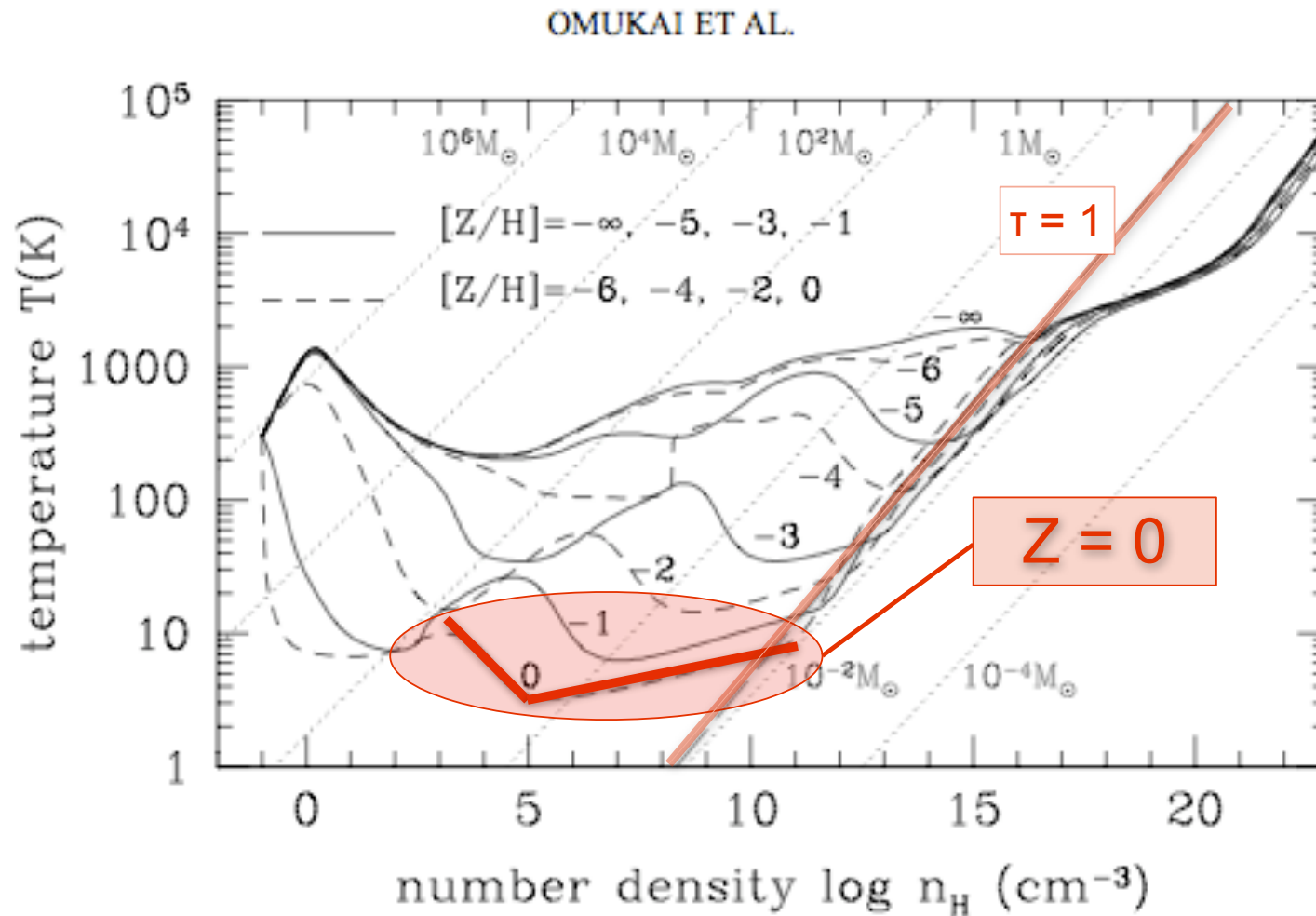
EOS as function of metallicity



(Omukai et al. 2005)



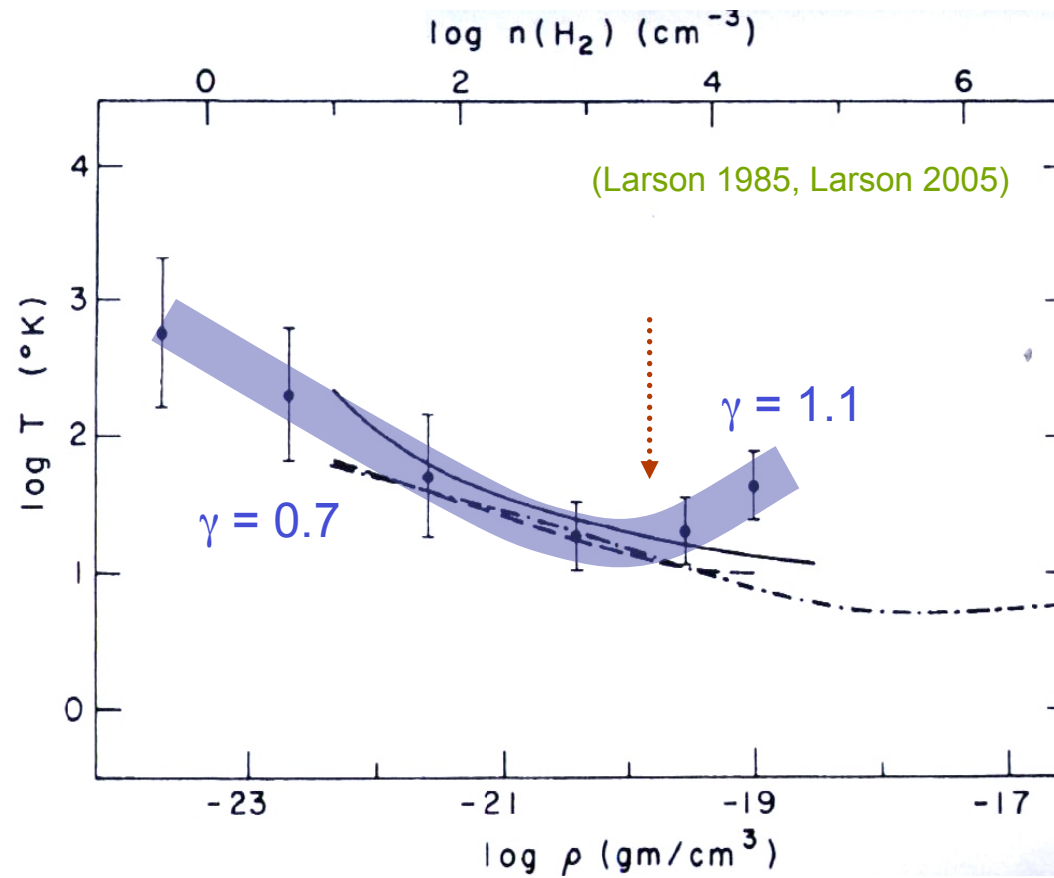
present-day star formation



(Omukai et al. 2005, Jappsen et al. 2005, Larson 2005)



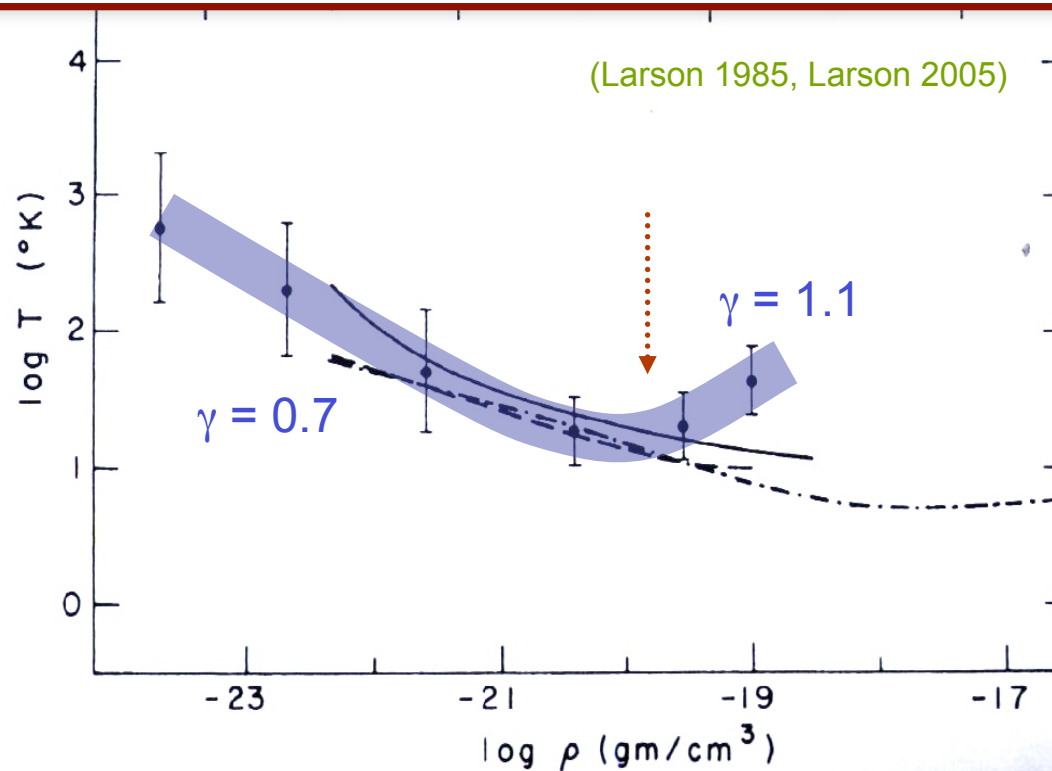
present-day star formation





present-day star formation

This kink in EOS is very insensitive to environmental conditions such as ambient radiation field
--> reason for universal form of the IMF? (Elmegreen et al. 2008)





IMF from simple piece-wise polytropic EOS

$$\gamma_1 = 0.7$$

$$\gamma_2 = 1.1$$

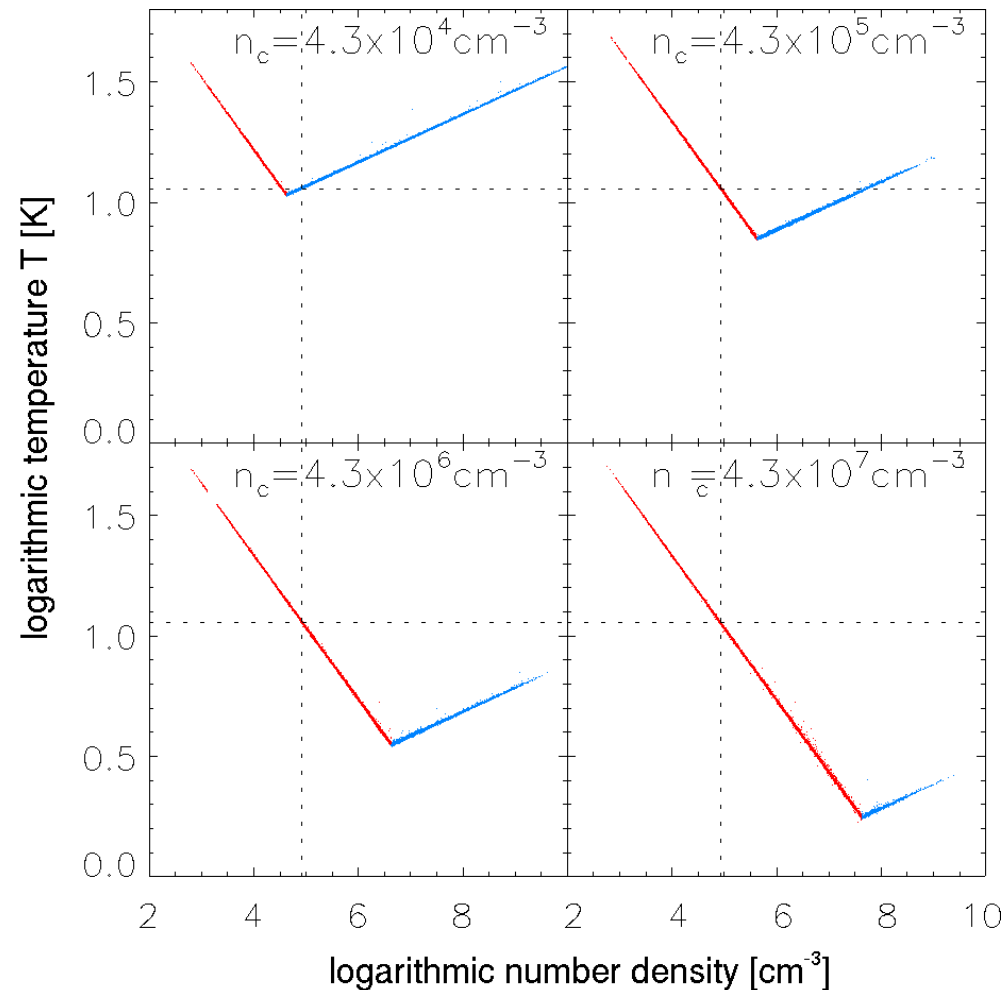
$$T \sim \rho^{\gamma-1}$$

EOS and Jeans Mass:

$$p \propto \rho^\gamma \rightarrow \rho \propto p^{1/\gamma}$$

$$M_{\text{jeans}} \propto \gamma^{3/2} \rho^{(3\gamma-4)/2}$$

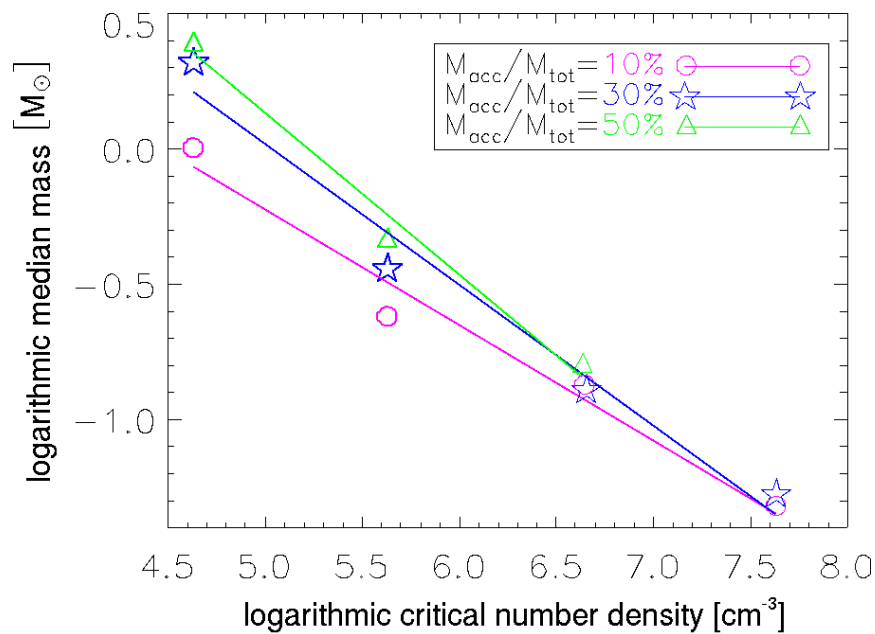
(Jappsen et al. 2005)



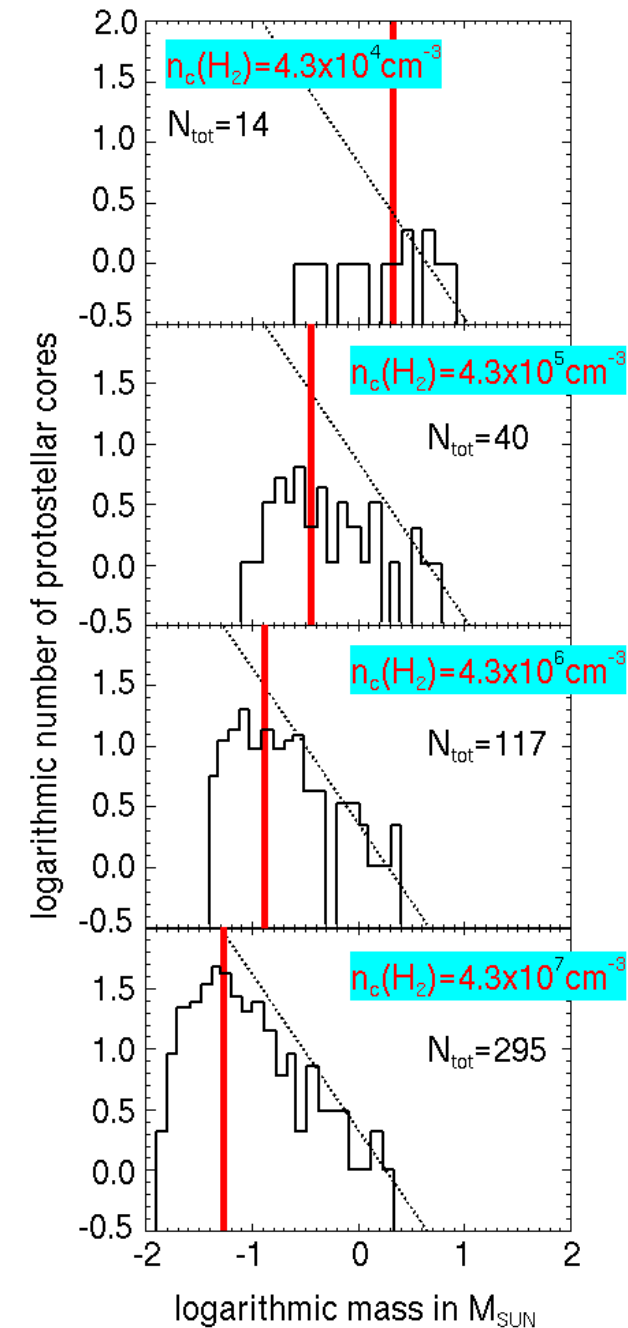


IMF from simple piece-wise EOS

critical density \uparrow \longrightarrow median mass \downarrow

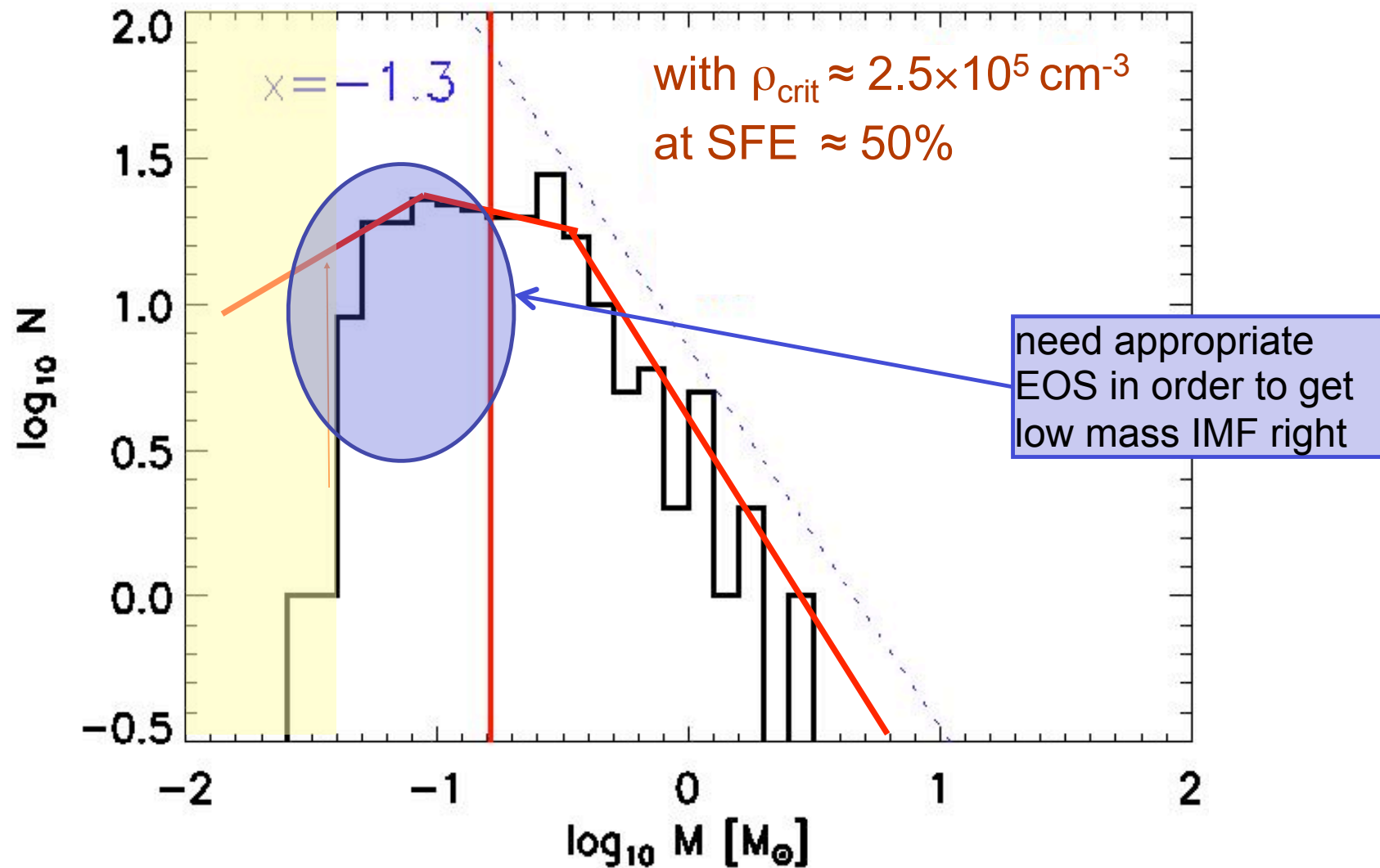


(Jappsen et al. 2005)



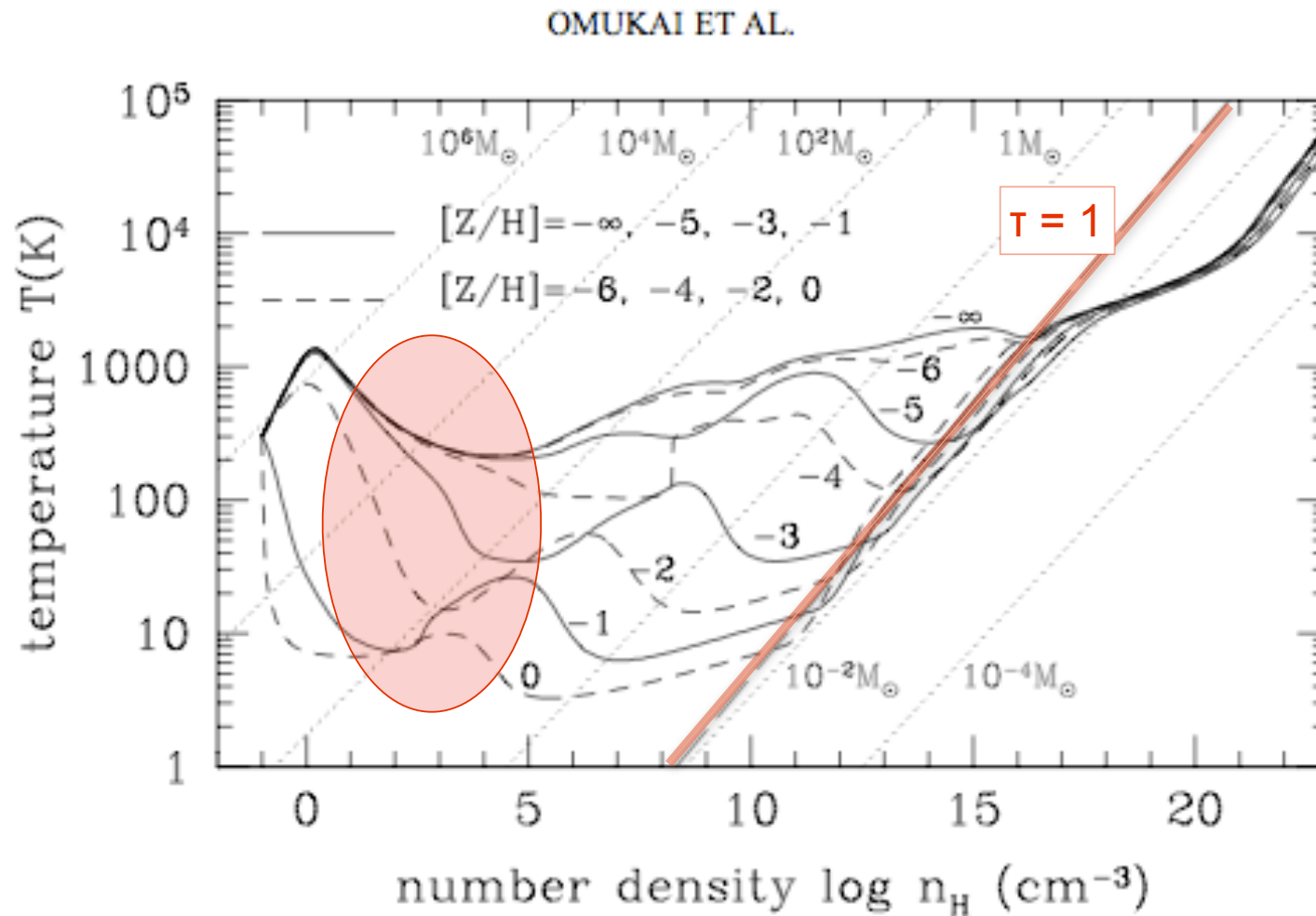


IMF in nearby molecular clouds





dependence on Z at low density



(Omukai et al. 2005)



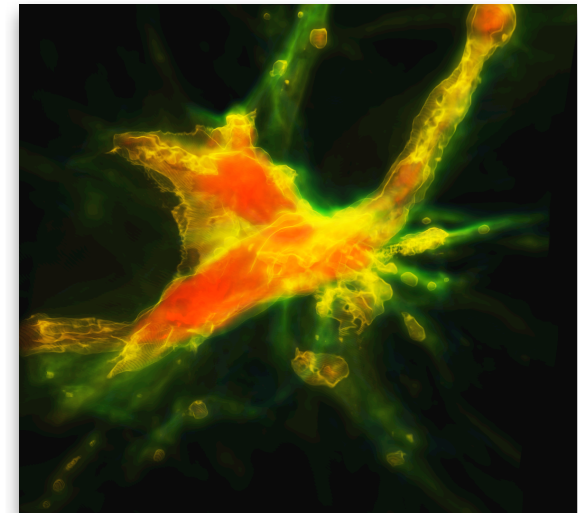
dependence on Z at low density

- at densities $n < 10^2 \text{ cm}^{-3}$ and metallicities $Z < 10^{-2}$ H_2 cooling dominates behavior.
(Jappsen et al. 2007)
- fragmentation depends on *initial conditions*
 - example 1: *solid-body rotating top-hat* initial conditions with dark matter fluctuations (a la Bromm et al. 1999) fragment no matter what metallicity you take (in regime $n \leq 10^6 \text{ cm}^{-3}$) \rightarrow because *unstable disk* builds up
(Jappsen et al. 2009a)
 - example 2: *centrally concentrated halo* does *not* fragment up to densities of $n \approx 10^6 \text{ cm}^{-3}$ up to metallicities $Z \approx -1$ (Jappsen et al. 2009b)



implications for Pop III

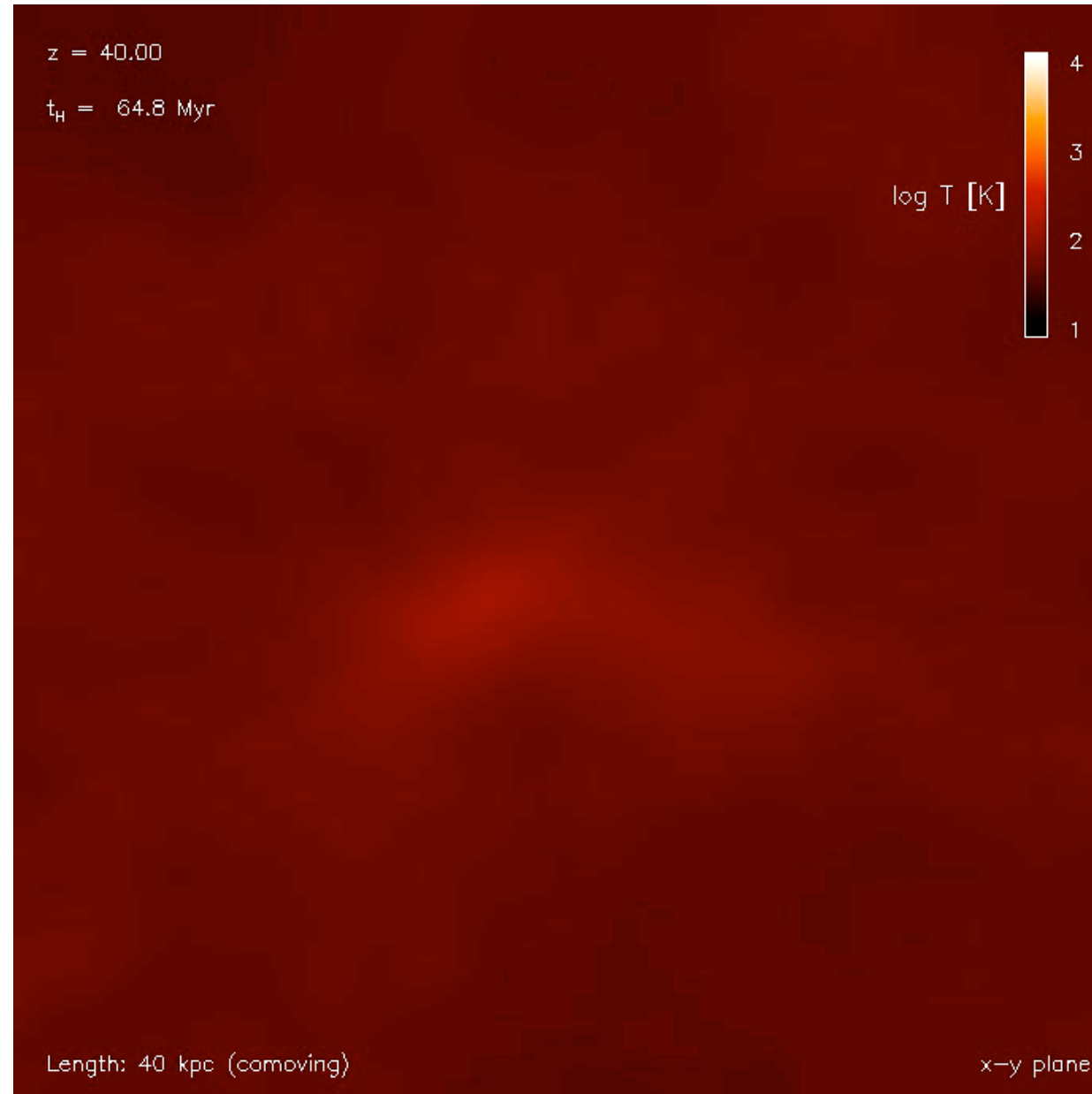
- star formation will depend on *degree of turbulence* in protogalactic halo
- speculation: *differences in stellar mass function?*
- speculation:
 - low-mass halos → low level of turbulence → relatively massive stars
 - high-mass halos (atomic cooling halos) → high degree of turbulence → wider mass spectrum with peak at lower-masses?



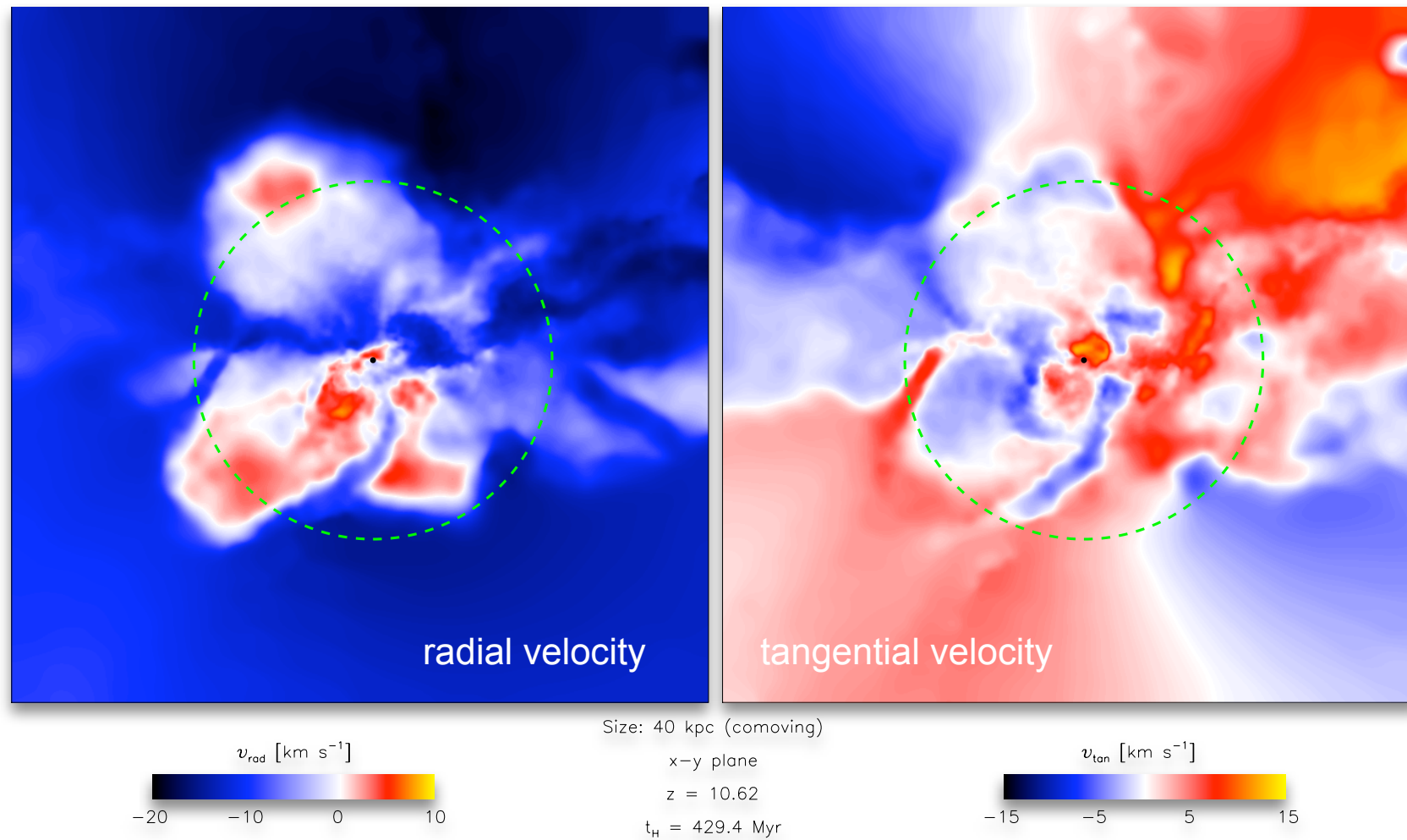
(Greif et al. 2008)



turbulence developing in an atomic cooling halo



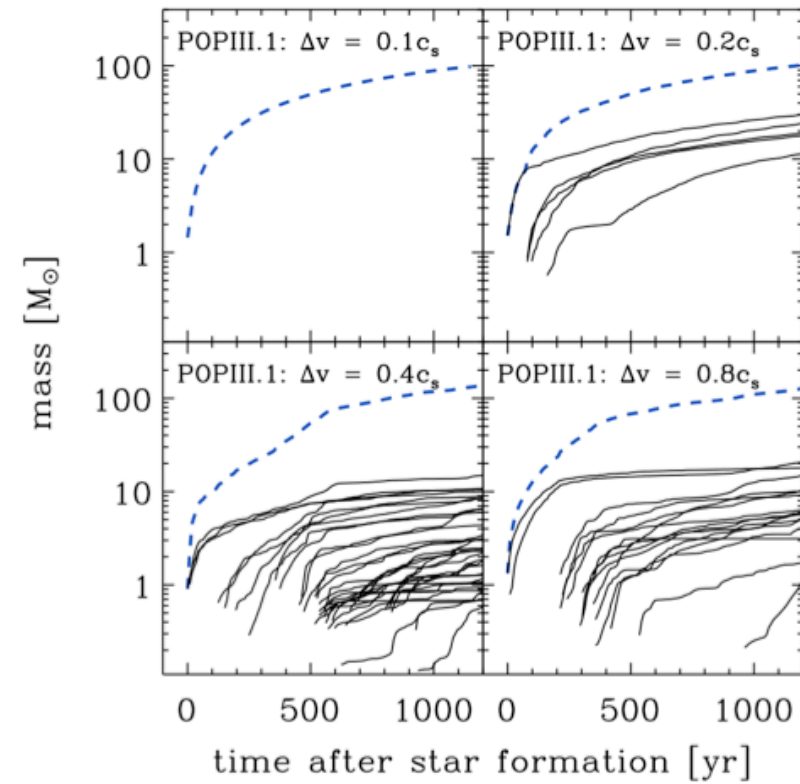
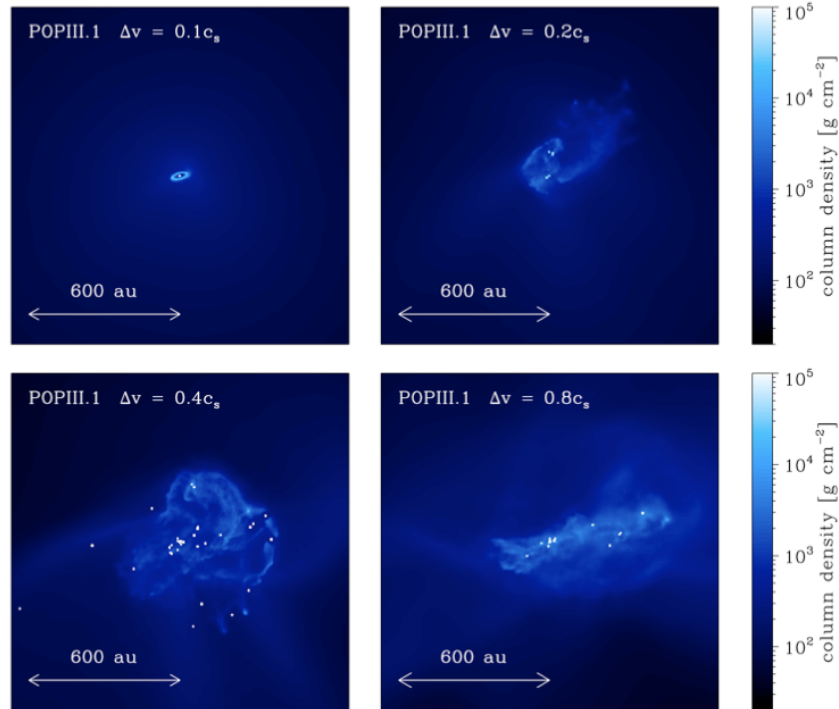
(Greif et al. 2008, see also Wise & Abel 2007)



turbulence developing in an atomic cooling halo (Greif et al. 2008)



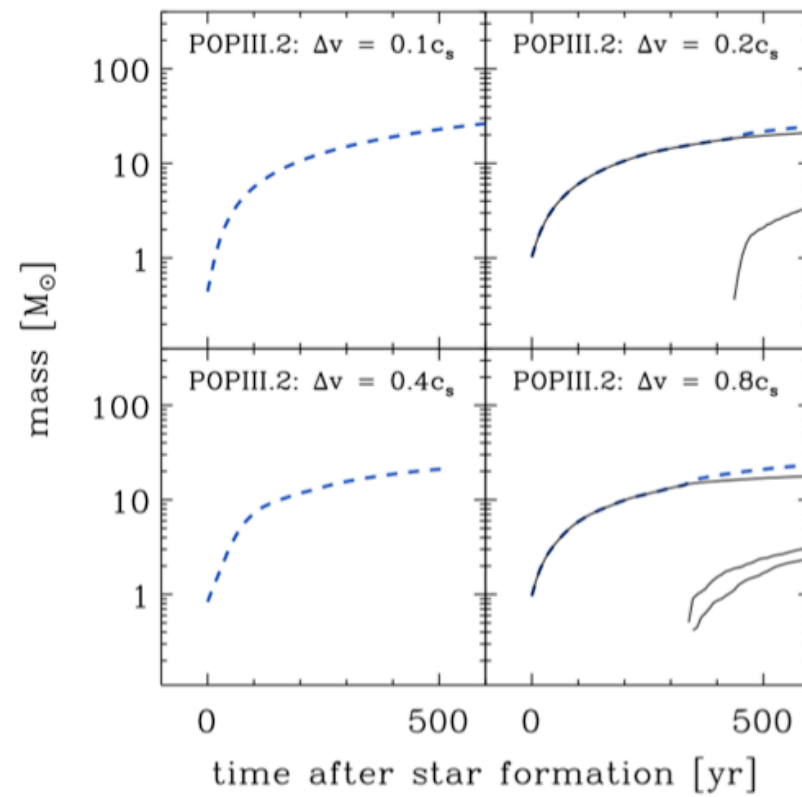
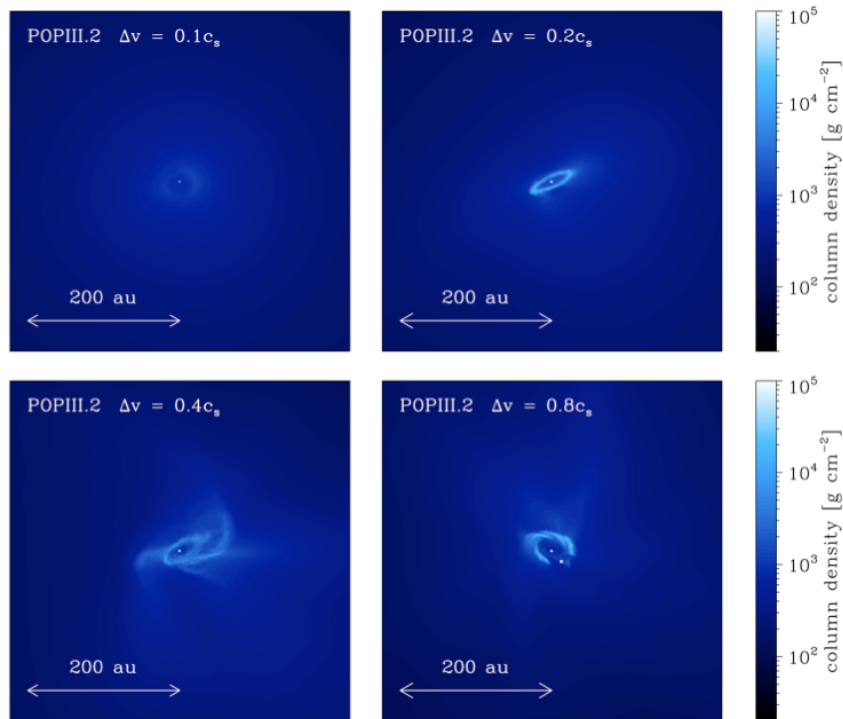
Pop III.1



(Clark et al, in prep.)



Pop III.2



(Clark et al, in prep.)



once again: thermodynamics

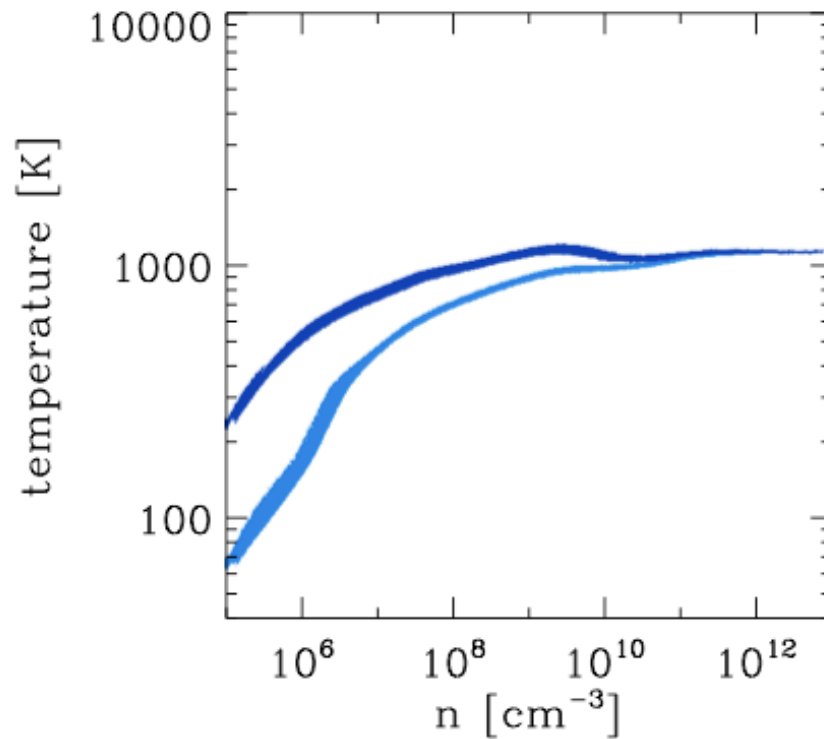


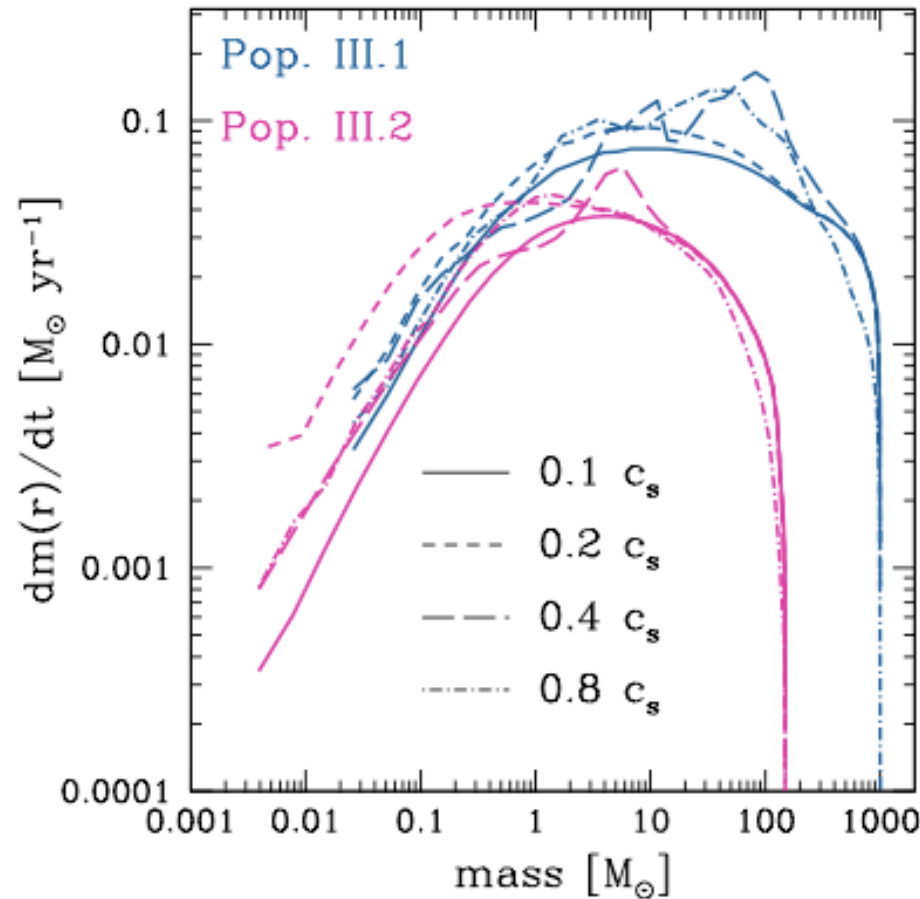
FIG. 6.— Temperature as a function of number density for the Pop. III.1 (dark blue) and Pop. III.2 (light blue) $\Delta v_{\text{turb}} = 0.1 c_s$ simulations. In both cases, the curves denote the state of the cloud at the point just before the formation of the sink particle.

also Pop III.2 gas heats up
above the CMB

--> weaker fragmentation!



once again: thermodynamics

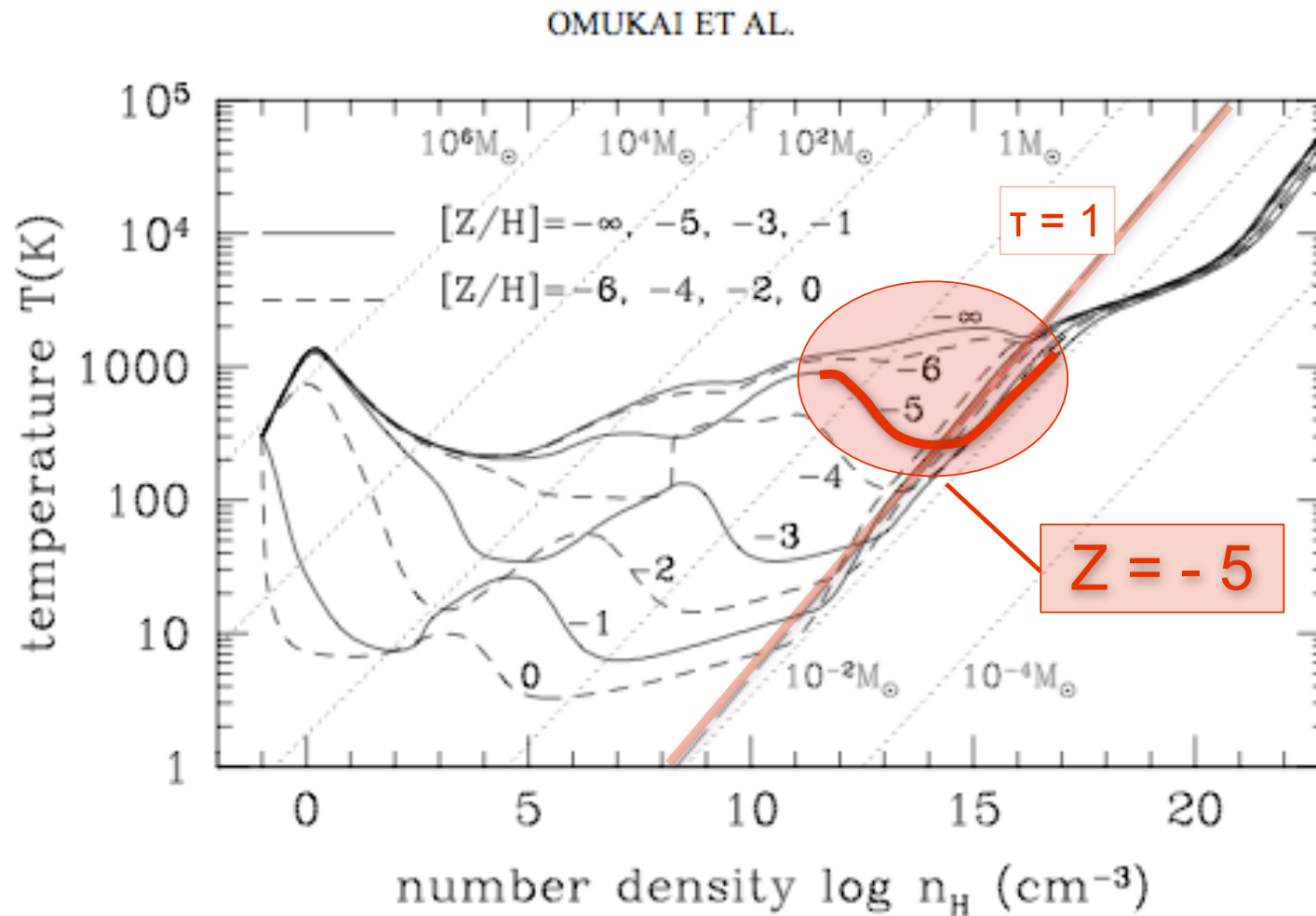


comparison of
accretion rates...

FIG. 8.— Accretion rates as a function of enclosed gas mass in the Pop. III.1 (upper lines; blue) and Pop. III.2 (lower lines; magenta) simulations, estimated as described in Section 4.1. Note that the sharp decline in the accretion rates for enclosed masses close to the initial cloud mass is an artifact of our problem setup; we would not expect to see this in a realistic Pop. III halo.



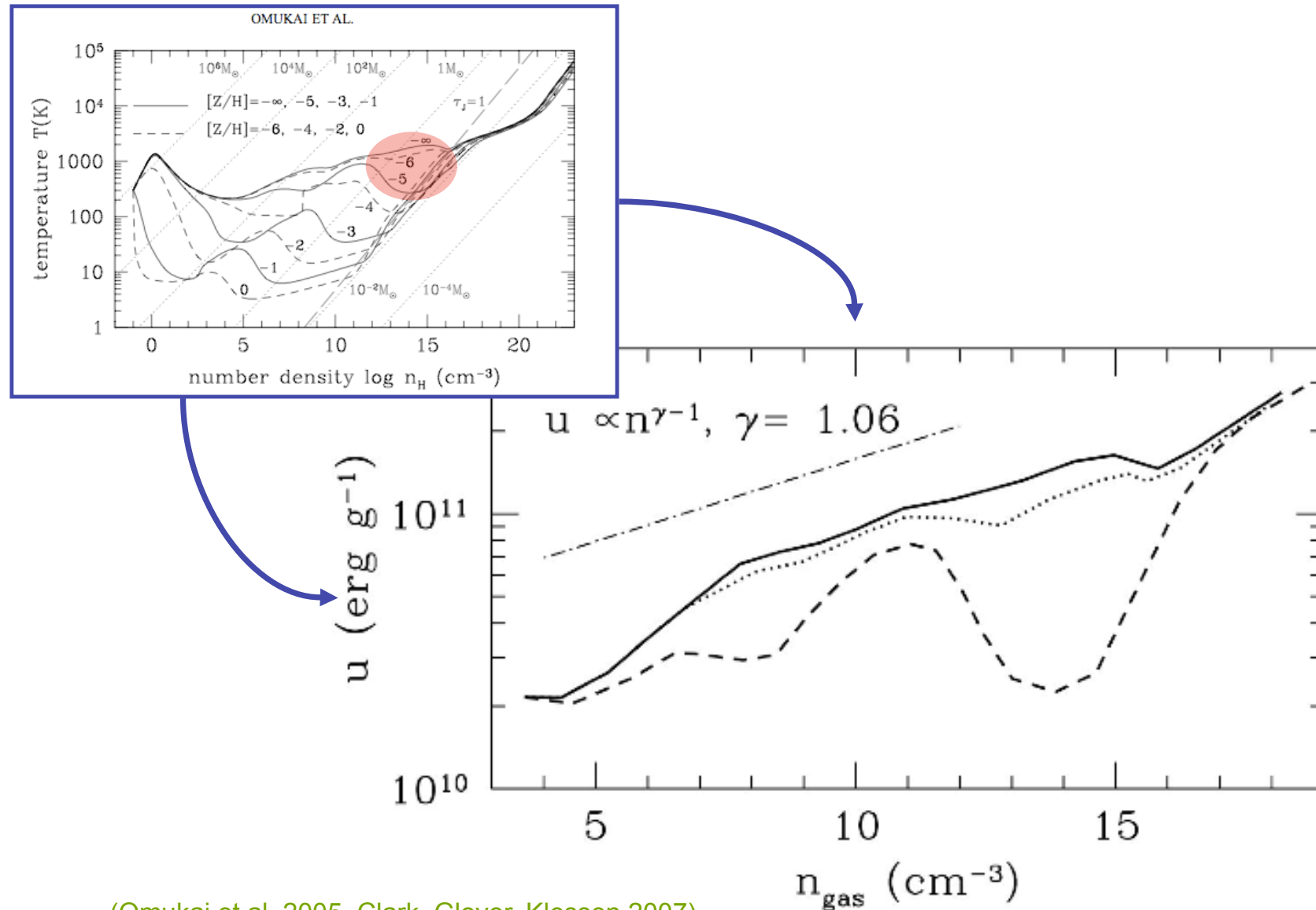
transition: Pop III to Pop II.5



(Omukai et al. 2005)



transition: Pop III to Pop II.5

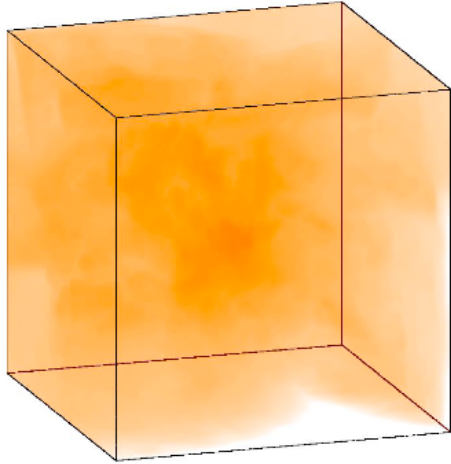


(Omukai et al. 2005, Clark, Glover, Klessen 2007)

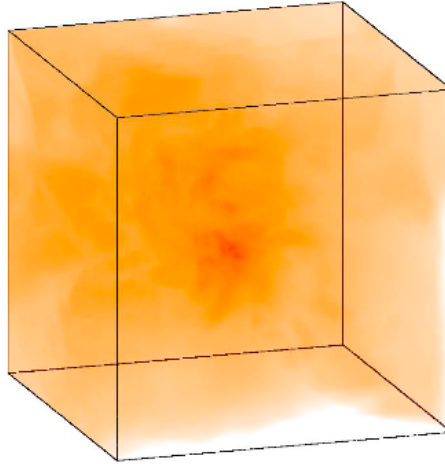


dust induced fragmentation at $Z=10^{-5}$

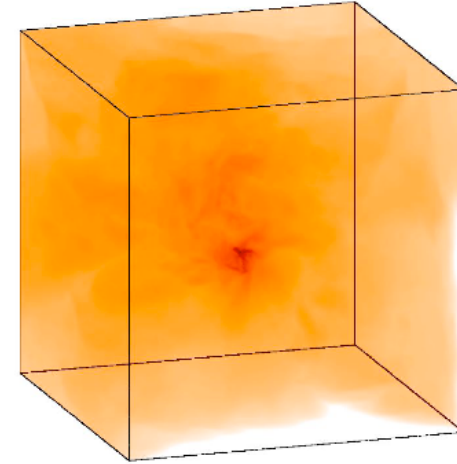
$t = t_{\text{SF}} - 67 \text{ yr}$



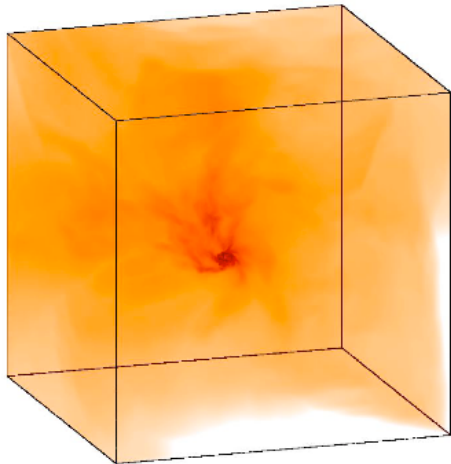
$t = t_{\text{SF}} - 20 \text{ yr}$



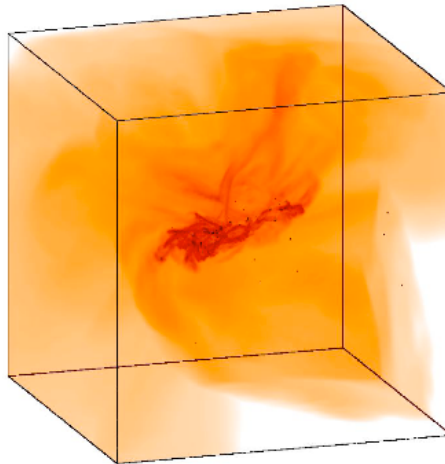
$t = t_{\text{SF}}$



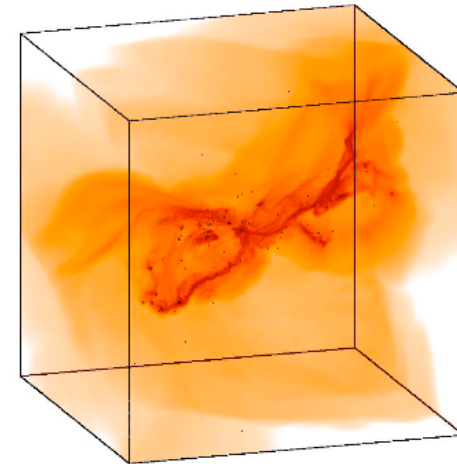
$t = t_{\text{SF}} + 53 \text{ yr}$



$t = t_{\text{SF}} + 233 \text{ yr}$



$t = t_{\text{SF}} + 420 \text{ yr}$

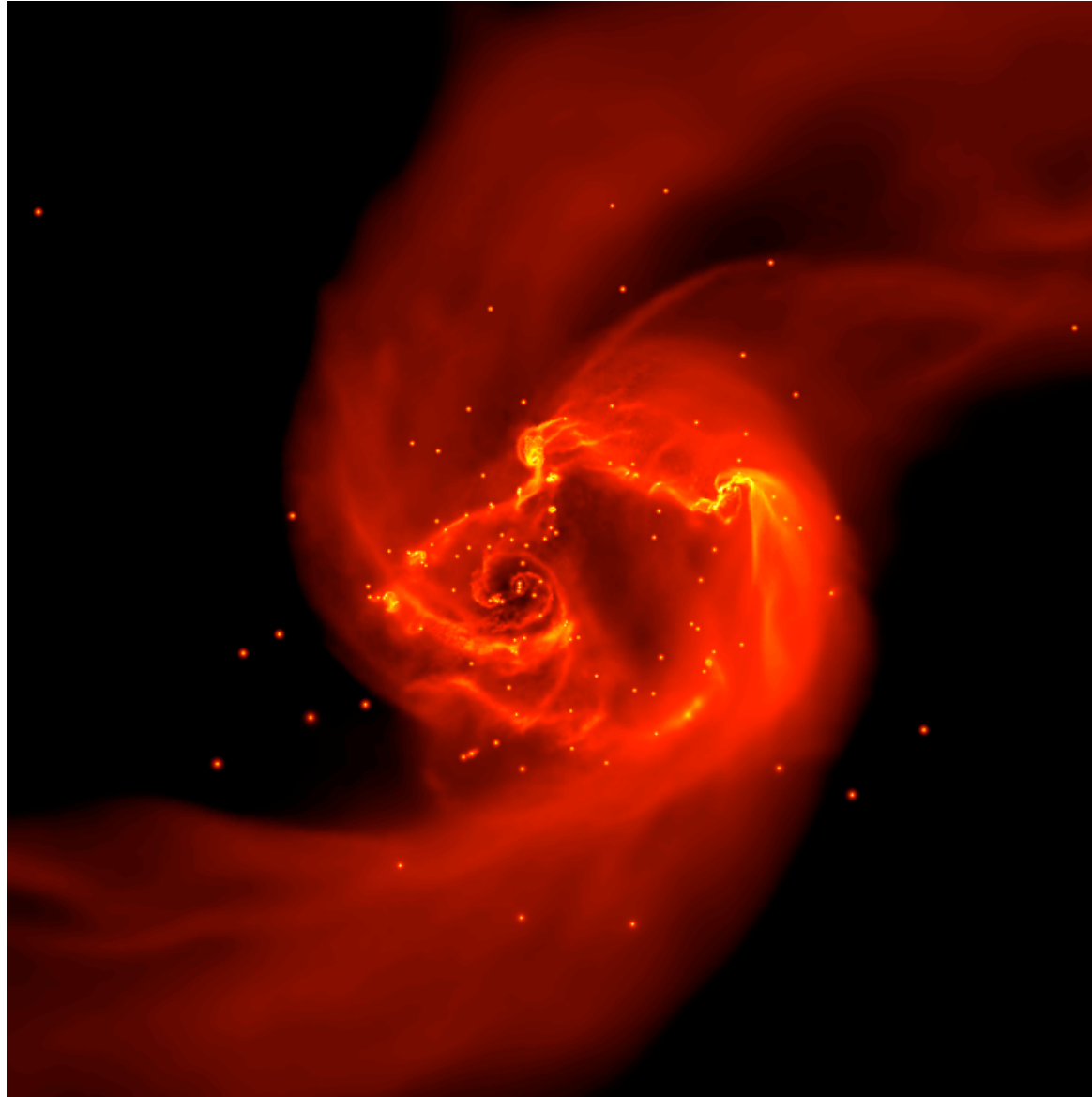


400 AU

(Clark et al. 2007)



dust induced fragmentation at $Z=10^{-5}$



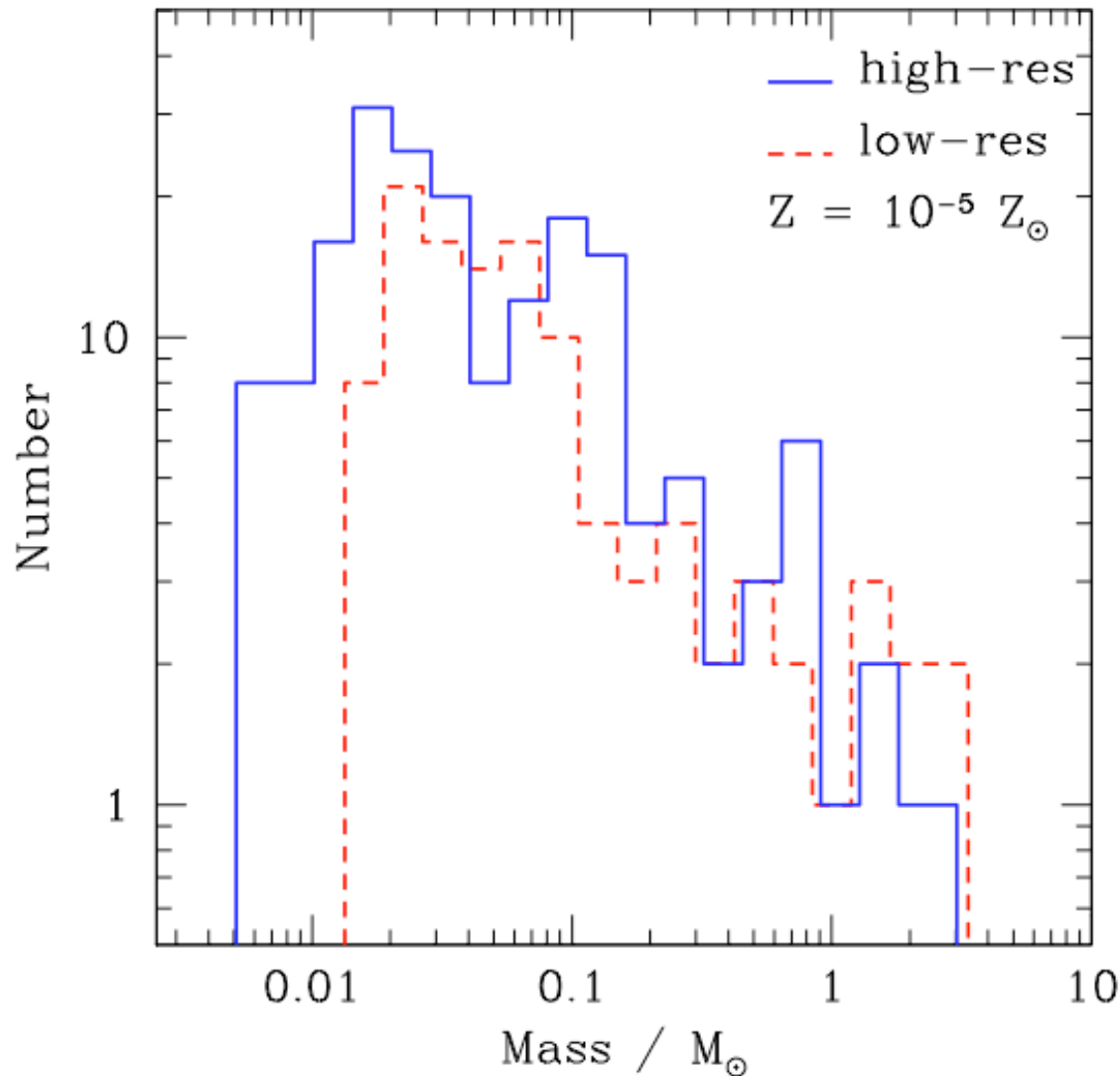
dense cluster of low-mass protostars builds up:

- mass spectrum peaks *below* $1 M_{\text{sun}}$
- cluster VERY dense
 $n_{\text{stars}} = 2.5 \times 10^9 \text{ pc}^{-3}$
- fragmentation at density
 $n_{\text{gas}} = 10^{12} - 10^{13} \text{ cm}^{-3}$

(Clark et al. 2008, ApJ 672, 757)



dust induced fragmentation at $Z=10^{-5}$



dense cluster of low-mass protostars builds up:

- mass spectrum peaks below $1 M_{\text{sun}}$
- cluster VERY dense
 $n_{\text{stars}} = 2.5 \times 10^9 \text{ pc}^{-3}$

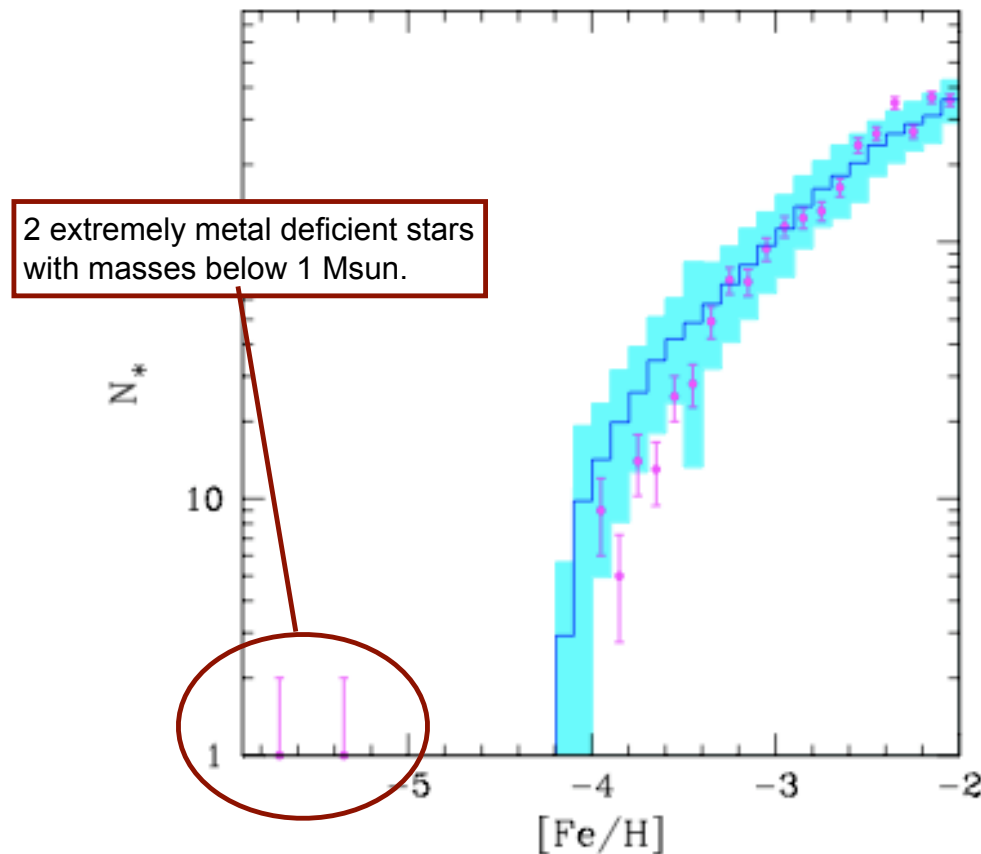
- *predictions:*

- * low-mass stars
with $[\text{Fe}/\text{H}] \sim 10^{-5}$
- * high binary fraction

(Clark et al. 2008)



dust induced fragmentation at $Z=10^{-5}$



(plot from Salvadori et al. 2006, data from Frebel et al. 2005)

dense cluster of low-mass protostars builds up:

- mass spectrum peaks below $1 M_{\text{sun}}$
- cluster VERY dense
 $n_{\text{stars}} = 2.5 \times 10^9 \text{ pc}^{-3}$

- *predictions:*

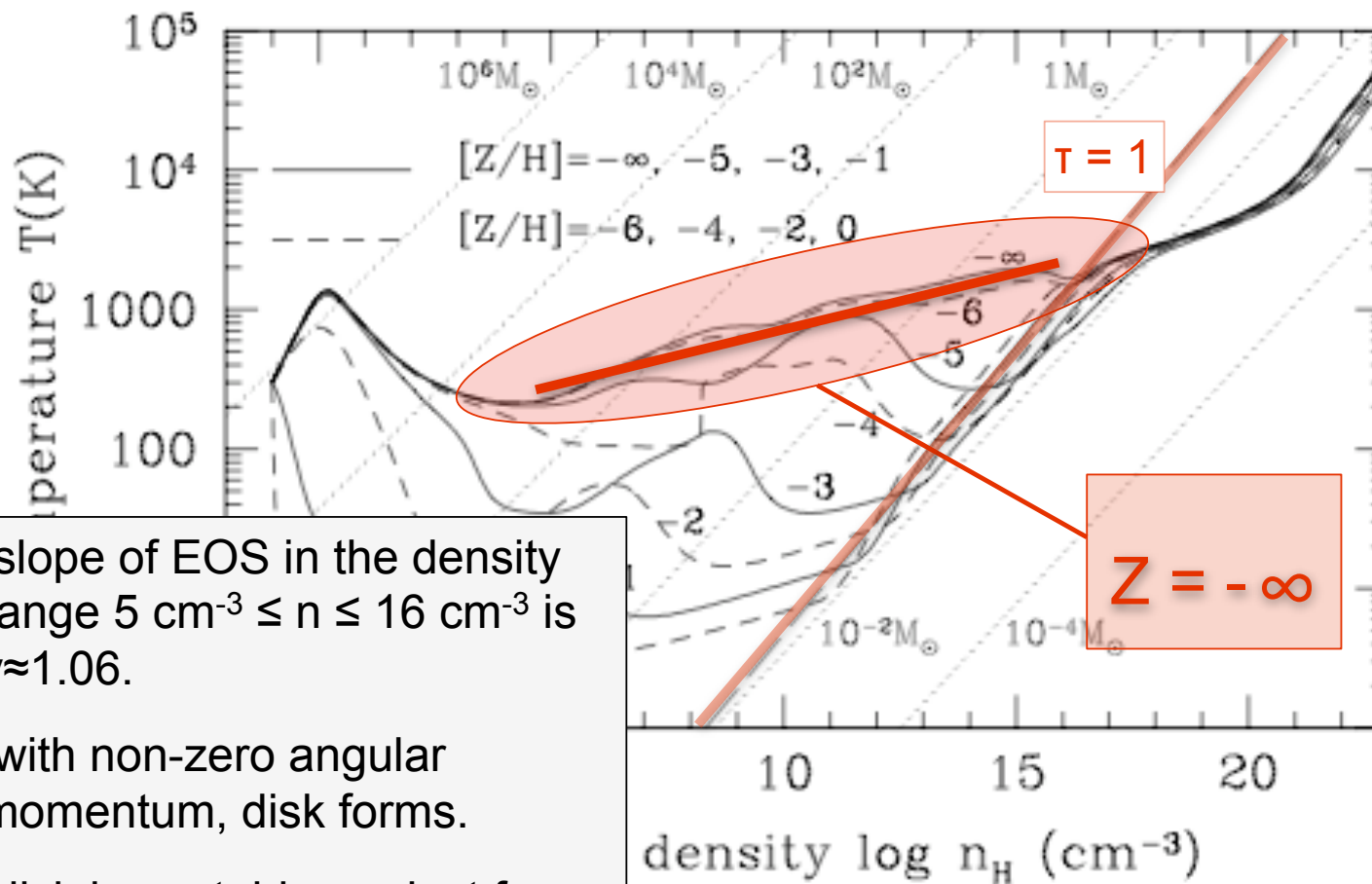
- * low-mass stars
with $[\text{Fe}/\text{H}] \sim 10^{-5}$
- * high binary fraction

(Clark et al. 2008)



metal-free star formation

OMUKAI ET AL.



- slope of EOS in the density range $5 \text{ cm}^{-3} \leq n \leq 16 \text{ cm}^{-3}$ is $\gamma \approx 1.06$.
- with non-zero angular momentum, disk forms.
- disk is unstable against fragmentation at high density



more on $Z=0$ star formation

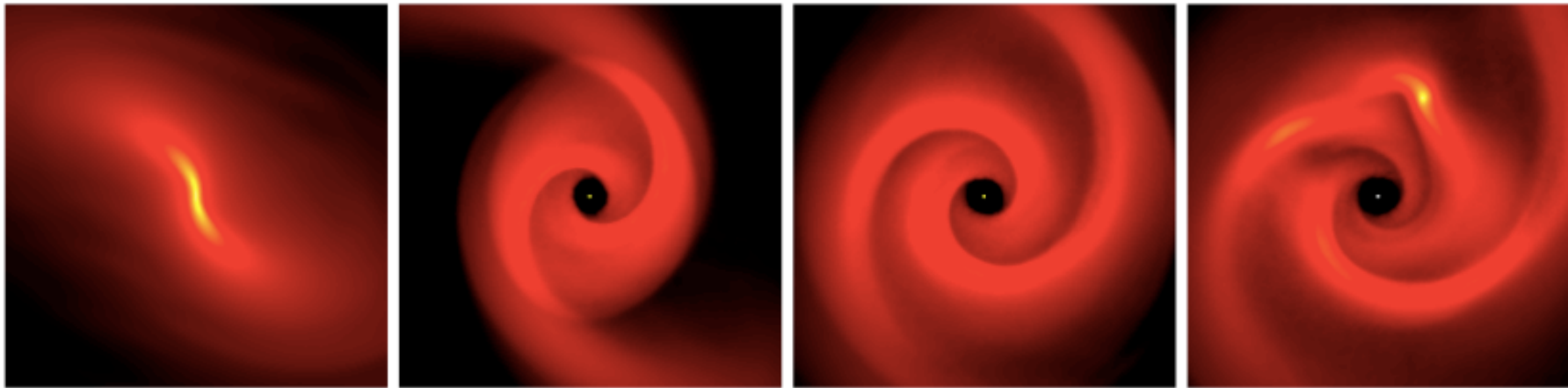


FIGURE 1. Column density images of the inner 66 au of the simulation, following the formation of the first protostar (sink particle) and the subsequent build-up of the protostellar disc and its eventual fragmentation. Starting from left-hand panel, which shows the gas at 1 yr before the protostar forms (t_{SF}), the next 3 panels show the evolution at times $t_{\text{SF}} + 76$ yr, $t_{\text{SF}} + 152$ yr and $t_{\text{SF}} + 228$ yr. The colour table is stretched from 10^3 g cm^{-2} to 10^6 g cm^{-2} .



more on $Z=0$ star formation

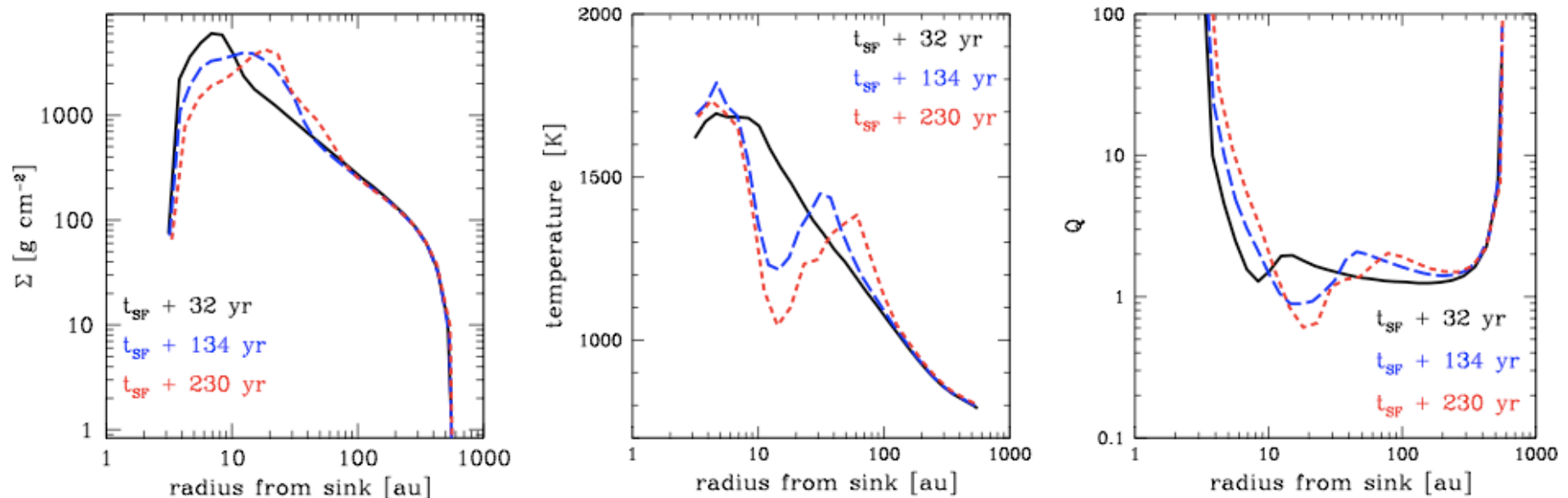


FIGURE 2. In the left-hand and central plots we show the radial profiles of the disc's surface density and gas temperature, centred on the first protostellar core to form in the simulation. The quantities are mass-weighted and taken from a slice through the midplane of the disc. In the right-hand plot we show the radial distribution of the corresponding Toomre parameter, $Q = c_s \kappa / \pi G \Sigma$, where c_s is the sound speed and κ is the epicyclic frequency. We adopt the standard simplification, and replace κ with the orbital frequency.



more on $Z=0$ star formation

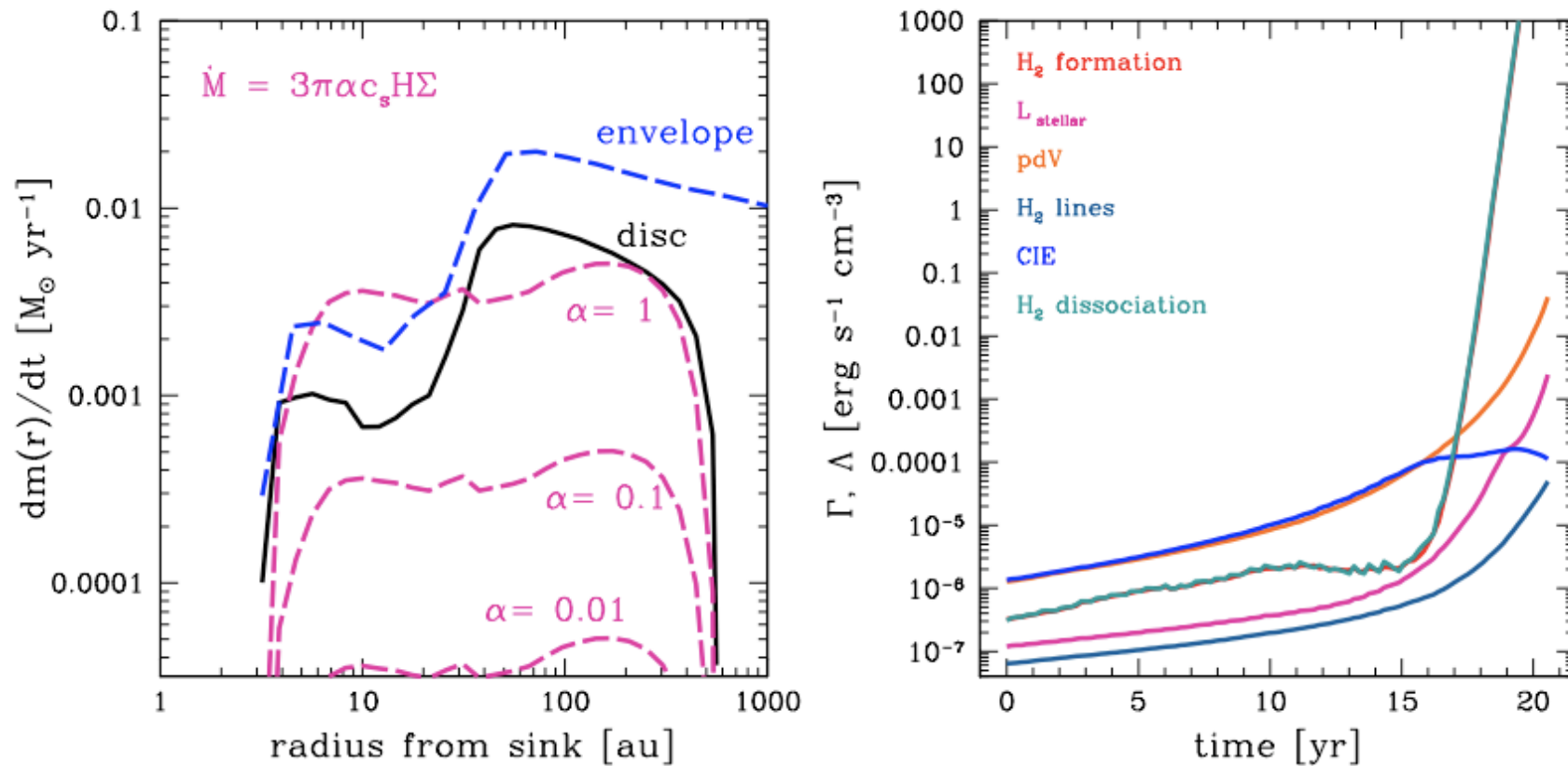


FIGURE 3. The left-hand plot shows the mass transfer through the disc. The solid black line shows the amount of mass moving inwards through each radial annulus in the disc per unit time. The dashed blue line shows the same quantity for the full spherical infalling envelope. The pink dashed lines show the accretion rates expected from an ‘alpha’ (thin) disc model, with three values of alpha. The right-hand plot shows the main heating and cooling processes that control the temperature evolution in the collapsing clump in the run-up to its eventual collapse.

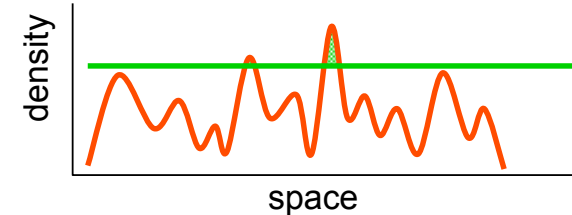


Summary



Summary I

- interstellar gas is highly *inhomogeneous*
 - ◆ *thermal instability*
 - ◆ *gravitational instability*
 - ◆ *turbulent compression* (in shocks $\delta\rho/\rho \propto M^2$; in atomic gas: $M \approx 1...3$)
- cold *molecular clouds* can form rapidly in high-density regions at *stagnation points of convergent large-scale flows*
 - ◆ chemical *phase transition*: atomic \rightarrow molecular
 - ◆ process is *modulated* by large-scale *dynamics* in the galaxy
- inside *cold clouds*: turbulence is highly supersonic ($M \approx 1...20$)
 \rightarrow *turbulence* creates density contrast, *gravity* selects for collapse
 \longrightarrow **GRAVOTUBULENT FRAGMENTATION**
- *turbulent cascade*: local compression *within* a cloud provokes collapse \rightarrow formation of individual *stars* and *star clusters*
- *star cluster*: gravity dominates in large region (\rightarrow competitive accretion)

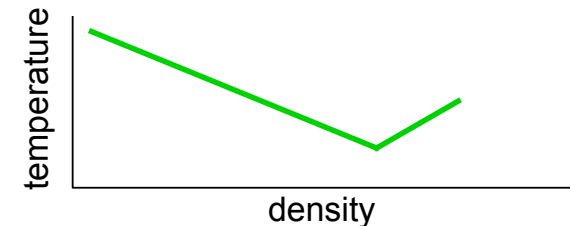




Summary II

- *thermodynamic response* (EOS) determines fragmentation behavior

- characteristic stellar mass from fundamental atomic and molecular parameters
--> explanation for quasi-universal IMF?



- *stellar feedback* is important

- accretion heating may reduce degree of fragmentation
- ionizing radiation will set efficiency of star formation

- *CAVEATS:*

- star formation is *multi-scale, multi-physics* problem --> VERY difficult to model
- in simulations: very small turbulent inertial range ($Re < 1000$)
- can we use EOS to describe thermodynamics of gas, or do we need time-dependent chemical network and radiative transport?
- stellar feedback requires (at least approximative) radiative transport, most numerical calculations so far have neglected that aspect



Thanks!